

Stephen F. Austin State University

SFA ScholarWorks

Electronic Theses and Dissertations

12-2022

Turning Ligands on Their Side: Computational Investigation into the Binding of N₂O and N₂ in Transition Metal Complexes

Cole Donald

Stephen F Austin State University, coledonald9211@gmail.com

Follow this and additional works at: <https://scholarworks.sfasu.edu/etds>

 Part of the [Computational Chemistry Commons](#), [Inorganic Chemistry Commons](#), and the [Physical Chemistry Commons](#)

[Tell us](#) how this article helped you.

Repository Citation

Donald, Cole, "Turning Ligands on Their Side: Computational Investigation into the Binding of N₂O and N₂ in Transition Metal Complexes" (2022). *Electronic Theses and Dissertations*. 484.

<https://scholarworks.sfasu.edu/etds/484>

This Thesis is brought to you for free and open access by SFA ScholarWorks. It has been accepted for inclusion in Electronic Theses and Dissertations by an authorized administrator of SFA ScholarWorks. For more information, please contact cdsscholarworks@sfasu.edu.

Turning Ligands on Their Side: Computational Investigation into the Binding of N₂O and N₂ in Transition Metal Complexes

Creative Commons License



This work is licensed under a [Creative Commons Attribution-Noncommercial-No Derivative Works 4.0 License](#).

TURNING LIGANDS ON THEIR SIDE:
COMPUTATIONAL INVESTIGATION INTO THE
BINDING OF N₂O AND N₂ IN TRANSITION METAL
COMPLEXES

By

COLE B. DONALD, B.S. Chemistry

Presented to the Faculty of the Graduate School of

Stephen F. Austin State University

In Partial Fulfillment

Of the Requirements

For the Degree of

Master of Science in Natural and Applied Science

STEPHEN F. AUSTIN STATE UNIVERSITY

December 2022

TURNING LIGANDS ON THEIR SIDE:
COMPUTATIONAL INVESTIGATION INTO THE
BINDING OF N₂O AND N₂ IN TRANSITION METAL
COMPLEXES

By

COLE B. DONALD, B.S. Chemistry

APPROVED:

J. Brannon Gary, Ph.D., Thesis Director

Brian Barngrover, Ph.D., Committee Member

Russell J. Franks, Ph.D., Committee Member

Alyx Frantzen, Ph.D., Committee Member

Christopher Aul, Ph.D., Committee Member

Sheryll Jerez, Ph.D.
Interim Dean of Research and Graduate Studies

© Cole Bridges Donald
All Rights Reserved
2022

Abstract

Common greenhouse gas nitrous oxide (N_2O) is a thermodynamically potent and environmentally benign oxidant, making it a desirable target for metal center activation. Unfortunately, N_2O is a poor ligand for transition metals due to its weak σ -donating and π -accepting properties; as a result, few transition metal complexes capable of interacting with N_2O have been found. As the primary source of all nitrogen in organisms, abundant gas dinitrogen (N_2) is a crucially important tiny molecule and an essential part of daily existence. However, due to its inertness, it has limited practical uses in this form. Through biological and commercial nitrogen fixation processes, one of the most inert substances, N_2 , is transformed into an accessible nitrogen supply, such as NH_3 , that may be incorporated into all nitrogen-containing biomolecules. Using computational chemistry, this work will highlight the energy differences between new potential N_2O binding modes. Insights into the comparison between the $\kappa\text{-N}$ and $\kappa\text{-O}$ versus the newly reported $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ binding modes will be discussed. Through the utilization of density functional theory, a low valent cobalt complex possessing N_2O in a $\eta^2\text{-NO}$ coordination is reported. These binding mode comparisons can be employed to develop N_2O as a "green" oxidant given the limited understanding of the coordination of N_2O to metal centers. This study will emphasize the energy disparities between putative N_2 binding mechanisms using computational chemistry. It will also be explored how the $\kappa\text{-N}$ binding mode compares to

the infrequently reported η^2 -NN binding mode. Given the partial knowledge of N₂'s coordination to metal centers, these binding mode comparisons can be used to improve understanding of N₂'s activation.

Dedication

This thesis work is completely dedicated too and would not have been feasible without the unwavering support of my dear parents, Nancy S. Walker-Copland and Therion C. Donald. They have consistently offered their moral, spiritual, and emotional support; and have been a great source of inspiration.

To Dad

Therion C. Donald

6/14/1975 – 7/4/2022

Though you never got to see this

you're in every page

Acknowledgments

First and foremost, I want to express my gratitude to my parents and extended family for their unwavering support and exemplary patience during my recent and ongoing academic endeavors. I want to express my thanks to my mom for supporting me through the highs and lows of my academic career. I owe my thanks to my father for teaching me to never give up on my ambitions-whatever the difficulties; without your "life lessons," I would not have been able to complete my thesis. I also owe a great deal of gratitude to my stepmom and stepdad for teaching me the value of perseverance, self-respect, and assertiveness. Mom, Dad, Rebecca, and Ray, words cannot express how thankful I am for everything you have done for me and for having me as your son.

Dr. Gary, you asked me to join "Ellis Island" as your second graduate student, and you helped me develop as an inorganic chemist over the years. I can only try to express how grateful I am for you. You have taught me so much, from technical know-how and background information to perceptions on potential future paths in industry. Throughout my stay at SFA, I have benefited greatly from your friendship and guidance.

Dr. Barngrover, you have served as my academic, research, and spiritual advisor, and an excellent friend for the past four years. I appreciate the amount of research you allowed me to do for my degrees and for helping me to grow as a computational chemist. You supported me at all the highs and lows of my academic career and any challenging times

in my personal life. No words or deeds can express how grateful we are for your advice and support. Thank you, Dr. Franks, for the humorous anecdotes you used to illustrate your teachings. Dr. Frantzen, I want to thank you for all the effort you spent over the years guiding me through my physical chemistry classes. I'm also grateful for all the useful advice and suggestions you gave me regarding experimental techniques. Along with offering advice, emotional support, and what seemed like an endless supply of sweets, Mrs. Stephanie Campbell and Ms. Heidi Wilson deserve my sincere appreciation.

I also want to express my gratitude to my friends and classmates for their assistance in getting me through some of the more challenging components of this degree. I am grateful that Gracie, Justin, and I were able to work together towards our graduate studies as "The Dream Team", and I am looking forward to following both of your academic and professional paths. Thank you for all your advice, encouragement, and humor throughout the years to numerous other instructors, friends, and classmates.

Finally, I want to thank the Chemistry & Biochemistry Department for being my academic "home" over the last four years, as well as the SFA ORSP RCA Grant, and the Robert A. Welch Foundation for supporting my research.

Table of Contents

Abstract	ii
Dedication	iv
Acknowledgments.....	v
List of Figures	xi
Abbreviations.....	xvii
Chapter 1	1
Small Molecule Activation Through Metal Binding.....	1
1.1 Green Oxidation	2
1.2 Why is N ₂ O an Attractive Oxidant?.....	3
1.3 Why is N ₂ Reduction Important?	5
1.4 Coordination Chemistry	7
1.5 Looking Ahead.....	11
1.6 References	12
Chapter 2.....	15
Computational Investigation of the Preferred Binding Modes of N ₂ O in Group 9 & 10 Metal Complexes.....	15
2.1 Introduction	15

2.2	Computation Details.....	19
2.3	Results and Discussion.....	19
2.4	Conclusion.....	30
2.5	References	31
Chapter 3		33
Computational Investigation into the Activation of N ₂ Through Binding in Transition Metal Complexes.....		33
3.1	Introduction	33
3.2	Computation Details.....	38
3.3	Results and Discussion.....	38
3.4	Conclusion.....	55
3.5	References	57
Chapter 4		64
Conclusions and Future Prospects.....		64
4.1	Conclusions	64
4.2	Future Prospects	66
4.3	References	68

Appendix.....	69
VITA	236

List of Schemes

Scheme 3.1 Isomerization of $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})_2]$ and $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})(\text{PMe}_3)]$ by the Sutton lab	53
Scheme 4.1 Binding between the nitrogen and oxygen atoms of the N_2O	65
Scheme 4.2 Electronic effect on N_2 binding.....	66

List of Figures

Figure 1.1 N ₂ O emissions, both natural and man-made, in Mt N ₂ O/yr (megatons of N ₂ O in equivalent nitrogen units per year)	4
Figure 1.2 Carbonyl ligand orbital interactions involved in binding with a transition metal.....	8
Figure 1.3 Orbital interactions of the olefin binding model by Chatt-Dewer-Duncanson.....	9
Figure 1.4 Metal-alkyne complexes exhibit σ -donation, π -acceptance, and π -donation.....	10
Figure 1.5 Trans influence series	10
Figure 2.1 Potential monometallic N ₂ O binding.....	16
Figure 2.2 Characterized structures with an N ₂ O unit	17
Figure 2.3 N ₂ O binding modes observed in previous work	18
Figure 2.4 Characterized structures with π -bound N ₂ O unit.....	19
Figure 2.5 Ligand modification effect on binding energy in [Ni(L) ₂ (N ₂ O)]	20
Figure 2.6 Ligand modification effect on binding in [Co(η^5 -Cp*)(N ₂ O)].....	22
Figure 2.7 Ligand modification effect on binding energy in [Co(η^5 -Cp)(L)(N ₂ O)].....	23
Figure 2.8 Para-substituted pyridine ligand modification effect on binding energy in [Co(η^5 -Cp)(L)(N ₂ O)]	25

Figure 2.9 Metal identity modification effect on η -NN binding in $[[M]CNMe(\eta^5-Cp)(N_2O)]$	25
Figure 2.10 Ligand identity modification effect on binding energy in $[[Co](L)_2Cl(N_2O)]$	27
Figure 2.11 Ligand modification effect on binding in $[Co(C_4N_2H_5)_2(NC_4H_8)(N_2O)]$	28
Figure 2.12 (a) Space-filling model of $[Co(C_4N_2H_5)_2(NC_4H_8)]$ in both the η^2 -NN and η^2 -NO configuration, (b) Space-filling model of $[Co(C_6N_2H_{11})_2(NC_4H_8)]$ in both the η^2 -NN and η^2 -NO configuration	29
Figure 2.13 Ligand modification effect on binding in $[Co(C_6N_2H_{11})_2(NC_4H_8)(N_2O)]$	30
Figure 3.1 Major transformations in the nitrogen cycle.....	34
Figure 3.2 FeMo-co, [7Fe-9S-Mo-C-homocitrate], with a cartoon of active Fe 2,3,6,7 FeMo-co face. PDB: 2AFI.....	36
Figure 3.3 Ligand modification effect on binding energy in $[Ni(L)_2(N_2)]$	39
Figure 3.4 Ligand modification effect on binding in $[Co(\eta^5-Cp^*)(CNMe)(N_2)]$	41
Figure 3.5 Ligand modification effect on binding energy in $[Co(\eta^5-Cp)(L)(N_2)]$	43
Figure 3.6 Ligand modification effect on binding energy in $[Co(\eta^5-Cp)(L)(N_2)]$	45
Figure 3.7 Metal identity modification effect on binding energy in $[[M]CNMe(\eta^5-Cp)(N_2)]$	47

Figure 3.8 Ligand identity modification effect on binding energy in $[\text{Co}(\text{L})_2\text{Cl}(\text{N}_2)]$	48
Figure 3.9 Metal identity modification effect on binding energy in $[[\text{M}](\eta^5-\text{Cp}^*)_2(\text{N}_2)]$	50
Figure 3.10 Ligand modification effect on binding in $[\text{Re}(\eta^5-\text{Cp}^*)(\text{CO})_2]$ and $[\text{Re}(\eta^5-\text{Cp}^*)(\text{CO})(\text{PMe}_3)]$	52
Figure 3.11 Metal identity modification effect on binding energy in $[[\text{M}](\text{NH}_3)_5(\text{N}_2)]$	53

List of Tables

Table 1.1 CO stretching frequencies of a variety of hexacarbonyl complexes ²⁰	11
Table 2.1 C-O and C-C stretch vibrational frequencies (in cm ⁻¹) for [Ni(L) ₂ (CO)] and [Ni(L) ₂ (CH) ₂] complexes	21
Table 2.2 C-O and C-C stretch vibrational frequencies (in cm ⁻¹) for [Co(η^5 -Cp)(L)(CO)] and [Co(η^5 -Cp)(L)(CH) ₂] complexes.....	24
Table 3.1 C-O and C-C stretch vibrational frequencies (in cm ⁻¹) for [Ni(L) ₂ (CO)] and [Ni(L) ₂ (CH) ₂] complexes	40
Table 3.2 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm ⁻¹) for [Ni(L) ₂ (N ₂)] complexes	41
Table 3.3 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm ⁻¹) for [Co(η^5 -Cp*)(CNMe)(N ₂)]	42
Table 3.4 C-O and C-C stretch vibrational frequencies (in cm ⁻¹) for [Co(η^5 -Cp)(L)(CO)] and [Co(η^5 -Cp)(L)(CH) ₂] complexes.....	44
Table 3.5 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm ⁻¹) for [Co(η^5 -Cp)(L)(N ₂)]	45
Table 3.6 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm ⁻¹) for para-substituted pyridine series [Co(η^5 -Cp)(L)(N ₂)]	46

Table 3.7 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm ⁻¹) for [[M](η ⁵ -Cp)(CNMe)(N ₂)]	48
Table 3.8 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm ⁻¹) for [Co(L) ₂ Cl(N ₂)].....	49
Table 3.9 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm ⁻¹) for [[M](η ⁵ -Cp [*]) ₂ (N ₂)]	51
Table 3.10 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm ⁻¹) for [[M](NH ₃) ₅ (N ₂)].....	54
Table 3.11 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm ⁻¹) for [[M](η ⁵ -Cp)(N ₂)Cl ₂]	55
Table 4.1 Trans ligand modification effect on terminal nitrogen vs oxygen binding	65

List of Equations

Equation 1.1	6
Equation 1.2	7

Abbreviations

N₂ dinitrogen

CH₄ Methane

CO₂ Carbon dioxide

EPA Environmental Protection Agency

N₂O Nitrous oxide

GWP Global warming potential

CFCs Chlorofluorocarbons

NH₃ Ammonia

FeMo-co FeMo-cofactor

CPⁱPr₃ Tris(phosphino)alkyl

DFT Density functional theory

MO Molecular orbital

DSC Differential Scanning Calorimetry

ECP Effective core potential

Cp Cyclopentadienyl

Cp* Pentamethylcyclopentadienyl

L Ligands

M Metal

Chapter 1

Small Molecule Activation Through Metal Binding

Since small molecule activation has a significant impact on environmental protection, industrial catalysis, medicine, and biology, chemistry research has long concentrated on the sustainable conversion of molecules like dinitrogen (N_2), methane (CH_4), and carbon dioxide (CO_2) into other compounds. The catalytic activation of generally inert small molecules has maintained a constant presence in the research community from the standpoint of inorganic chemistry.¹

Although molecules like N_2 , and CO_2 are plentiful and easily available, the majority of them are inert, and their chemical transformations are difficult from a thermodynamic and mechanistic standpoint. Such changes frequently occur along difficult-to-control reaction pathways that combine multielectron redox reactions with proton transfers. For the design and development of effective catalysts, it is imperative to understand how these small molecules bind to metals to determine their modes of activation and solve the mechanistic foundation, which provides a significant task for the chemistry community.¹

Bioinorganic chemistry has been a major source of inspiration for the subject since nature mostly employs metal ions to activate these relatively inert compounds and control their reactivity. But now that industrial processes of small-molecule transformations that encompass enzymatic processes, homogeneous organometallic chemistry, and

heterogeneous catalysis have been developed, attempts to further the catalysis efficiency, utilization, and waste elimination of these processes are currently under investigation.²

1.1 Green Oxidation

The current most crucial chemical process for giving a molecule functionality is oxidation. Almost every area of the chemical and related industries will continue to depend on oxidation, particularly selective oxidation, even if hydrocarbons remain our main feedstock. The US Environmental Protection Agency (EPA) launched the Green Chemistry Movement in the early 1990s to persuade businesses and academics to utilize chemistry to stop pollution. The objective of "green chemistry" was to "advance novel chemical technologies that decrease or eliminate the use or creation of hazardous compounds in the design, manufacture, and usage of chemical goods," to be more precise.⁵ The employment of metallic oxidants, such as permanganate, is one particularly dangerous and harmful way of oxidation that has a long history in the chemistry of oxidation. The design of cleaner oxidation processes depends critically on the supply of oxygen. The most appealing option is frequently air or oxygen, and hydrogen peroxide, which produces just water as a byproduct, is a good second choice.⁴

Current oxidation chemistry is at a turning point, particularly as it relates to industrial operations. Commodity oxidations generate many megatons of products with a market value of tens of billions of dollars. Due to their high prices and environmental implications, reagents like permanganate and chromate have lost their appeal. The creation of greener

and more effective oxidative processes depends on an understanding of the interactions between metal ions and oxidants.⁶

1.2 Why is N₂O an Attractive Oxidant?

The greatest stratospheric ozone-depleting gas and the third most important anthropogenic greenhouse gas is nitrous oxide (N₂O).⁷ Due to its long atmospheric lifespan of approximately 114 years, N₂O, in particular, has a global warming potential (GWP) that is almost 300 times greater than that of CO₂ and depletes the ozone layer at the same time, much like chlorofluorocarbons (CFCs). Therefore, knowledge of its causes and management could significantly aid in the stability of the planet's climate.⁸ A third of N₂O production is emitted by industrial sources, including the manufacturing of nitric acid and nylon, as well as from soils, particularly those that have had fertilizer treatment.⁵ Agricultural operations account for 4.1 Mt N₂O/yr, or 66 percent of all gross anthropogenic emissions, making them by far the greatest source of N₂O emissions. Agricultural N₂O emissions are mostly caused by manure, synthetic fertilizers, and crop leftovers.⁹ Forest fires, burning of crop residues, and burning of biomass for cooking and heating are the main sources of N₂O emissions from biomass. As of 2015, biomass-derived N₂O emissions amount to 0.7 Mt N₂ON/yr, or 11% of all gross anthropogenic emissions (Figure 1.1).⁸

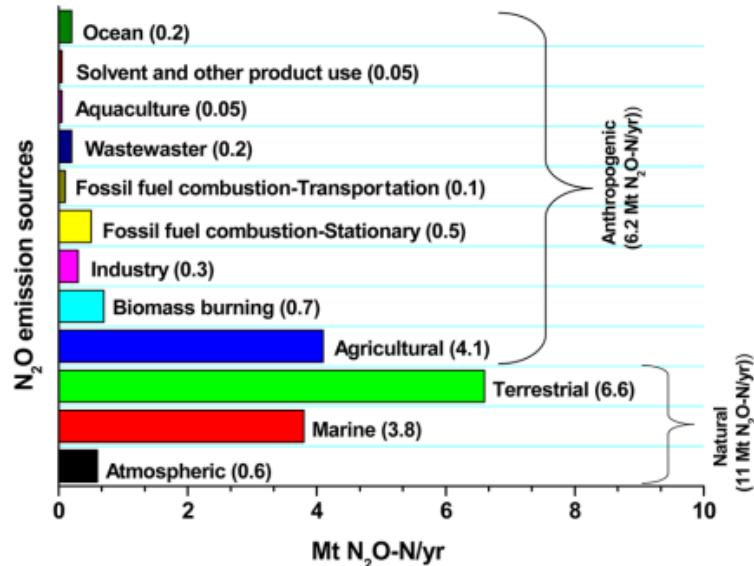


Figure 1.1 N₂O emissions, both natural and man-made, in Mt N₂O/yr (megatons of N₂O in equivalent nitrogen units per year)⁸

Understanding the reactivity of N₂O with metal centers in several contexts, such as the gas phase, solid substrates, enzymes, and soluble complexes in solution has been one key research goal motivated by these metal-promoted reactions. Few transition metal complexes are reported that are capable of interacting with N₂O.^{10–16} Given the potential applications of N₂O as an oxidant for O-atom transfer reactions, broadening the scope of this class of compounds would allow for the development of a variety of useful processes.¹⁷ Complexes that can activate nitrous oxide for useful processes are extremely rare. Since N₂O is a potent thermodynamic oxidant and can act as an oxygen-atom transfer reagent giving the ideal chemical byproduct of N₂, it is a preferred oxidant for chemical synthesis. Since N₂O has a strong oxidizing potential and doesn't present the same explosive risks as O₂, it is also considered to be an effective substitute for O₂. Future discoveries could

advance our knowledge of N₂O's known binding properties and open the door to more research on this peculiar and potentially important oxidant.

1.3 Why is N₂ Reduction Important?

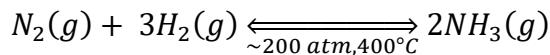
N₂ is the main element that makes up the Earth's atmosphere. Thereby, N₂ is a continually abundant molecule around the globe, yet in this form, it has few practical applications due to its inertness. However, molecular nitrogen is the initial source of all nitrogen in organisms, making it a crucial component of life's chemistry. This rich supply of nitrogen is only useful to a few species. Nitrogen fixation is the method by which N₂ is added to biological systems; it involves the nitrogenase enzyme, which has a metal-sulfido cluster in the active site.¹⁸

One of the most inert compounds, N₂, is converted into a bioavailable nitrogen source, such as NH₃, that may be integrated into all nitrogen-containing biomolecules through biological and industrial nitrogen fixation processes. As a result, nitrogen fixation has drawn close attention from the chemical community for decades since it is crucial to maintaining life on this planet. Currently, it is still not understood mechanistically how the nitrogenase enzymes encourage the biological reduction of nitrogen in the presence of ambient conditions.⁶

The energy-intensive Haber-Bosch process, which produces millions of tons of ammonia each year straight from three equivalents of dinitrogen and dihydrogen molecules with the help of a metal catalyst, is perhaps the most well-known and oldest of these processes.¹⁹ (Equation 1.1). This reaction is exothermic and thermodynamically favorable at ambient

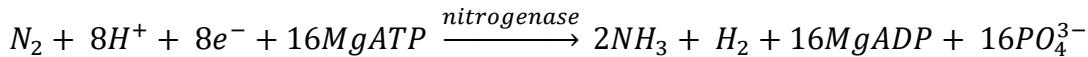
conditions although the process is not kinetically favorable since the feedstock gases must be compressed to several hundred atmospheres to promote ammonia synthesis at the high temperatures currently required.¹⁸ By raising the temperature at which the reaction is conducted, the kinetics can be overcome, but the process's thermodynamics becomes less favorable.

Equation 1.1¹⁸



In 1886, it was first shown that some species could fix atmospheric N₂. The biological fixation of N₂ by nitrogenase-type enzymes is performed at ambient temperature and pressure, although using a reaction that needs an electron supply, in contrast to the high-temperature commercial procedures mentioned above (Equation 1.2).¹⁸ It takes approximately sixteen equivalents of MgATP to complete this reaction, which is energy-intensive. Even while there have been several attempts to identify new processes that are superior to the Haber-Bosch and nitrogenase processes, none of them have been successful, but future discoveries could advance our knowledge of dinitrogen's reduction by nitrogenase and lead to a greener alternative to the century-old Haber-Bosch process.

Equation 1.2¹⁸



1.4 Coordination Chemistry

Many ligands in coordination and organometallic chemistry act as π -acceptors, π -donors, or both in addition to being σ -donors. The π^* -orbitals of several ligands that are involved in π -bonding with transition metal centers are unoccupied.²⁰ As demonstrated for a carbonyl ligand in Figure 1.2,²¹ these π^* -orbitals overlap with filled metal d-orbitals to produce a lower energy-filled bonding orbital and a higher energy empty antibonding orbital. The interaction between the metal and the ligand π^* -orbital results in the shift of electron density from the metal onto the carbonyl ligand because the filled bonding molecular orbital has some carbonyl π^* character. This results in the reduction of the CO bond order and more than makes up for the lone pair on carbon's contribution of electron density to the metal. Back bonding is the term for this transfer of electron density from the metal to the π^* -orbital. In short, the carbonyl group's infrared stretching frequency is lower than that of free CO, while the C-O distance in carbonyl ligands is longer than that of free

CO. These demonstrate how the back bonding interaction reduces the order of the C-O bonds.^{20,21}

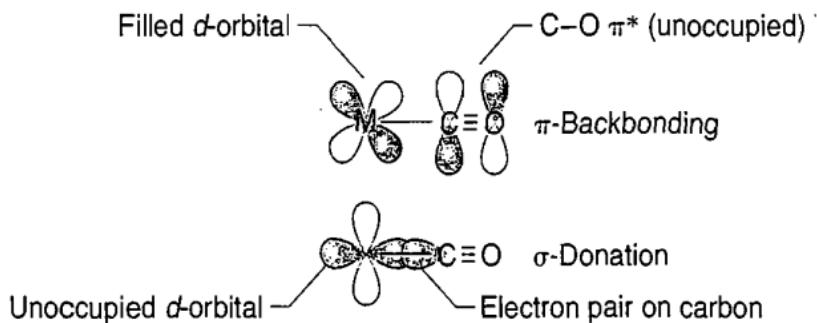


Figure 1.2 Carbonyl ligand orbital interactions involved in binding with a transition metal.²¹

Olefins are a significant type of π -acceptor ligands that attach to the metal via their π -bonds. The orbital interactions in a metal-olefin complex are depicted in Figure 1.3,²¹ which was initially created by Chatt, Dewar, and Duncanson.^{20,21} The interactions between the olefin ligand and the metal during metal-olefin binding include both σ -donating and π -accepting interactions. One interaction includes the donation of electron density from the olefin's π -bonding orbital to a metal's open orbital. Since there is no nodal plane along this bond axis, coordination to the olefin's face results in the formation of one σ -bonding interaction. In the second interaction, an occupied orbital at the metal and the olefin's π -antibonding orbital overlap. The overlap between the metal's d-orbitals and the p-orbitals of two carbon atoms at the face of the π -system transpires instead of the overlap between the metal and the π -system of CO through the p-orbital of one atom at the end of the ligand. One nodal plane exists on the face of the π^* -orbital of an olefin, and one nodal plane exists

along the bond axis in the orbital that results from the interaction of the filled orbital at the metal with the π^* -orbital of the olefin. As a result, this interaction is regarded as a π -bond. The interaction of an olefin's π^* -orbital with a full metal orbital results in a flow of electron density into the ligand's π^* -orbital and lowers the bond order of the π -bond, similar to the interaction of CO's π^* -orbital with a filled metal d-orbital. As a result, the coordination of the olefin in such a complex causes the carbon-carbon bond order to drop and the length of the C-C bond to increase relative to the free olefin. In addition, the olefin's substituents are twisted away from the metal, and hybridization at the olefinic carbons moves from sp^2 to sp^3 .^{20,21}

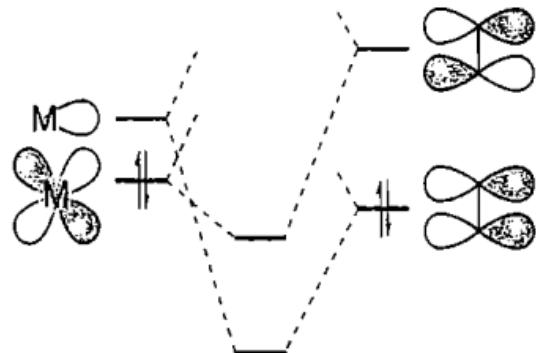


Figure 1.3 Orbital interactions of the olefin binding model by Chatt-Dewar-Duncanson²¹

Alkynes, like olefins, may act as an σ -donor, a π -acceptor, as well as a π -donor. Figure 1.4,²¹ illustrates three interactions that can be used to explain the bonding of alkynes: σ -donation by one set of π -bonding orbitals, π -bonding by an orthogonal set of π -bonding orbitals, and π -back bonding into an alkyne's π^* orbital.^{20,21}

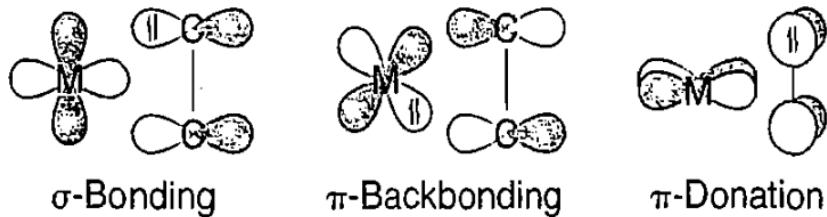


Figure 1.4 Metal-alkyne complexes exhibit σ -donation, π -acceptance, and π -donation²¹

In this research, the electronic and trans-influence characteristics of the metals and ligands are modified. The impact of a ligand on the length of the bond trans to it in the ground state of a complex is known as the *trans*-influence, which is a thermodynamic effect. One ligand, for instance, might have an impact on the length of the metal-ligand bond involving the ligand *trans* to it. Based on the respective σ -donor characteristics of the ligands, the ranking in Figure 1.5 forecasts the order for the trans influence.²⁰



Figure 1.5 Trans influence series²⁰

Five isoelectronic hexacarbonyls are shown to have the following CO stretching frequencies in Table 1.1,²⁰ which illustrate how a complex's electronics affect the CO bond. These numbers can be contrasted with the free CO's 2143 cm^{-1} stretching frequency. From the table, it can be seen that $[\text{Ti}(\text{CO})_6]^{2-}$ includes the most highly reduced metal, technically Ti(2-), which indicates that titanium has the greatest potential to back-donate electron density to CO and the least capacity to attract electrons. With $[\text{Fe}(\text{CO})_6]^{2+}$, the formal charges on the metals go from -2 for $[\text{Ti}(\text{CO})_6]^{2-}$ to +2. With the highest negative formal

charge, the titanium in $[\text{Ti}(\text{CO})_6]^{2-}$ has the strongest propensity to donate to CO. The result is a high population of CO's π^* orbitals in $[\text{Ti}(\text{CO})_6]^{2-}$ and a weakening of the CO bond. In general, the likelihood of the metal donating electrons to the π^* orbitals of CO increases with the negative charge on the organometallic species, and the energy of the CO stretching vibrations decreases.²⁰

Table 1.1 CO stretching frequencies of a variety of hexacarbonyl complexes²⁰

Complex	$\nu(\text{CO})$, cm^{-1}
$[\text{Ti}(\text{CO})_6]^{2-}$	1748
$[\text{V}(\text{CO})_6]^-$	1859
$\text{Cr}(\text{CO})_6$	2000
$[\text{Mn}(\text{CO})_6]^+$	2100
$[\text{Fe}(\text{CO})_6]^{2+}$	2204

1.5 Looking Ahead

The energy differences between putative new N_2O binding modes will be highlighted in Chapter 2 along with comparisons of the previously known $\kappa\text{-N}$ and $\kappa\text{-O}$ binding modes to the recently revealed $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ binding modes in order to gain a better understanding of how modifying the metal, whether it be through metal identity or ligand changes, relates to small molecule binding preferences. In Chapter 3, N_2 is discussed in relation to how it binds to and is activated by transition metal complexes to reduce the N_2 triple bond.

1.6 References

- (1) Roesky, P. W.; Fout, A. R. Diversity in Small-Molecule Activation: The Adventure Continues. *Inorg. Chem.* **2021**, *60* (18), 13757–13758.
<https://doi.org/10.1021/acs.inorgchem.1c02529>.
- (2) Meyer, F.; Tolman, W. B. Forums on Small-Molecule Activation: From Biological Principles to Energy Applications. *Inorg. Chem.* **2015**, *54* (11), 5039.
<https://doi.org/10.1021/acs.inorgchem.5b00768>.
- (3) Parent, K. E. Bleaching with Green Oxidation Chemistry.
- (4) Hill, C. L. Controlled Green Oxidation. *Nat. Chem.* **1999**, *401* (6752), 436–437.
<https://doi.org/10.1038/46704>.
- (5) Irles, A.; Gonçalves, I. C.; Lopes, M. C.; Fernandes, A. C.; Ramalho, A. G.; Pertusa, J. *Handbook of Green Chemistry and Technology*; 2001; Vol. 17.
- (6) Tolman, W. B. *Activation of Small Molecules: Organometallic and Bioinorganic Perspectives*; 2006. <https://doi.org/10.1002/9783527609352>.
- (7) Bange, H. W.; Arévalo-Martínez, D. L.; Paz, M. de la; Farías, L.; Kaiser, J.; Kock, A.; Law, C. S.; Rees, A. P.; Rehder, G.; Tortell, P. D.; Upstill-Goddard, R. C.; Wilson, S. T. A Harmonized Nitrous Oxide (N_2O) Ocean Observation Network for the 21st Century. *Front Mar Sci* **2019**, *6* (APR), 157.
<https://doi.org/10.3389/FMARS.2019.00157/BIBTEX>.
- (8) Konsolakis, M. Recent Advances on Nitrous Oxide (N_2O) Decomposition over Non-Noble-Metal Oxide Catalysts: Catalytic Performance, Mechanistic Considerations, and Surface Chemistry Aspects. *ACS Catal.* **2015**, *5* (11), 6397–6421.
https://doi.org/10.1021/ACSCATAL.5B01605/ASSET/IMAGES/LARGE/CS-2015-01605M_0019.JPG.
- (9) Ravishankara, A. R.; Daniel, J. S.; Portmann, R. W. Nitrous Oxide (N_2O): The Dominant Ozone-Depleting Substance Emitted in the 21st Century. *Science* (1979) **2009**, *326* (5949), 123–125.
https://doi.org/10.1126/SCIENCE.1176985/SUPPL_FILE/RAVISHANKARA.SOM.PDF.
- (10) Ben-Daniel, R.; Neumann, R. Activation of Nitrous Oxide and Selective Oxidation of Alcohols and Alkylarenes Catalyzed by the $[\text{PV}_2\text{MO}_{10}\text{O}_{40}]^{5-}$ Polyoxometalate Ion.

Angew. Chem., Int. Ed. **2003**, *42* (1), 92–95.
<https://doi.org/10.1002/ANIE.200390062>.

- (11) Harman, W. H.; Chang, C. J. N₂O Activation and Oxidation Reactivity from a Non-Heme Iron Pyrrole Platform. *J. Am. Chem. Soc.* **2007**, *129* (49), 15128–15129.
<https://doi.org/10.1021/JA076842G>.
- (12) Leont'ev, A. v; Fomicheva, O. A.; Proskurnina, M. v; Zefirov, N. S. Modern Chemistry of Nitrous Oxide. *Russ. Chem. Rev.* **2001**, *70* (2), 91–104.
<https://doi.org/10.1070/RC2001V070N02ABEH000631>.
- (13) Andino, J. G.; Caulton, K. G. Mechanism of N/O Bond Scission of N₂O by an Unsaturated Rhodium Transient. *J. Am. Chem. Soc.* **2011**, *133* (32), 12576–12583.
<https://doi.org/10.1021/JA202439G>.
- (14) Otsuka, K.; Wang, Y. Partial Oxidation of Methane with N₂O over Fe₂(MoO₄)₃ Catalyst. *Catal. Lett.* **1994**, *24* (1–2), 85–94. <https://doi.org/10.1007/BF00807378>.
- (15) Parmon, V. N.; Panov, G. I.; Uriarte, A.; Noskov, A. S. Nitrous Oxide in Oxidation Chemistry and Catalysis: Application and Production. *Catal. Today* **2005**, *100* (1–2), 115–131. <https://doi.org/10.1016/J.CATTOD.2004.12.012>.
- (16) Xiao, J. D. ; et. al. Oxidation of Ethane to Ethanol by N₂O in a Metal-Organic Framework with Coordinatively Unsaturated Iron(II) Sites. *Nat. Chem.* **2014**, *6* (7), 590–595. <https://doi.org/10.1038/NCHEM.1956>.
- (17) Tolman, W. B. Binding and Activation of N₂O at Transition-Metal Centers: Recent Mechanistic Insights. *Angew. Chem., Int. Ed.* **2010**, *49* (6), 1018–1024.
<https://doi.org/10.1002/ANIE.200905364>.
- (18) Fryzuk, M. D.; Johnson, S. A. The Continuing Story of Dinitrogen Activation. *Coord. Chem. Rev.* **2000**, *200*–*202*, 379–409. [https://doi.org/10.1016/S0010-8545\(00\)00264-2](https://doi.org/10.1016/S0010-8545(00)00264-2).
- (19) Basch, H.; Musaev, D. G.; Morokuma, K. Why Does the Reaction of the Dihydrogen Molecule with [P₂n₂]Zr(μ-H₂-N₂)Zr[P₂n₂] Produce [P₂n₂]Zr(μ-H₂-N₂H)Zr[P₂n₂](μ-H) but Not the Thermodynamically More Favorable [P₂n₂]Zr(μ-NH)₂Zr[P₂n₂]. *J. Am. Chem. Soc.* **1999**, *121* (24), 5754–5761.
<https://doi.org/10.1021/ja984408w>.

- (20) Miessler, G. L.; Fischer, P. J.; Tarr, D. A. *Inorganic Chemistry 5th Ed.*; 2014; Vol. 41.
- (21) Stahl, S. S. *Organotransition Metal Chemistry: From Bonding to Catalysis*; 2010; Vol. 132. <https://doi.org/10.1021/ja103695e>.

Chapter 2

Computational Investigation of the Preferred Binding Modes of N₂O in Group 9 & 10 Metal Complexes

2.1 Introduction

Nitrous oxide (N₂O), a common greenhouse gas, is a thermodynamically powerful and ecologically friendly oxidant that makes it an appealing target for metal center activation.¹

N₂O is a notoriously poor ligand, and its coordination chemistry has been limited to a few terminal, end-on κ -N complexes despite its utility as an oxygen-atom transfer reagent for transition metals.² As a result, nitrous oxide is still being studied as a potentially "green" oxidant in inorganic and organometallic systems. Despite its attraction as an oxidant, N₂O's practical utility is limited by its high kinetic stability. Due to a high activation barrier (ca. 59 kcal/mol), breaking down N₂O into its constituents, N₂ and O₂, requires considerable temperatures (400 °C) in the absence of catalysts.³ In the presence of many transition-metal-based systems, however, N₂O has been well-reported as a highly efficient oxygen-atom transfer reagent.^{4–6} Tolman's has reported multiple potential binding modes of N₂O to a single metal center (Figure 2.1).⁶

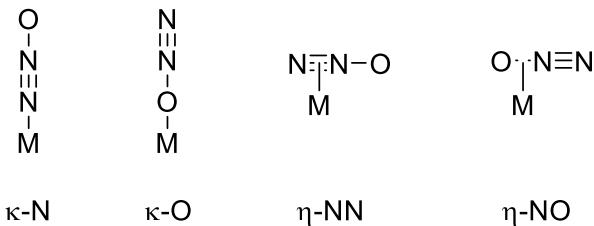


Figure 2.1 Potential monometallic N_2O binding

Characterized complexes of N_2O bond to transition metals are quite rare (Figure 2.2).^{1,7–}

¹² The Lehnert lab used vibrational spectroscopy and computational analysis to demonstrate that the first coordination compound with a N_2O unit, $[\text{Ru}(\text{NH}_3)_5(\text{N}_2\text{O})]^{2+}$, reported by Taube⁷, is a κ -N isomer.⁹ James has also reported a more recent case of a ruthenium octahedral nitrogen-bound N_2O complex found by NMR at low temperatures.¹⁰ X-ray crystallography, NMR spectroscopy, and computational chemistry have been used to describe two nitrogen-bound rhodium(I) pincer complexes.¹¹ X-ray crystallography and vibrational spectroscopy have also been used to characterize a nitrogen-bound copper (I) complex.¹² Another κ -N isomer of N_2O with a vanadium tris-pyrrolide system, which has been confirmed by X-ray crystallography, vibrational spectroscopy, and computational chemistry, has been reported by the Chang lab.¹ The N_2O binds to the metal through the

nitrogen atom in all of these characterized examples of transition metal complexes with N₂O attached to a terminal atom.

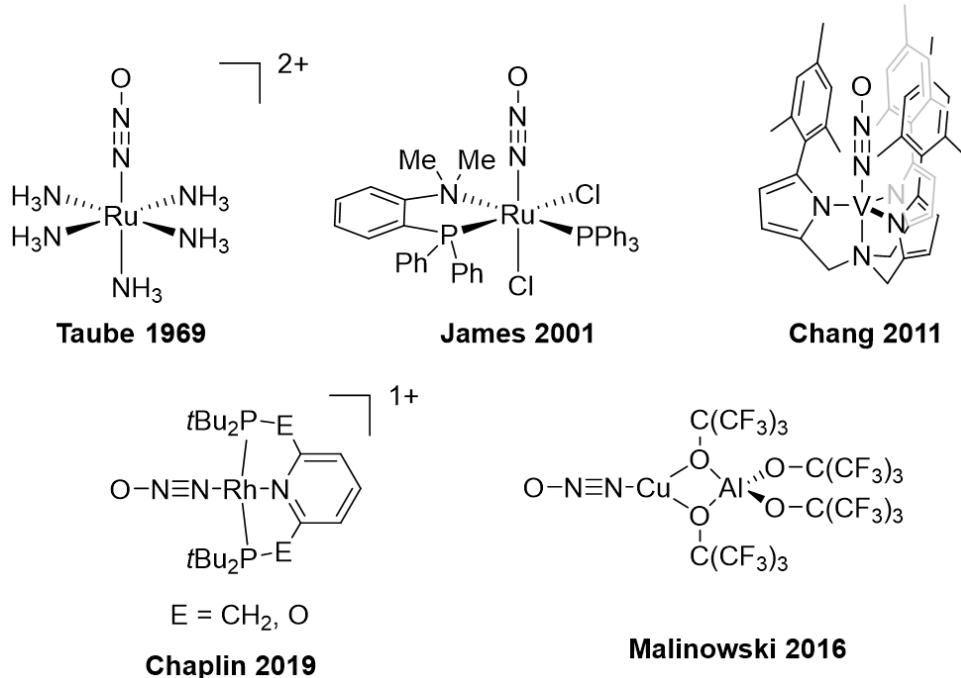


Figure 2.2 Characterized structures with an N₂O unit

The Long lab has reported an iron-based metal-organic framework that can bind N₂O. The N₂O unit has been discovered to bind in both the κ -N and κ -O isomers in this system, and the isomer distribution has been modeled at a 40 percent to 60 percent ratio using powder neutron diffraction.¹³ Recent computational studies of this iron-based metal-organic framework have also suggested that organic molecules settling in the pores of the structure may modify the binding preference.¹⁴

In a previous study, the potential binding effects in a variety of group 8 metal complexes were investigated using density functional theory. Since the N₂O unit is a poor π -acceptor,

it was observed that it preferred to bind in a linear configuration through the terminal nitrogen atom. By changing the nature of the metal center as a π -donor, calculations revealed that the N₂O unit can prefer to bind through the oxygen atom, in which case this binding mode is a σ -donor ligand (Figure 2.3).¹⁵

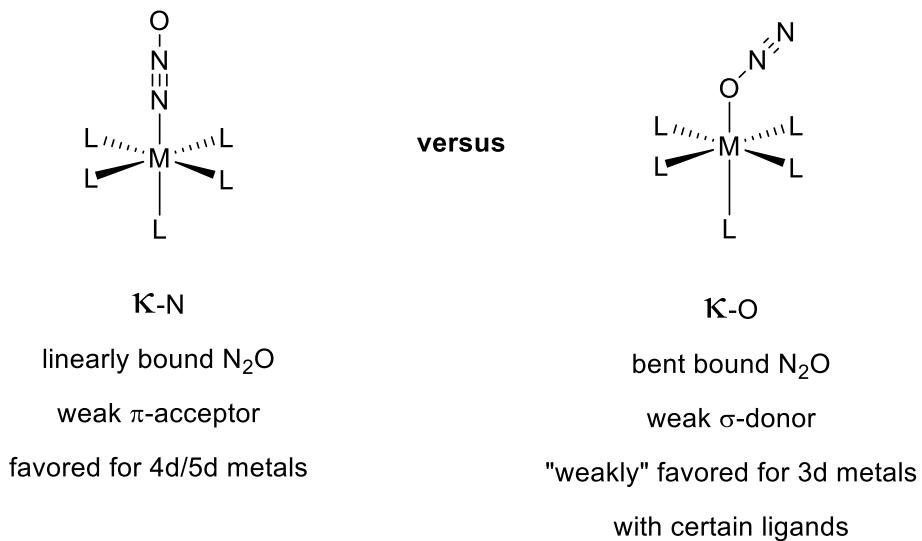


Figure 2.3 N₂O binding modes observed in previous work

In a recent and interesting study by the Roesler lab, a nickel complex possessing a η^2 -NN bound N₂O unit was characterized by X-ray crystallography, IR, and solid-state NMR spectroscopy.¹⁶ Also, the Figueroa lab has reported another η^2 -NN coordination complex of N₂O with a mononuclear cobalt system that has been characterized by IR spectroscopy and computational chemistry (Figure 2.4).² The possible binding effects of a range of group 9 and 10 metal complexes are investigated using density functional theory in this study.

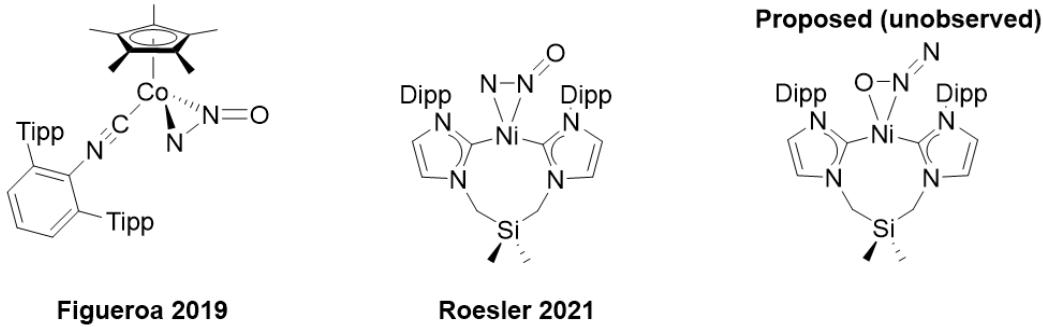


Figure 2.4 Characterized structures with π -bound N_2O unit

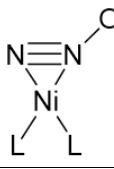
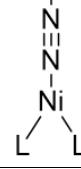
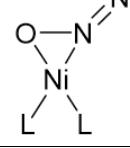
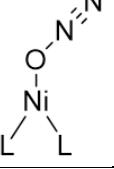
2.2 Computation Details

For all the calculations presented here, the Gaussian16 code was utilized.¹⁷ The level of theory used was B3LYP/6-31G(d). Heavy metals used an effective core potential (ECP) basis set, specifically LANL2DZ. In the gas phase, all singlet spin states were assessed using restricted Kohn-Sham procedures, whereas triplets and quintets were assessed using unrestricted Kohn-Sham techniques. The electronic energy is corrected for enthalpy and entropy using unscaled vibrational frequencies at 1 atm and 298.15 K. By calculating structures with no imaginary frequencies, all stationary points are classified as minima.¹⁵

2.3 Results and Discussion

The possible binding of N_2O to the previously investigated pseudo-square planar Ni system by the Roselar lab was the first comparison made to better understand the binding of N_2O .² Since this approach has been used in the iron, osmium, and ruthenium systems previously studied,¹⁵ the B3LYP/6-31G(d,p) was chosen as the suitable level of theory. With this system, the impact of altering ligands *cis* to the N_2O unit was investigated to determine the influence on binding preference. Interestingly, despite the Roselar lab's

proposal of η^2 -NO as an intermediate, the η^2 -NO binding mode could not be optimized without geometric restrictions of the N₂O unit in four of the five systems. The κ -O binding mode could not be successfully optimized as the structure favored the formation of a metal oxo. In the system with the CO-supporting ligand, all proposed N₂O binding mechanisms were optimized.⁶ The η^2 -NN bound N₂O unit is the energetically favored binding mode with the majority of the ligands, according to the data in Figure 2.5.

					
L=NHC	0.0 kcal/mol	+ 15.6 kcal/mol	IS		oxo
L=PM ₃	0.0 kcal/mol	+ 14.8 kcal/mol	IS		+ 38.1 kcal/mol
L=PH ₃	0.0 kcal/mol	+ 8.9 kcal/mol	IS		+ 30.2 kcal/mol
L=CNMe	0.0 kcal/mol	+ 5.7 kcal/mol	IS		+ 22.3 kcal/mol
L=CO	+ 5.4 kcal/mol	0.0 kcal/mol	+ 18.4 kcal/mol		+ 12.5 kcal/mol

“oxo” represents the formation of a metal oxo due to NO bond breaking and “IS” represents an incorrect structure.

Figure 2.5 Ligand modification effect on binding energy in [Ni(L)₂(N₂O)]

As one progresses from an electron-rich to an electron-deficient metal system through donating supporting ligands, the relative energy difference between η^2 -NN and κ -N decreases. This observation is supported by the vibrational frequencies of the κ -CO and η^2 -C₂H₂ complexes of [Ni(L)₂] as seen in Table 2.1. When the ligand *cis* to the N₂O unit

is substituted with CO, the binding preference favors the κ -N bound N₂O unit by 5.4 kcal/mol.

Table 2.1 C-O and C-C stretch vibrational frequencies (in cm⁻¹) for [Ni(L)₂(CO)] and [Ni(L)₂(CH)₂] complexes

Complex	v _{CO} (cm ⁻¹)	v _{CC} (cm ⁻¹)
[NiC ₁₀ N ₄ H ₁₆]	2009.4	1652.2
[Ni(PMe ₃) ₂]	2036.0	1672.0
[Ni(PH ₃) ₂]	2071.0	1696.4
[Ni(CNMe) ₂] ^a	2061.8	1709.4
[Ni(CO) ₂] ^b	2100.0, 2102.1, 2189.0	1749.9

^aThe C-O stretch vibrational frequency for [Ni(CNMe)₂(CO)] does not follow the trend as the structure breaks planarity. ^bThe [Ni(CO)₂(CO)] complex has three C-O stretch vibrational frequencies due to symmetry coupling.

In an attempt to get a better understanding of the preference between the η^2 -NN and η^2 -NO binding modes, the next system investigated was one similar to the Figueroa lab.² Of the proposed binding modes of N₂O by Tolman, all binding modes were optimized.⁶ The η^2 -NN isomer was calculated to be the most stable in the [CoCp*CNMe] system by 6.2 kcal/mol (Figure 2.6), which is consistent with the Figueroa lab observations in which the η^2 -NN isomer was experimentally characterized and computationally shown to be the most stable isomer.²

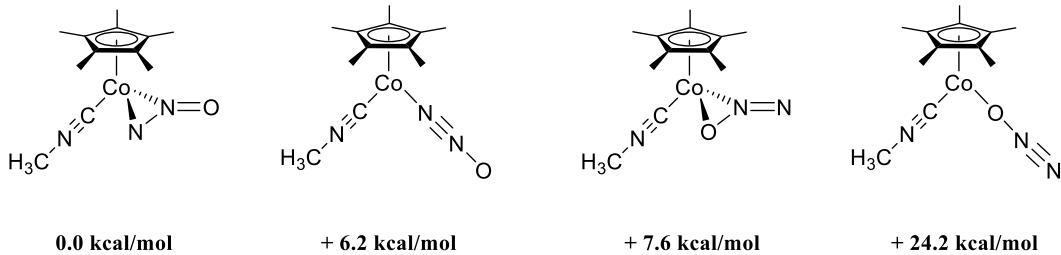
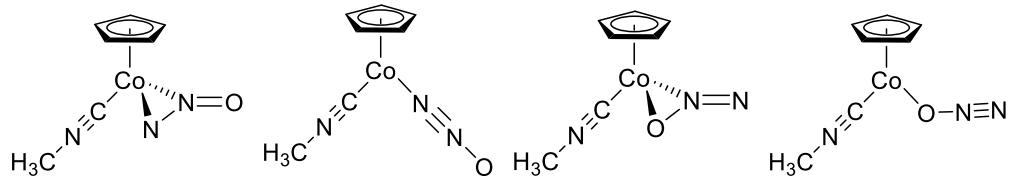


Figure 2.6 Ligand modification effect on binding in $[\text{Co}(\eta^5\text{-Cp}^*)(\text{N}_2\text{O})]$

The pentamethylcyclopentadienyl (Cp^*) ligand was replaced with a cyclopentadienyl (Cp) ligand to see if the electronics of the system altered the $\eta^2\text{-NN}$ or $\kappa\text{-N}$ binding preference. The $\eta^2\text{-NN}$ isomer was determined to be the most stable by 2.9 kcal/mol in the $[\text{CoCpCNMe}]$ system, as shown in Figure 2.7, which is consistent with the previously examined system. The relative energy difference between the preferred binding mode and the subsequent mode diminishes when a supportive ligand is switched from an electron-rich Cp^* to a more electron-deficient Cp . This is comparable to what was observed with the initial Ni system. To evaluate the ligand's influence on the preferred binding mode, the effects of altering the ligand *cis* to the N_2O unit were investigated. The $\eta^2\text{-NN}$ bound N_2O unit is the energetically favored binding mode with the majority of the ligands, as shown by the data in Figure 2.7.



	L=PMe ₃	0.0 kcal/mol	+ 6.6 kcal/mol	+ 7.5 kcal/mol	+ 31.2 kcal/mol
L=PH ₃	0.0 kcal/mol	+ 6.0 kcal/mol	+ 9.2 kcal/mol	+ 26.4 kcal/mol	
L=CNMe	0.0 kcal/mol	+ 2.9 kcal/mol	+ 11.2 kcal/mol	+ 21.0 kcal/mol	
L=CO	+ 3.3 kcal/mol	0.0 kcal/mol	+ 12.5 kcal/mol	+ 17.1 kcal/mol	

Figure 2.7 Ligand modification effect on binding energy in $[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{N}_2\text{O})]$

When the ligand *cis* to the N_2O unit is substituted with CO, however, the binding preference favors the κ -N bound N_2O unit by 3.3 kcal/mol. This is not unexpected considering CO is a strong π -acceptor that efficiently lowers the electron richness of the low valent Co metal center, giving favorability to the κ -N bound binding state. This observation is supported by the vibrational frequencies of the κ -CO and $\eta^2\text{-C}_2\text{H}_2$ complexes of $[\text{Co}(\eta^5\text{-Cp})(\text{L})]$ as seen in Table 2.2, which follows a similar trend seen with the previously investigated Ni systems.

Table 2.2 C-O and C-C stretch vibrational frequencies (in cm^{-1}) for $[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{CO})]$ and $[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{CH}_2)_2]$ complexes

Complex	$\nu_{\text{CO}} (\text{cm}^{-1})$	$\nu_{\text{CC}} (\text{cm}^{-1})$
$[\text{Co}(\eta^5\text{-Cp})(\text{PMe}_3)]$	2032.4	1739.5
$[\text{Co}(\eta^5\text{-Cp})(\text{PH}_3)]$	2049.2	1749.1
$[\text{Co}(\eta^5\text{-Cp})(\text{CNMe})]$	2062.5	1765.2
$[\text{Co}(\eta^5\text{-Cp})(\text{CO})]^{\text{a}}$	2071.2, 2117.6	1780.7

^aThe $[\text{Co}(\eta^5\text{-Cp})(\text{CO})_2]$ complex has two C-O stretch vibrational frequencies due to symmetry coupling.

Several para-substituted pyridine ligands *cis* to the N_2O unit in the $[\text{CoCp}]$ system were studied to determine the influence of preference for $\eta^2\text{-NN}$ or $\kappa\text{-N}$ binding with minimal impact from sterics. According to the results in Figure 2.8, the $\eta^2\text{-NN}$ bound isomer is energetically preferred over the $\kappa\text{-N}$ isomer, independent of the ligand identity. Because the structures formed a metal oxo complex, energy estimates for the O-bound isomers of the $[\text{Co}(\text{Cp})(\text{NMe})_2(\text{N}_2\text{O})]$ and $[\text{Co}(\text{Cp})(\text{NO})_2(\text{N}_2\text{O})]$ systems could not be determined without constraining the N_2O unit.

L= py(p-NMe) ₂	0.0 kcal/mol	+ 9.7 kcal/mol	oxo	oxo
L=py	0.0 kcal/mol	+ 9.0 kcal/mol	oxo	+ 30.3 kcal/mol
L= py(p-NO) ₂	0.0 kcal/mol	+ 5.3 kcal/mol	+ 6.6 kcal/mol	+ 21.6 kcal/mol

“oxo” represents the formation of a metal oxo due to NO bond breaking.

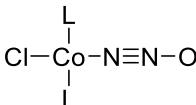
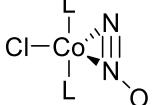
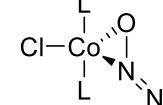
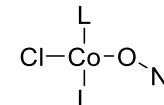
Figure 2.8 Para-substituted pyridine ligand modification effect on binding energy in $[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{N}_2\text{O})]$

The metal center was then exchanged with other metals in the $[[\text{M}] \text{Cp}(\text{CNMe})(\text{N}_2\text{O})]$ complexes to understand how it affected the binding preference (Figure 2.9). Interestingly with Ir and Rh, the preferred binding mode was calculated to be the $\kappa\text{-N}$ isomer with a linear orientation.

M=Co	0.0 kcal/mol	+ 2.9 kcal/mol	+ 11.2 kcal/mol	+ 21.0 kcal/mol
M-Rh	+ 6.0 kcal/mol	0.0 kcal/mol	+ 14.2 kcal/mol	+ 11.4 kcal/mol
M=Ir	+ 3.5 kcal/mol	0.0 kcal/mol	+ 14.9 kcal/mol	+ 16.2 kcal/mol

Figure 2.9 Metal identity modification effect on $\eta\text{-NN}$ binding in $[[\text{M}] \text{CNMe}(\eta^5\text{-Cp})(\text{N}_2\text{O})]$

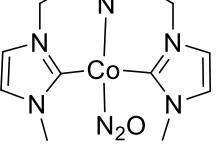
A series of square planar $[\text{Co}(\text{L})_2\text{Cl}(\text{N}_2\text{O})]$ complexes were investigated using a variety of ligands (L) in an effort to reduce the relative energy gap between the $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ binding modes using the observations made (Figure 2.10). Similar to the observations in the $[\text{CoCp}(\text{L})_2(\text{N}_2\text{O})]$ and $[\text{Ni}(\text{L})_2(\text{N}_2\text{O})]$ systems, it is evident that the relative energy difference between $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ reduces as one moves from an electron-rich to an electron-deficient system. The $\kappa\text{-O}$ binding modes for 4 out of the 5 systems investigated could not be obtained without geometric restrictions of the N_2O unit, as they originally optimized to the $\eta^2\text{-NO}$ binding mode. These complexes are indicated by the symbol "IS". By continuing with a pseudo-square planar geometry, similar to the Figueroa lab system, and making the metal center more electron-rich, the relative energy difference between $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ binding modes was decreased. It should also be noted that the second most favored binding mode of N_2O is $\eta^2\text{-NO}$ in both the $[\text{Co}(\text{PH}_3)_2(\text{N}_2\text{O})\text{Cl}]$ and $[\text{Co}(\text{PMe}_3)_2(\text{N}_2\text{O})\text{Cl}]$ systems, which has not been seen before. Previously, $\kappa\text{-N}$ and $\eta^2\text{-NN}$ have been the two most favored binding modes.

					
L=PM ₃	+ 7.8 kcal/mol	0.0 kcal/mol	+ 2.5 kcal/mol		IS
L=PH ₃	+ 3.4 kcal/mol	0.0 kcal/mol	+ 3.1 kcal/mol		IS
L=CNMe	0.0 kcal/mol	+ 1.0 kcal/mol	+ 5.6 kcal/mol		IS
L=CO	0.0 kcal/mol	+ 9.1 kcal/mol	+ 13.5 kcal/mol	+ 12.7 kcal/mol	

“IS” represents an incorrect structure.

Figure 2.10 Ligand identity modification effect on binding energy in $[[\text{Co}](\text{L})_2\text{Cl}(\text{N}_2\text{O})]$

In view of the decrease in the relative energy difference between not only $\eta^2\text{-NN}$ and $\kappa\text{-N}$ but also $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ by changing the geometry and electronics of the metal system, more attempts were made to locate a system in which $\eta^2\text{-NO}$ was the favored binding mode. To perhaps shift the electronics in favor of the $\eta^2\text{-NO}$ binding mode while maintaining the pseudo square planar geometry, the PMe₃ supporting ligands were substituted for an N-heterocyclic carbene (NHC). Figure 2.11's results over $[\text{Co}(\text{C}_4\text{N}_2\text{H}_5)_2(\text{NC}_4\text{H}_8)(\text{N}_2\text{O})]$ show that the $\eta^2\text{-NN}$ binding mode is still the energetically preferred one by 0.9 kcal/mol. The $\kappa\text{-O}$ binding mode could not be successfully optimized as the structure favored the formation of a metal oxo.

		$\eta^2\text{-NN}$	$\eta^2\text{-NO}$	
		+ 8.3 kcal/mol	0.0 kcal/mol	+ 0.9 kcal/mol
				oxo

“oxo” represents the formation of a metal oxo due to NO bond breaking.

Figure 2.11 Ligand modification effect on binding in $[\text{Co}(\text{C}_4\text{N}_2\text{H}_5)_2(\text{NC}_4\text{H}_8)(\text{N}_2\text{O})]$

The twisting of the pincer ligand in the $[\text{Co}(\text{C}_4\text{N}_2\text{H}_5)_2(\text{NC}_4\text{H}_8)(\text{N}_2\text{O})]$ system is shown in Figure 2.12a. This resulted in the N_2O unit being driven up out of plane by the steric bulk of the methyl groups, which significantly reduced the relative energy disparity between the $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ binding modes. In correspondence to this observation, the methyl groups on the NHCs were altered to t-butyl in a last attempt to pinpoint a system in which $\eta^2\text{-NO}$ is the favored binding mode. As seen in Figure 2.12b, the substitution of methyl groups for t-butyl groups caused the pincer ligand to twist further, and in turn, caused the N_2O unit to rise higher out of plane.

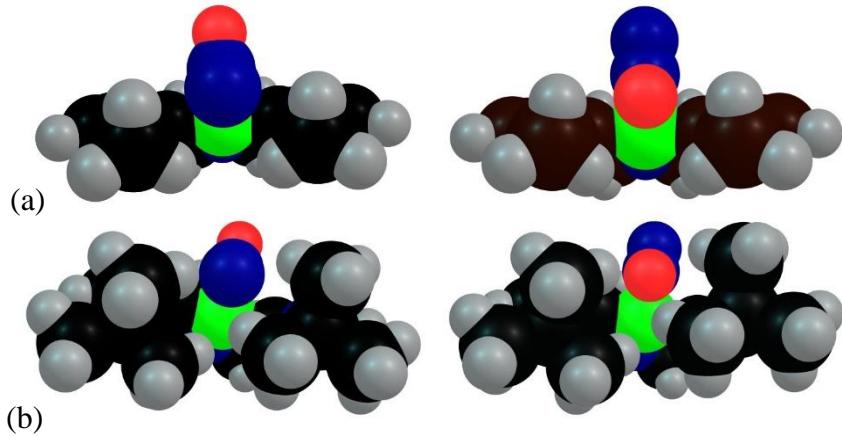
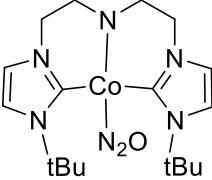
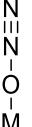


Figure 2.12 (a) Space-filling model of $[\text{Co}(\text{C}_4\text{N}_2\text{H}_5)_2(\text{NC}_4\text{H}_8)]$ in both the $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ configuration, (b) Space-filling model of $[\text{Co}(\text{C}_6\text{N}_2\text{H}_{11})_2(\text{NC}_4\text{H}_8)]$ in both the $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ configuration

Color Scheme: green = cobalt, blue = nitrogen, red = oxygen, black = carbon, grey = hydrogen

Figure 2.13's results over $[\text{Co}(\text{C}_6\text{N}_2\text{H}_{11})_2(\text{NC}_4\text{H}_8)(\text{N}_2\text{O})]$ show that the $\eta^2\text{-NO}$ is in energetic equilibrium with the $\eta^2\text{-NN}$ binding mode. It should be noted that for “consistency”, the values are reported to one decimal place, making the $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ appear in energetic equilibrium. However, these two binding modes vary by 0.04 kcal/mol, giving preference to the $\eta^2\text{-NO}$ binding mode, which has not previously been seen. This being said, the 0.04 kcal/mol difference between the $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ binding modes falls into the margin of error for density functional theory (DFT) calculations.

				
	+ 5.4 kcal/mol	0.0 kcal/mol	0.0 kcal/mol	oxo

“oxo” represents the formation of a metal oxo due to NO bond breaking.

Figure 2.13 Ligand modification effect on binding in $[\text{Co}(\text{C}_6\text{N}_2\text{H}_{11})_2(\text{NC}_4\text{H}_8)(\text{N}_2\text{O})]$

2.4 Conclusion

The N_2O unit often prefers to be bonded through both nitrogens in a bent configuration to low valent metal centers, according to this study's use of computational chemistry. This favored binding mode is associated with a low valent metal's strong π -donor characteristics. Alternately, by increasing the electron deficiency of the metal complexes, this binding preference can be switched to the $\kappa\text{-N}$ binding mode. It was interestingly observed that the geometry and electronics of the structure also had an impact on the preference between the $\eta^2\text{-NN}$ and $\eta^2\text{-NO}$ binding modes. Binding between the nitrogen and oxygen atoms of the N_2O can be favored by utilizing a pseudo-square planar geometry, similar to the Figueroa lab system, and making the metal center more electron-rich. These insights aid in understanding the known binding properties of N_2O , opening the door for further research into the chemistry of this peculiar and potentially major oxidant.

2.5 References

- (1) Piro, N. A.; Lichterman, M. F.; Harman, W. H.; Chang, C. J. A Structurally Characterized Nitrous Oxide Complex of Vanadium. *J. Am. Chem. Soc.* **2011**, *133* (7), 2108–2111. <https://doi.org/10.1021/JA110798W>.
- (2) Mokhtarezadeh, C. C.; Chan, C.; Moore, C. E.; Rheingold, A. L.; Figueroa, J. S. Side-On Coordination of Nitrous Oxide to a Mononuclear Cobalt Center. *J. Am. Chem. Soc.* **2019**, *141* (38), 15003–15007. <https://doi.org/10.1021/JACS.9B08241>.
- (3) Volman, D. H. Photochemistry of Small Molecules. *J. Photochem.* **1984**, *25* (1), 15–19. [https://doi.org/10.1016/0047-2670\(84\)85004-2](https://doi.org/10.1016/0047-2670(84)85004-2).
- (4) Masaya Matsuoka; Woo-Sung Ju; Kenzo Takahashi; Hiromi Yamashita, and; Anpo*, M. Photocatalytic Decomposition of N₂O into N₂ and O₂ at 298 K on Cu(I) Ion Catalysts Anchored onto Various Oxides. The Effect of the Coordination State of the Cu(I) Ions on the Photocatalytic Reactivity. *J. Phys. Chem. B* **2000**, *104* (20), 4911–4915. <https://doi.org/10.1021/JP9940001>.
- (5) Severin, K. Synthetic Chemistry with Nitrous Oxide. *Chem. Soc. Rev.* **2015**, *44* (17), 6375–6386. <https://doi.org/10.1039/C5CS00339C>.
- (6) Tolman, W. B. Binding and Activation of N₂O at Transition-Metal Centers: Recent Mechanistic Insights. *Angew. Chem., Int. Ed.* **2010**, *49* (6), 1018–1024. <https://doi.org/10.1002/ANIE.200905364>.
- (7) Armor, J. N.; Taube, H. Formation and Reactions of [(NH₃)₅RuN₂O₂⁺]. *J. Am. Chem. Soc.* **1969**, *91* (24), 6874–6876. <https://doi.org/10.1021/JA01052A069>.
- (8) Walstrom, A.; Pink, M.; Fan, H.; Tomaszewski, J.; Caulton, K. G. Radical (NO) and Nonradical (N₂O) Reagents Convert a Ruthenium(IV) Nitride to the Same Nitrosyl Complex. *Inorg. Chem.* **2007**, *46* (19), 7704–7706. <https://doi.org/10.1021/IC700789Y>.
- (9) Paulat, F.; Kuschel, T.; Näther, C.; Praneeth, V. K. K.; Sander, O.; Lehnert, N. Spectroscopic Properties and Electronic Structure of Pentammineruthenium(II) Dinitrogen Oxide and Corresponding Nitrosyl Complexes: Binding Mode of N₂O and Reactivity. *Inorg. Chem.* **2004**, *43* (22), 6979–6994. <https://doi.org/10.1021/IC049302I>.

- (10) Pamplin, C. B.; Ma, E. S. F.; Safari, N.; Rettig, S. J.; James, B. R. The Nitrous Oxide Complex, RuCl₂(H₁-N₂O)(P-N)(PPh₃) (P-N = [o-(N,N-Dimethylamino)Phenyl]Diphenylphosphine); Low Temperature Conversion of N₂O to N₂ and O₂ [2]. *J. Am. Chem. Soc.* **2001**, *123* (35), 8596–8597.
<https://doi.org/10.1021/JA0106319>.
- (11) Gyton, M. R.; Leforestier, B.; Chaplin, A. B. Rhodium(I) Pincer Complexes of Nitrous Oxide. *Angew. Chem., Int. Ed.* **2019**, *58* (43), 15295–15298.
<https://doi.org/10.1002/ANIE.201908333>.
- (12) Zhuravlev, V.; Malinowski, P. J. A Stable Crystalline Copper(I)–N₂O Complex Stabilized as the Salt of a Weakly Coordinating Anion. *Angew. Chem., Int. Ed.* **2018**, *57* (36), 11697–11700. <https://doi.org/10.1002/ANIE.201806836>.
- (13) Xiao, J. D. ; et. al. Oxidation of Ethane to Ethanol by N₂O in a Metal-Organic Framework with Coordinatively Unsaturated Iron(II) Sites. *Nat. Chem.* **2014**, *6* (7), 590–595. <https://doi.org/10.1038/NCHEM.1956>.
- (14) Suh, B. L.; Kim, J. Ligand Insertion in MOF-74 as Effective Design for Oxidation of Ethane to Ethanol. *J. Phys. Chem. C* **2018**, *122* (40), 23078–23083.
<https://doi.org/10.1021/ACS.JPCC.8B07857>.
- (15) Fields, K.; Barngrover, B. M.; Gary, J. B. Computational Investigation of the Preferred Binding Modes of N₂O in Group 8 Metal Complexes. *Inorg. Chem.* **2020**, *59* (24), 18314–18318. <https://doi.org/10.1021/acs.inorgchem.0c02903>.
- (16) Lombardi, B. M. P.; Gendy, C.; Gelfand, B. S.; Bernard, G. M.; Wasylissen, R. E.; Tuononen, H. M.; Roesler, R. Side-on Coordination in Isostructural Nitrous Oxide and Carbon Dioxide Complexes of Nickel. *Angew. Chem., Int. Ed.* **2021**, *60* (13), 7077–7081. <https://doi.org/10.1002/ANIE.202011301>.
- (17) Frisch, M. J. ; et. al. Gaussian 16, Revision A.03. Gaussian Inc.: Wallingford, CT 2009.

Chapter 3

Computational Investigation into the Activation of N₂ Through Binding in Transition Metal Complexes

3.1 Introduction

One of the essential elements for the existence of all living things is nitrogen. It is a need for the formation of several classes of biomolecules, such as proteins, DNA, and chlorophyll. Although nitrogen is present in the atmosphere in huge quantities as dinitrogen gas (N₂), most organisms cannot access it in this form, making it a limited resource and frequently restricting primary production in many ecosystems. Nitrogen is not readily accessible to primary producers like plants until it is changed from dinitrogen gas into ammonia (NH₃).¹

Along with N₂ and NH₃, nitrogen may also take on inorganic (such as nitrate) and organic (such as amino and nucleic acids) forms. Thus, when organisms use nitrogen for development and, in certain circumstances, energy, it passes through a variety of distinct changes throughout the environment. Nitrogen is transformed primarily by nitrogen fixation, nitrification, denitrification, ammonification, and anammox (Figure 3.1). The productivity of the biosphere depends heavily on the oxidation of nitrogen into its many oxidation states, which is carried out by a wide variety of microorganisms including bacteria, archaea, and fungi.¹

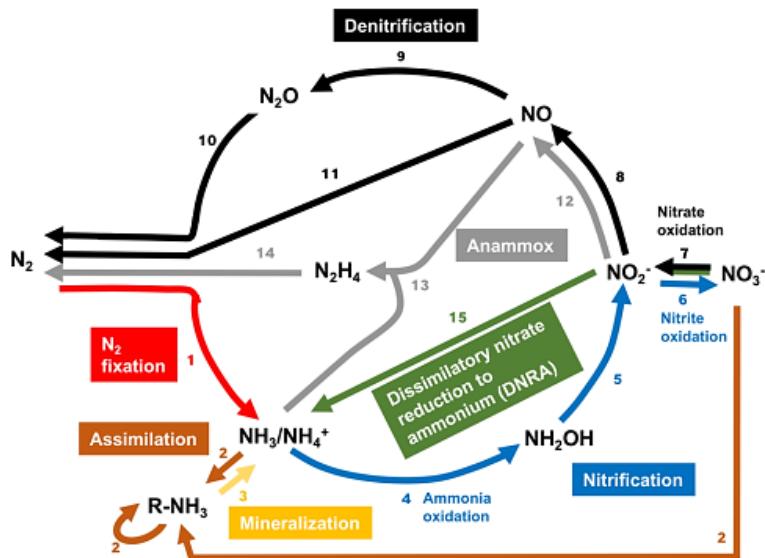


Figure 3.1 Major transformations in the nitrogen cycle

Diazotrophic bacteria carry out biological nitrogen fixation, which is the conversion of N_2 into two NH_3 molecules, an accessible source of nitrogen. These specialized microbes can exist independently or in symbiotic partnerships with specific animals and plants, such as legumes (e.g., termites).¹ Before the development of the Haber-Bosch process for the synthesis of commercial ammonia, these microorganisms were the primary source of the majority of nitrogen atoms in living things. They possess nitrogenase enzymes, which are the only kind of enzymes capable of carrying out the crucial N_2 -reducing process.^{2–12} Despite being an exergonic process under ambient conditions, the full reduction of atmospheric N_2 gas to two molecules of NH_3 is kinetically unfavorable due to the high energy barrier for activating the inert N_2 triple bond.^{13–17}

The well-known Haber-Bosch process, which sustains the reaction of N_2 and H_2 on an iron catalyst at high temperature and pressure, is the technical answer to dinitrogen

fixation.¹⁸ Due to the process, ammonia fertilizer became widely available, which contributed to a rise in the global population by hastening the rate at which agricultural yields rose. Unfortunately, although being a significant technological development, the Haber-Bosch process has always been energy-intensive, leaving a significant carbon footprint. The reaction consumes around 1% of the world's total energy production while operating at temperatures of around 500 °C and pressures of up to 20 MPa.^{19,20} According to the Institute for Industrial Productivity, it released up to 451 million tons of CO₂ in 2010. More than any other industrial chemical-making process, this sum is responsible for around 1.4% of the world's yearly CO₂ emissions.^{19,20} On the other hand, biological nitrogen fixation is limited to ambient circumstances and the usage of ATP, a biological energy source. At the active site of the proteins, metal ions like molybdenum and iron are often present, which are essential for activating N₂.^{21–28} Metalloenzyme nitrogenase, which is made up of the electron-donating Fe cluster and the catalytic MoFe active site, serves as a catalyst for biological N₂ fixation. N₂ is reduced at the FeMo-cofactor (FeMo-co) active

site of the MoFe protein, [7Fe-9S-Mo-C-homocitrate], on the particular 4Fe-face picture in Figure 3.2.²

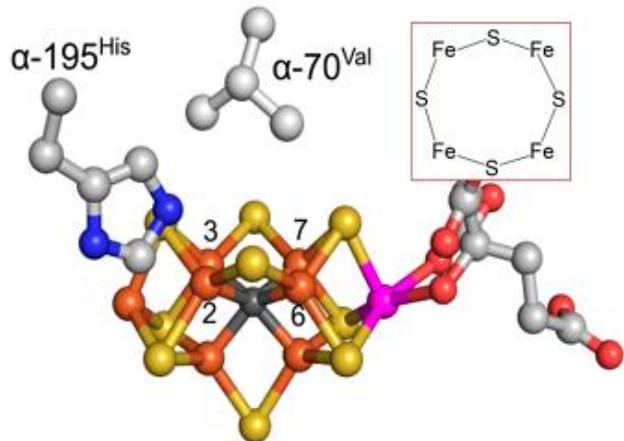


Figure 3.2 FeMo-co, [7Fe-9S-Mo-C-homocitrate], with a cartoon of active Fe 2,3,6,7 FeMo-co face. PDB: 2AFI

Exploring the most efficient catalysts for use in industry while drawing lessons from nature has proven to be one of the biggest obstacles. Numerous investigations have been done on metal-containing catalysts for nitrogen fixation at mild temperatures and pressures (298 K and 1 atm), where the metallic elements play important roles.^{22,29-41}

The reduction of N₂ to ammonia using molybdenum catalysts that include tetradeinate [HIPTN₃N]³⁻ triamidoamine ligands discovered by X-ray crystallography was reported by the Schrock lab at room temperature and 1 atmosphere.³⁷ The Peters team also reported the structurally and computationally supported reduction of N₂ to ammonia through the Fe complex with tris(phosphino)alkyl (CP^{iPr}₃) ligand containing an axial carbon donor.⁴² A report on terminal vs bridging end-on N₂ coordination in transition metal complexes was published by the Goldman, Miller, and Holland laboratories. They use both quantitative

DFT and qualitative molecular orbital (MO) analyses to elucidate the fundamental factors governing the two binding modes and determine which is preferred for a given metal-ligand system. They concluded that an electronic factor, such as the number of π -bonds in the triatomic MNN moiety of an MNN complex, plays a critical role in determining these trends.⁴⁰

In a prior, yet intriguing, investigation, the Lappert group used IR and ESR spectroscopy to analyze a zirconium complex that contained a η^2 -NN bound N₂ unit.⁴³ The Sutton group has reported side-bound coordination of N₂ with a mononuclear rhenium system that has been characterized by NMR spectroscopy.⁴⁴ A η^2 -NN coordination complex of N₂ has been reported by the Coppens lab using the photocrystallographic method, which enables the study of metastable or transient species formed by *in situ* laser irradiation of diffractometer-mounted samples at low temperatures. These findings have been supported by low-temperature IR and Differential Scanning Calorimetry (DSC) measurements as well as by quantum mechanical calculations.⁴⁵ The Bercaw lab has reported the discovery of yet another η^2 -NN bound N₂ unit with a mononuclear titanium system that has been characterized by IR and NMR spectroscopy.⁴⁶ We hope to shed some insight on yet another significant small molecule and its binding and activation towards bond reduction with transition metal complexes based on our earlier research on terminal vs side-on N₂O coordination in transition metal complexes.

3.2 Computation Details

For all the calculations presented here, the Gaussian16 code was utilized.⁴⁷ The level of theory used was B3LYP/6-31G(d). Heavy metals used an effective core potential (ECP) basis set, specifically LANL2DZ. In the gas phase, all singlet spin states were assessed using restricted Kohn-Sham procedures, whereas triplets and quintets were assessed using unrestricted Kohn-Sham techniques. The electronic energy is corrected for enthalpy and entropy using unscaled vibrational frequencies at 1 atm and 298.15 K. By calculating structures with no imaginary frequencies, all stationary points are classified as minima.⁴⁸

3.3 Results and Discussion

The initial systems investigated involved the potential binding of N₂ to the previously studied [Ni(L)₂] in our lab. These systems were chosen as they are known to bind small molecules in both the terminal and side-bound coordination, as seen with the Roselar lab's work with N₂O.⁴⁹ This initial comparison, which was done to better understand the binding and activation of N₂, included a pseudo-trigonal planar Ni system. The B3LYP/6-31G(d,p) level of theory was selected as it has been employed in the iron, osmium, and ruthenium systems previously studied.⁴⁸ With this system, the effect of changing the ligands' *cis* to the N₂ unit was examined to assess its impact on bond activation and binding. The findings in Figure 3.3 show that the energetically preferred binding mode for all of the ligands is the κ -N bound N₂ unit.

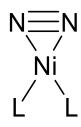
		
L=NHC	0.0 kcal/mol	+3.3 kcal/mol
L=PMe3	0.0 kcal/mol	+1.7 kcal/mol
L=PH3	0.0 kcal/mol	+3.8 kcal/mol
L=CNMe	0.0 kcal/mol	+4.7 kcal/mol
L=CO	0.0 kcal/mol	+6.8 kcal/mol

Figure 3.3 Ligand modification effect on binding energy in $[\text{Ni}(\text{L})_2(\text{N}_2)]$

The relative energy difference between $\eta^2\text{-NN}$ and $\kappa\text{-N}$ increases as one moves from an electron-rich to an electron-deficient metal system through donating supporting ligands. The NHC ligand deviates from the trend because of the steric encumbrance in the complex. This observation is supported by the vibrational frequencies of the $\kappa\text{-CO}$ and $\eta^2\text{-C}_2\text{H}_2$ complexes of $[\text{Ni}(\text{L})_2]$ as seen in Table 3.1. It can be seen from the data that with this Ni^0 system, the relative energy between the $\eta^2\text{-NN}$ and $\kappa\text{-N}$ binding modes is fairly small, indicating that the $\eta^2\text{-NN}$ binding mode on N_2 is achievable, as seen with the Figueroa and Roselar labs.

Table 3.1 C-O and C-C stretch vibrational frequencies (in cm^{-1}) for $[\text{Ni}(\text{L})_2(\text{CO})]$ and $[\text{Ni}(\text{L})_2(\text{CH})_2]$ complexes

Complex	$\nu_{\text{CO}} (\text{cm}^{-1})$	$\nu_{\text{CC}} (\text{cm}^{-1})$
$[\text{NiC}_{10}\text{N}_4\text{H}_{16}]$	2009.4	1652.2
$[\text{Ni}(\text{PMe}_3)_2]$	2036.0	1672.0
$[\text{Ni}(\text{PH}_3)_2]$	2071.0	1696.4
$[\text{Ni}(\text{CNMe})_2]^{\text{a}}$	2061.8	1709.4
$[\text{Ni}(\text{CO})_2]^{\text{b}}$	2100.0, 2102.1, 2189.0	1749.9

^aThe C-O stretch vibrational frequency for $[\text{Ni}(\text{CNMe})_2(\text{CO})]$ does not follow the trend as the structure breaks planarity. ^bThe $[\text{Ni}(\text{CO})_2(\text{CO})]$ complex has three C-O stretch vibrational frequencies due to symmetry coupling.

In this $[\text{Ni}(\text{L})_2(\text{N}_2)]$ series, the N-N bond distance decreases from 1.13 Å in the terminal bound $[\text{Ni}(\text{NHC})_2(\text{N}_2)]$ complex to 1.12 Å in the terminal bound $[\text{Ni}(\text{CO})_2(\text{N}_2)]$ complex while the N-N stretching frequency in the terminal bound $[\text{Ni}(\text{NHC})_2(\text{N}_2)]$ complex increases from 2194.0 cm^{-1} to 2331.4 cm^{-1} in the terminal bound $[\text{Ni}(\text{CO})_2(\text{N}_2)]$ complex. Similar observations can be seen with the side-bound N_2 unit (Table 3.2), in which the N-N bond distance decreases from 1.18 Å in the side-bound $[\text{Ni}(\text{NHC})_2(\text{N}_2)]$ complex to 1.15 Å in the side-bound $[\text{Ni}(\text{CO})_2(\text{N}_2)]$ complex while the N-N stretching frequency in the side-bound $[\text{Ni}(\text{NHC})_2(\text{N}_2)]$ complex increases from 1882.9 cm^{-1} to 2068.3 cm^{-1} in the side-bound $[\text{Ni}(\text{CO})_2(\text{N}_2)]$ complex. This demonstrates that the N-N bond distance decreases as the N-N stretching frequency increases when a supporting ligand is switched from an electron-rich to an electron-deficient alternative.

Table 3.2 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm⁻¹) for [Ni(L)₂(N₂)] complexes

[Ni(L) ₂ (N ₂)]	κ-N	η²-NN		
	N-N (Å)	v _{NN} (cm ⁻¹)	N-N (Å)	v _{NN} (cm ⁻¹)
L=NHC	1.13	2194.0	1.18	1882.9
L=PM ₃	1.13	2217.7	1.17	1917.5
L=PH ₃	1.13	2249.1	1.17	1955.3
L=CNMe	1.12	2217.9	1.16	1988.6
L=CO	1.12	2331.4	1.15	2068.3

The next system examined, which was another that has been shown to bind small molecules in both the terminal and side-bound coordination, was comparable to the Figueroa lab to gain a better grasp of the η²-NN and κ-N binding modes.⁵⁰ According to calculations, the κ-N isomer is 8.6 kcal/mol more stable in the [Co(η⁵-Cp*)(CNMe)] system (Figure 3.4).

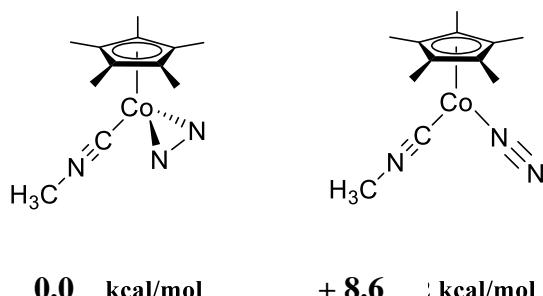


Figure 3.4 Ligand modification effect on binding in [Co(η⁵-Cp*)(CNMe)(N₂)]

In this $[\text{Co}(\eta^5\text{-Cp}^*)(\text{CNMe})(\text{N}_2)]$ series, the N-N bond distance increases from 1.13 Å in the terminal N_2 bound complex to 1.16 Å in the side-bound N_2 complex (Table 3.3). From this, it can be seen that the N-N stretching frequency in the terminal N_2 bound complex decreases from 2257.7 cm^{-1} to 2003.7 cm^{-1} in the side-bound N_2 complex. The relative energy difference and N-N stretching frequency in the preceding Ni system are less than those in the $[\text{Co}(\eta^5\text{-Cp}^*)(\text{CNMe})(\text{N}_2)]$ system, indicating that this system does not activate the N_2 bond as much as the previous systems.

Table 3.3 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm^{-1}) for $[\text{Co}(\eta^5\text{-Cp}^*)(\text{CNMe})(\text{N}_2)]$

$[\text{Co}(\text{Cp}^*)(\text{CNMe})(\text{N}_2)]$	$\kappa\text{-N}$ N-N (Å)	$\eta^2\text{-NN}$ N-N (Å)	ν_{NN} (cm^{-1})	ν_{NN} (cm^{-1})
CNMe	1.13	1.16	2257.7	2003.7

To determine if the electronics of the system changed the binding of $\eta^2\text{-NN}$ or $\kappa\text{-N}$ as well as N_2 activation, the cyclopentadienyl (Cp) ligand was substituted with the pentamethylcyclopentadienyl (Cp^*) ligand. In the $[\text{CoCpCNMe}]$ system, as seen in Figure 3.5, the $\kappa\text{-N}$ isomer was shown to be the most stable by 12.5 kcal/mol. When a supporting ligand is changed from an electron-rich to a more electron-deficient alternative, the relative energy gap between the favored binding mode and the succeeding mode increases. Similar results were seen with the original Ni system. The impacts of changing the ligand's *cis* to the N_2 unit were examined to assess the ligand's impact on the binding mode and N_2

activation. According to the results in Figure 3.5, the κ -N bound N_2 unit is the energetically preferred binding form with all of the ligands. Due to the ligand's broken linearity, which caused some distortion in the relative energy, the CNMe supporting ligand did not follow the trend.

L=PMe ₃	0.0 kcal/mol	+ 9.7 kcal/mol
L=PH ₃	0.0 kcal/mol	+ 9.9 kcal/mol
L=CNMe	0.0 kcal/mol	+ 12.5 kcal/mol
L=CO	0.0 kcal/mol	+ 10.7 kcal/mol

Figure 3.5 Ligand modification effect on binding energy in $[Co(\eta^5\text{-Cp})(L)(N_2)]$

The binding preference varies by 10.7 kcal/mol when CO is replaced for the ligand *cis* to the N_2 unit. This is not unexpected considering CO is a strong π -acceptor that efficiently lowers the electron richness of the low valent Co metal center, giving favorability to the κ - N bound binding mode. The vibrational frequencies of the κ -CO and η^2 - C_2H_2 complexes of $[Co(\eta^5\text{-Cp})(L)]$ shown in Table 3.4 complement this observation and show a similar pattern to the Ni systems that were previously studied.

Table 3.4 C-O and C-C stretch vibrational frequencies (in cm^{-1}) for $[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{CO})]$ and $[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{CH}_2)_2]$ complexes

Complex	$\nu_{\text{CO}} (\text{cm}^{-1})$	$\nu_{\text{CC}} (\text{cm}^{-1})$
$[\text{Co}(\eta^5\text{-Cp})(\text{PMe}_3)]$	2032.4	1739.5
$[\text{Co}(\eta^5\text{-Cp})(\text{PH}_3)]$	2049.2	1749.1
$[\text{Co}(\eta^5\text{-Cp})(\text{CNMe})]$	2062.5	1765.2
$[\text{Co}(\eta^5\text{-Cp})(\text{CO})]^{\text{a}}$	2071.2, 2117.6	1780.7

^aThe $[\text{Co}(\eta^5\text{-Cp})(\text{CO})_2]$ complex has two C-O stretch vibration frequencies due to symmetry coupling.

In this $[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{N}_2)]$ series, the N-N bond distance decreases from 1.13 Å in the terminal bound $[\text{Co}(\eta^5\text{-Cp})(\text{PMe}_3)(\text{N}_2)]$ complex to 1.12 Å in the terminal bound $[\text{Co}(\eta^5\text{-Cp})(\text{CO})(\text{N}_2)]$ complex, similar observations can be seen with the side-bound N_2 unit (Table 3.5). From this, it can be seen that the N-N stretching frequency in the terminal bound $[\text{Co}(\eta^5\text{-Cp})(\text{PMe}_3)(\text{N}_2)]$ complex increases from 2226.8 cm^{-1} to 2294.9 cm^{-1} in the terminal bound $[\text{Co}(\eta^5\text{-Cp})(\text{CO})(\text{N}_2)]$ complex, similar observations can be seen with the side-bound N_2 coordination. It can be seen that changing the electronics of the metal system had quite an effect on N_2 activation, however, this observation is complicated by the bending of the CNMe.

Table 3.5 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm⁻¹) for [Co(η^5 -Cp)(L)(N₂)]

[Co(Cp)(L)(N ₂)]	κ -N N-N (Å)	v _{NN} (cm ⁻¹)	N-N (Å)	η^2 -NN v _{NN} (cm ⁻¹)
L=PM ₃	1.13	2226.8	1.16	1975.7
L=PH ₃	1.13	2244.4	1.16	1992.8
L=CNMe	1.12	2274.7	1.15	2022.3
L=CO	1.12	2294.9	1.15	2048.9

In order to ascertain the influence that the electronics have on η^2 -NN or κ -N binding as well as N₂ activation with consistent steric influence, a number of para-substituted pyridine ligands *cis* to the N₂ unit in the [Co(η^5 -Cp)(L)(N₂)] system were examined. According to the findings in Figure 3.6, regardless of the kind of ligand, the κ -N bound isomer is energetically favored over the η^2 -NN isomer.

L= py-(p-NMe) ₂	0.0 kcal/mol	+6.2 kcal/mol
L=py	0.0 kcal/mol	+6.7 kcal/mol
L= py-(p-NO) ₂	0.0 kcal/mol	+8.4 kcal/mol

Figure 3.6 Ligand modification effect on binding energy in [Co(η^5 -Cp)(L)(N₂)]

The N-N bond distance reduces in this $[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{N}_2)]$ series from 1.13 Å in the terminal bound $[\text{Co}(\eta^5\text{-Cp})\text{py-(p-NMe)}_2(\text{N}_2)]$ complex to 1.13 Å in the terminal bound $[\text{Co}(\eta^5\text{-Cp})\text{py-(p-NO)}_2(\text{N}_2)]$ complex. Similar findings may be made with the side-bound N_2 unit (Table 3.6). In contrast, the terminal bound $[\text{Co}(\eta^5\text{-Cp})\text{py-(p-NO)}_2(\text{N}_2)]$ complex has a higher N-N stretching frequency than the terminal bound $[\text{Co}(\eta^5\text{-Cp})\text{py-(p-NMe)}_2(\text{N}_2)]$ complex, increasing from 2216.3 cm^{-1} to 2248.4 cm^{-1} . Similar findings may be made for the side-bound N_2 coordination. It is clear from the minimal steric interference that the earlier observation—that the N_2 bond length and activation were mostly unaffected by electronic factors—was correct.

Table 3.6 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm^{-1}) for para-substituted pyridine series $[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{N}_2)]$

$[\text{Co}(\eta^5\text{-Cp})(\text{L})(\text{N}_2)]$	$\kappa\text{-N}$	$\eta^2\text{-NN}$		
	N-N (Å)	ν_{NN} (cm^{-1})	N-N (Å)	ν_{NN} (cm^{-1})
L= py-(p-NMe ₂)	1.13	2216.3	1.16	1963.7
L=py	1.13	2225.3	1.16	1972.7
L= py-(p-NO ₂)	1.13	2248.4	1.16	2003.0

To examine the periodic trend in previously investigated systems, the metal center was then replaced with other metals in the $[[\text{M}](\eta^5\text{-Cp})(\text{CNMe})(\text{N}_2)]$ complexes (Figure 3.7). Interestingly, it was observed that the preferred binding mode for all three metal centers was the $\kappa\text{-N}$ isomer with a linear orientation. With respect to Co and Ir, the data in the table

shows a periodic trend where the relative energy between the η^2 -NN or κ -N binding mode increases as one descends the group. It should also be noted that Rh does not follow any sort of trend, which is similar to observations in chapter 2 as well as in previous work.⁴⁸

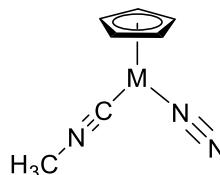
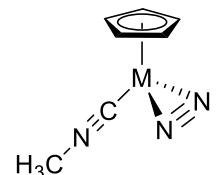
		
M=Co	0.0 kcal/mol	+12.5 kcal/mol
M=Rh	0.0 kcal/mol	+9.4 kcal/mol
M=Ir	0.0 kcal/mol	+15.6 kcal/mol

Figure 3.7 Metal identity modification effect on binding energy in $[[M]CNMe(\eta^5\text{-}Cp)(N_2)]$

In this $[[M](\eta^5\text{-}Cp)(CNMe)(N_2)]$ series, the N-N bond distance in all the terminal bound complexes is 1.12 Å, similar observations can be seen with the side-bound N₂ unit (Table 3.7). From this, it can be seen that the N-N stretching frequency in the terminal [Co(η^5 -Cp)(CNMe)(N₂)] complex decreases from 2274.7 cm⁻¹ to 2264.0 cm⁻¹ in the terminal bound [Ir(η^5 -Cp)(CNMe)(N₂)] complex, opposite observations can be seen with the side-bound N₂ coordination.

Table 3.7 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm⁻¹) for [[M](η⁵-Cp)(CNMe)(N₂)]

		κ-N		η ² -NN	
	[[M](η ⁵ -Cp)(CNMe)(N ₂)]	N-N (Å)	v _{NN} (cm ⁻¹)	N-N (Å)	v _{NN} (cm ⁻¹)
M=Co		1.12	2274.7	1.15	2022.3
M=Rh		1.12	2300.1	1.14	2134.7
M=Ir		1.12	2264.0	1.15	2040.6

Sticking with a d⁸ system that has a similar geometry as the previously studied system, a number of square planar [Co(L)₂Cl(N₂)] complexes were studied employing a range of ligands (L) (Figure 3.8). It is clear that the relative energy difference between η²-NN and κ-N grows as one advances from an electron-rich to an electron-deficient system, which is in line with the observations in the [CoCp(L)₂(N₂)] and [Ni(L)₂(N₂)] systems.

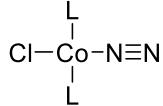
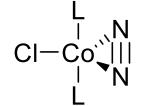
			
L=PMe ₃	0.0 kcal/mol		+10.3 kcal/mol
L=PH ₃	0.0 kcal/mol		+10.9 kcal/mol
L=CNMe	0.0 kcal/mol		+11.8 kcal/mol
L=CO	0.0 kcal/mol		+12.9 kcal/mol

Figure 3.8 Ligand identity modification effect on binding energy in [Co(L)₂Cl(N₂)]

In this $[\text{Co}(\text{L})_2\text{Cl}(\text{N}_2)]$ series, the N-N bond distance decreases from 1.13 Å in the terminal bound $[\text{Co}(\text{PMe}_3)_2\text{Cl}(\text{N}_2)]$ complex to 1.12 Å in the terminal bound $[\text{Co}(\text{CO})_2\text{Cl}(\text{N}_2)]$ complex, similar observations can be seen with the side-bound N_2 unit (Table 3.8). From this, it can be seen that the N-N stretching frequency in the terminal bound $[\text{Co}(\text{PMe}_3)_2\text{Cl}(\text{N}_2)]$ complex increases from 2209.1 cm^{-1} to 2335.8 cm^{-1} in the terminal bound $[\text{Co}(\text{CO})_2\text{Cl}(\text{N}_2)]$ complex, similar observations can be seen with the side-bound N_2 coordination.

Table 3.8 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm^{-1}) for $[\text{Co}(\text{L})_2\text{Cl}(\text{N}_2)]$

$[\text{Co}(\text{L})_2\text{Cl}(\text{N}_2)]$	$\kappa\text{-N}$ N-N (Å)	ν_{NN} (cm^{-1})	$\eta^2\text{-NN}$ N-N (Å)	ν_{NN} (cm^{-1})
L=PMe ₃	1.13	2209.1	1.16	1965.5
L=PH ₃	1.13	2241.9	1.16	1991.8
L=CNMe	1.12	2264.3	1.15	2050.0
L=CO	1.12	2335.8	1.14	2095.8

In an effort to further comprehend the reduction in energy differences between the $\eta^2\text{-NN}$ and $\kappa\text{-N}$ binding modes utilizing the findings made, a number of pseudo-trigonal planar $[[\text{M}]((\eta^5\text{-Cp}^*)_2\text{N}_2)]$ complexes were studied (Figures 3.9). These systems were picked since it has been shown experimentally that the respective binding energies of N_2 's $\eta^2\text{-NN}$ and $\kappa\text{-N}$ binding modes are generally close to one another and occasionally even in

equilibrium as well as that the relative energy for the η^2 -NN and κ -N binding modes and N_2 activation depended on the structure's shape, electronics, the metals oxidation state, and the number of electrons. As demonstrated by the work of the Chirik and Bercaw laboratories, these structures are also known to favor the binding of N_2 in a side-bound coordination.^{46,51} These systems were chosen when it was discovered that the relative energy for the η^2 -NN and κ -N binding modes and N_2 activation depended on the structure's shape, electronics, the metals oxidation state, and the number of electrons. According to the results, the κ -N binding mode is the energetically favored binding mode for the triplet and singlet forms of $[Ti(\eta^5\text{-Cp}^*)_2(N_2)]$ as well as for $[Zr(\eta^5\text{-Cp}^*)_2(N_2)]$ as a triplet. It's interesting to note that the η^2 -NN binding mode was discovered to be N_2 's preferred energetic binding mode for $[Zr(\eta^5\text{-Cp}^*)_2(N_2)]$ as a singlet. It should be noticed that when one moves from Ti to Zr, the relative energy between the η^2 -NN and κ -N binding modes diminishes.

		Multiplicity		
M=Ti	1		0.0 kcal/mol	+0.9 kcal/mol
M=Ti	3		0.0 kcal/mol	+8.7 kcal/mol
M=Zr	1		+2.6 kcal/mol	0.0 kcal/mol
M=Zr	3		0.0 kcal/mol	+4.0 kcal/mol

Figure 3.9 Metal identity modification effect on binding energy in $[[M](\eta^5\text{-Cp}^*)_2(N_2)]$

In this $[[M](\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ series, the N-N bond distance is 1.14 Å in the terminal bound $[\text{Ti}(\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ singlet complex and the terminal bound $[\text{Zr}(\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ singlet complex but increases from 1.13 Å in the terminal bound $[\text{Ti}(\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ triplet complex to 1.14 Å in the $[\text{Zr}(\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ triplet complex, similar observations can be seen with the side-bound N_2 unit (Table 3.9). In direct contrast, the N-N stretching frequency in the terminal $[\text{Ti}(\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ singlet decreases from 2117.0 cm^{-1} to 2101.9 cm^{-1} in the terminal bound $[\text{Zr}(\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ singlet complex and from 2101.9 cm^{-1} in the terminal bound $[\text{Ti}(\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ triplet complex to 2076.5 cm^{-1} in the $[\text{Zr}(\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ triplet complex, similar observations can be seen with the side-bound N_2 coordination.

Table 3.9 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm^{-1}) for $[[M](\eta^5\text{-Cp}^*)_2(\text{N}_2)]$

$[[M](\eta^5\text{-Cp}^*)_2(\text{N}_2)]$	Multiplicity	N-N (Å)	ν_{NN} (cm^{-1})	N-N (Å)	ν_{NN} (cm^{-1})
M=Ti	1	1.14	2117.0	1.18	1877.2
M=Ti	3	1.13	2101.9	1.15	2030.4
M=Zr	1	1.14	2101.9	1.18	1815.7
M=Zr	3	1.14	2076.5	1.15	1999.2

As these complexes are known to favor the side-bound N_2 unit or for the $\eta^2\text{-NN}$ and $\kappa\text{-N}$ binding modes of N_2 to be in equilibrium, the next two systems to be examined were ones similar to the Sutton lab.⁴⁴ The $\kappa\text{-N}$ isomer is 19.4 kcal/mol more stable in the $[\text{Re}(\eta^5\text{-Cp}^*)_2(\text{N}_2)]$ series than the $\eta^2\text{-NN}$ isomer.

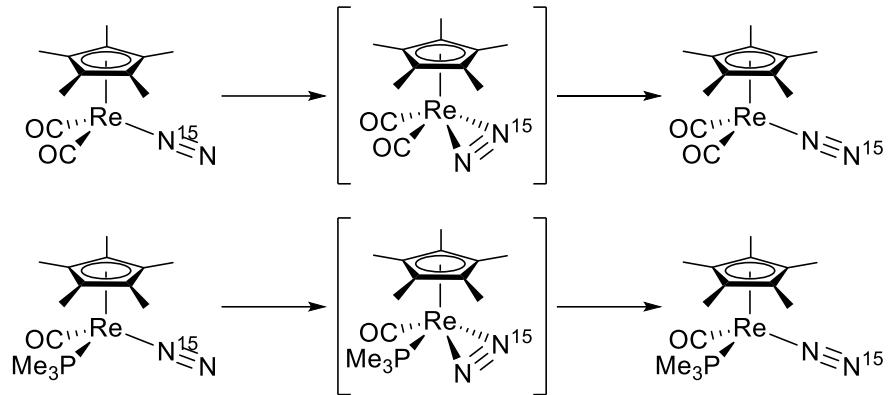
$\text{Cp}^*)(\text{CO})_2]$ system, according to calculations (Figure 3.10). It can also be seen that the κ -N isomer is 20.3 kcal/mol more stable in the $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})(\text{PMe}_3)]$ system.

κ -N	η^2 -NN	κ -N	η^2 -NN
0.0 kcal/mol	+19.4 kcal/mol	0.0 kcal/mol	+20.3 kcal/mol

Figure 3.10 Ligand modification effect on binding in $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})_2]$ and $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})(\text{PMe}_3)]$

Through a time-dependent study by the Sutton lab, using ^{15}N spectroscopy, the formation of $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})_2]$ occurred after 7 hours at room temperature, while the formation of $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})(\text{PMe}_3)]$ was not observed at room temperature, formation did occurred after 3 hours at 330 K (Scheme 3.1).⁴⁴ It can be understood that the faster formation corresponds to the molecule that is lower in energy and thereby, more accessible. This rhenium data agrees with the finding of the Sutton lab in which $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})_2]$ was observed to be lower in energy than $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})(\text{PMe}_3)]$.

Scheme 3.1 Isomerization of $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})_2]$ and $[\text{Re}(\eta^5\text{-Cp}^*)(\text{CO})(\text{PMe}_3)]$ by the Sutton lab⁴⁴



In an effort to comprehend how the geometry of the structure, the d electron count, and the metal center's oxidation state impact the relative binding of the two binding modes and N_2 activation, a number of octahedral $[[\text{M}](\text{NH}_3)_5(\text{N}_2)]^{2+}$ complexes were studied (Figure 3.11). As one goes down the group, as was the case in the prior systems, it can be seen that the relative energy between the $\eta^2\text{-NN}$ and $\kappa\text{-N}$ binding modes rises.

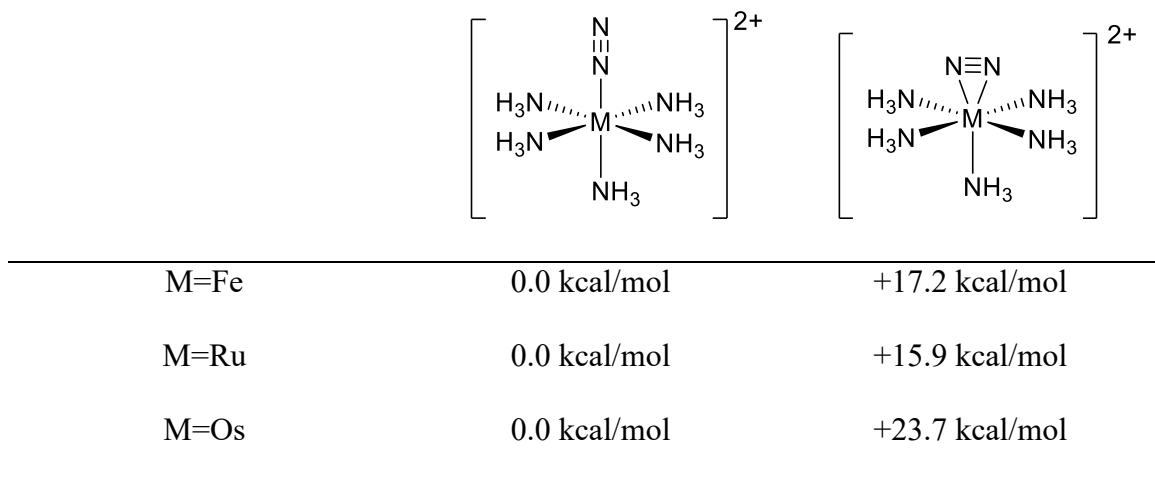


Figure 3.11 Metal identity modification effect on binding energy in $[[\text{M}](\text{NH}_3)_5(\text{N}_2)]$

In this $[[M](NH_3)_5(N_2)]^{2+}$ series, the N-N bond distance increases from 1.11 Å in the terminal bound $[Fe(NH_3)_5(N_2)]^{2+}$ complex to 1.12 Å in the terminal bound $[Os(NH_3)_5(N_2)]^{2+}$ complex, similar observations can be seen with the side-bound N₂ unit (Table 3.10). From this, it can be seen that the N-N stretching frequency in the terminal-bound $[Fe(NH_3)_5(N_2)]^{2+}$ complex decreased from 2347.0 cm⁻¹ to 2268.5 cm⁻¹ in the terminal-bound $[Os(NH_3)_5(N_2)]^{2+}$ complex, similar observations can be seen with the side-bound N₂ coordination.

Table 3.10 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm⁻¹) for $[[M](NH_3)_5(N_2)]$

$[[M](NH_3)_5(N_2)]^{2+}$	$\kappa\text{-N}$ N-N (Å)	ν_{NN} (cm ⁻¹)	$\eta^2\text{-NN}$ N-N (Å)	ν_{NN} (cm ⁻¹)
M=Fe	1.11	2347.0	1.12	2309.9
M=Ru	1.11	2339.8	1.12	2311.6
M=Os	1.12	2268.5	1.14	2139.4

From the data in chapter 2 over N₂O, and previous work,⁴⁸ it can be seen that $\eta^2\text{-NN}$ is a better π -acceptor than $\kappa\text{-N}$. The favorability of $\eta^2\text{-NN}$ was observed in low oxidation states systems with a good electron-rich π -donor metal. Lastly, in view of the decrease in the relative energy between the $\eta^2\text{-NN}$ and $\kappa\text{-N}$ binding modes and N₂ activation by changing the electronics, and oxidation state of the metal system, more attempts were made to locate a system in which the $\eta^2\text{-NN}$ mode of N₂ was not plausible. To do this, a series

of $[[M](\eta^5\text{-Cp})(\text{N}_2)\text{Cl}_2]^{3+}$ complexes similar to the previous system were investigated however, an additional supporting ligand was added to increase the oxidation state of the metal center. It should be noted that the $\eta^2\text{-NN}$ binding mode could not be successfully optimized as the N_2 unit broke off of the metal center in all three systems, which is a similar observation from previous work.⁴⁸ It can be seen in Table 3.11 that the N_2 bond distance and stretching frequency of the $\kappa\text{-N}$ binding mode do not follow any periodic trend.

Table 3.11 N-N bond distances (in Å) and N-N stretch vibrational frequencies (in cm^{-1}) for $[[M](\eta^5\text{-Cp})(\text{N}_2)\text{Cl}_2]$

$[[M](\eta^5\text{-Cp})(\text{N}_2)\text{Cl}_2]^{3+}$	N-N (Å)	$\nu_{\text{NN}} (\text{cm}^{-1})$
M=Co	1.11	2416.2
M=Rh	1.11	2427.7
M=Ir	1.11	2378.1

3.4 Conclusion

According to this study's use of computational chemistry, the linear configuration of the N_2 unit bonding through the terminal nitrogen is preferred. However, it was shown that the strong π -donor properties of a low valent metal might alter the preference in favor of $\eta^2\text{-NN}$. It was interestingly shown that the geometry and electronics of the structure also affected the favorability between the $\eta^2\text{-NN}$ and $\kappa\text{-N}$ binding modes. It may be said that $\eta^2\text{-NN}$ is a superior π -acceptor than $\kappa\text{-N}$ and requires a suitable electron-rich low valent π -

donor metal in order to be possibly favorable. These observations help to explain the N₂'s known binding characteristics and pave the way for more investigation into the chemistry of this remarkable small molecule.

3.5 References

- (1) Zhang, X.; Ward, B. B.; Sigman, D. M. Global Nitrogen Cycle: Critical Enzymes, Organisms, and Processes for Nitrogen Budgets and Dynamics. *Chem. Rev.* **2020**, *120* (12), 5308–5351.
https://doi.org/10.1021/ACS.CHEMREV.9B00613/ASSET/IMAGES/LARGE/CR9B00613_0002.JPG.
- (2) Yang, H.; Rittle, J.; Marts, A. R.; Peters, J. C.; Hoffman, B. M. ENDOR Characterization of $(N_2)Fe^{II}(\mu-H)_2Fe^{I}(N_2)\text{-}$: A Spectroscopic Model for N_2 Binding by the Di- μ -Hydrido Nitrogenase Janus Intermediate. *Inorg. Chem.* **2018**, *57* (19), 12323–12330.
https://doi.org/10.1021/ACS.INORGCHEM.8B02021/ASSET/IMAGES/MEDIUM/IC-2018-02021U_M010.GIF.
- (3) Vela, J.; Cirera, J.; Smith, J. M.; Lachicotte, R. J.; Flaschenriem, C. J.; Alvarez, S.; Holland, P. L. Quantitative Geometric Descriptions of the Belt Iron Atoms of the Iron-Molybdenum Cofactor of Nitrogenase and Synthetic Iron(II) Model Complexes. *Inorg. Chem.* **2007**, *46* (1), 60–71.
https://doi.org/10.1021/IC0609148/SUPPL_FILE/IC0609148SI20060525_034035.CIF.
- (4) Čorić, I.; Holland, P. L. Insight into the Iron-Molybdenum Cofactor of Nitrogenase from Synthetic Iron Complexes with Sulfur, Carbon, and Hydride Ligands. *J. Am. Chem. Soc.* **2016**, *138* (23), 7200–7211.
https://doi.org/10.1021/JACS.6B00747/ASSET/IMAGES/LARGE/JA-2016-00747G_0018.JPG.
- (5) Horitani, M.; Grubel, K.; McWilliams, S. F.; Stubbert, B. D.; Mercado, B. Q.; Yu, Y.; Gurubasavaraj, P. M.; Lees, N. S.; Holland, P. L.; Hoffman, B. M. ENDOR Characterization of an Iron-Alkene Complex Provides Insight into a Corresponding Organometallic Intermediate of Nitrogenase. *Chem. Sci. J.* **2017**, *8* (9), 5941–5948.
<https://doi.org/10.1039/C7SC01602F>.
- (6) Rohde, M.; Laun, K.; Zebger, I.; Stripp, S. T.; Einsle, O. Two Ligand-Binding Sites in CO-Reducing V Nitrogenase Reveal a General Mechanistic Principle. *Sci. Adv.* **2021**, *7* (22), 4474–4502.
https://doi.org/10.1126/SCIAADV.ABG4474/SUPPL_FILE/SCIAADV.ABG4474_SM.PDF

- (7) Mishina, M.; Tanabe, T.; Doya, K.; Wickens, J.; Toyoizumi, T.; Kasai, K.; Spatzal, T.; Perez, K. A.; Einsle, O.; Howard, J. B.; Rees, D. C. Ligand Binding to the FeMo-Cofactor: Structures of Co-Bound and Reactivated Nitrogenase. *Science* (1979) **2014**, 345 (6204), 1620–1623. <https://doi.org/10.1126/SCIENCE.1256679>.
- (8) Spatzal, T.; Perez, K. A.; Einsle, O.; Howard, J. B.; Rees, D. C. Ligand Binding to the FeMo-Cofactor: Structures of Co-Bound and Reactivated Nitrogenase. *Science* (1979) **2014**, 345 (6204), 1620–1623.
https://doi.org/10.1126/SCIENCE.1256679/SUPPL_FILE/1256679.SPATZAL.SM.PDF.
- (9) Skubi, K. L.; Holland, P. L. So Close, yet Sulfur Away: Opening the Nitrogenase Cofactor Structure Creates a Binding Site. *Biochemistry* **2018**, 57 (26), 3540–3541.
https://doi.org/10.1021/ACS.BIOCHEM.8B00529/ASSET/IMAGES/LARGE/BI-2018-005298_0003.JPG.
- (10) Kang, W.; Lee, C. C.; Jasniewski, A. J.; Ribbe, M. W.; Hu, Y. Structural Evidence for a Dynamic Metallocofactor during N₂ Reduction by Mo-Nitrogenase. *Science* (1979) **2020**, 368 (6497), 1381–1385.
https://doi.org/10.1126/SCIENCE.AAZ6748/SUPPL_FILE/AAZ6748_MDAR REP_RODUCIBILITY_CHECKLIST.PDF.
- (11) Holland, P. L. Low-Coordinate Iron Complexes as Synthetic Models of Nitrogenase. <https://doi.org/10.1139/v05-005> **2011**, 83 (4), 296–301. <https://doi.org/10.1139/V05-005>.
- (12) *Nitrogenase and Nitrogen Activation / The Holland Group.*
<https://holland.chem.yale.edu/research/nitrogenase-and-nitrogen-activation> (accessed 2022-06-30).
- (13) Bezdek, M. J.; Chirik, P. J. Expanding Boundaries: N₂ Cleavage and Functionalization beyond Early Transition Metals. *Angew. Chem., Int. Ed.* **2016**, 55 (28), 7892–7896. <https://doi.org/10.1002/ANIE.201603142>.
- (14) Knobloch, D. J.; Lobkovsky, E.; Chirik, P. J. Functionalization of Hafnium Oxamidide Complexes Prepared from CO-Induced N₂ Cleavage. *J. Am. Chem. Soc.* **2010**, 132 (43), 15340–15350.
https://doi.org/10.1021/JA106848Y/SUPPL_FILE/JA106848Y_SI_002.PDF.

- (15) MacLeod, K. C.; Vinyard, D. J.; Holland, P. L. A Multi-Iron System Capable of Rapid N₂ Formation and N₂ Cleavage. *J. Am. Chem. Soc.* **2014**, *136* (29), 10226–10229. <https://doi.org/10.1021/JA505193Z>.
- (16) MacLeod, K. C.; Menges, F. S.; McWilliams, S. F.; Craig, S. M.; Mercado, B. Q.; Johnson, M. A.; Holland, P. L. Alkali-Controlled C-H Cleavage or N-C Bond Formation by N₂-Derived Iron Nitrides and Imides. *J. Am. Chem. Soc.* **2016**, *138* (35), 11185–11191. <https://doi.org/10.1021/JACS.6B04984>.
- (17) Semproni, S. P.; Chirik, P. J. Synthesis of a Base-Free Hafnium Nitride from N₂ Cleavage: A Versatile Platform for Dinitrogen Functionalization. *J. Am. Chem. Soc.* **2013**, *135* (30), 11373–11383.
https://doi.org/10.1021/JA405477M/SUPPL_FILE/JA405477M_SI_002.PDF.
- (18) Holland, P. L. Introduction: Reactivity of Nitrogen from the Ground to the Atmosphere. *Chem. Rev.* **2020**, *120* (12), 4919–4920.
<https://doi.org/10.1021/ACS.CHEMREV.0C00361>.
- (19) Capdevila-Cortada, M. Electrifying the Haber–Bosch. *Nat. Catal.* **2019**, *2* (12), 1055–1055. <https://doi.org/10.1038/s41929-019-0414-4>.
- (20) Smith, C.; Hill, A. K.; Torrente-Murciano, L. Current and Future Role of Haber–Bosch Ammonia in a Carbon-Free Energy Landscape. *Energy Environ. Sci.* **2020**, *13* (2), 331–344. <https://doi.org/10.1039/C9EE02873K>.
- (21) Rohde, M.; Sippel, D.; Trncik, C.; Andrade, S. L. A.; Einsle, O. The Critical E4 State of Nitrogenase Catalysis. *Biochemistry* **2018**, *57* (38), 5497–5504.
https://doi.org/10.1021/ACS.BIOCHEM.8B00509/ASSET/IMAGES/LARGE/BI-2018-00509N_0006.JPG.
- (22) del Castillo, T. J.; Thompson, N. B.; Peters, J. C. A Synthetic Single-Site Fe Nitrogenase: High Turnover, Freeze-Quench 57Fe Mössbauer Data, and a Hydride Resting State. *J. Am. Chem. Soc.* **2016**, *138* (16), 5341–5350.
https://doi.org/10.1021/JACS.6B01706/ASSET/IMAGES/LARGE/JA-2016-01706N_0012.JPG.
- (23) Vela, J.; Stoian, S.; Flaschenriem, C. J.; Münck, E.; Holland, P. L. Erratum: A Sulfido-Bridged Diiron(II) Compound and Its Reactions with Nitrogenase-Relevant Substrates (Journal of the American Chemical Society (2004) 126 (4522-4523)). *J. Am. Chem. Soc.* **2010**, *132* (35), 12517. <https://doi.org/10.1021/JA106351N>.

- (24) Arnet, N. A.; Dugan, T. R.; Menges, F. S.; Mercado, B. Q.; Brennessel, W. W.; Bill, E.; Johnson, M. A.; Holland, P. L. Synthesis, Characterization, and Nitrogenase-Relevant Reactions of an Iron Sulfide Complex with a Bridging Hydride. *J. Am. Chem. Soc.* **2015**, 137 (41), 13220–13223.
https://doi.org/10.1021/JACS.5B06841/SUPPL_FILE/JA5B06841_SI_002.CIF.
- (25) Arnet, N. A.; Dugan, T. R.; Menges, F. S.; Mercado, B. Q.; Brennessel, W. W.; Bill, E.; Johnson, M. A.; Holland, P. L. Synthesis, Characterization, and Nitrogenase-Relevant Reactions of an Iron Sulfide Complex with a Bridging Hydride. *J. Am. Chem. Soc.* **2015**, 137 (41), 13220–13223. <https://doi.org/10.1021/JACS.5B06841>.
- (26) Wilson, D. W. N.; Holland, P. L. Nitrogenases and Model Complexes in Bioorganometallic Chemistry. *Reference Module in Chemistry, Molecular Sciences and Chemical Engineering* **2021**. <https://doi.org/10.1016/B978-0-12-820206-7.00035-4>.
- (27) Vela, J.; Stoian, S.; Flaschenriem, C. J.; Münck, E.; Holland, P. L. A Sulfido-Bridged Diiron(II) Compound and Its Reactions with Nitrogenase-Relevant Substrates. *J. Am. Chem. Soc.* **2004**, 126 (14), 4522–4523.
https://doi.org/10.1021/JA049417L/SUPPL_FILE/JA049417LSI20040301_084105.PDF.
- (28) Speelman, A. L.; Čorić, I.; van Stappen, C.; Debeer, S.; Mercado, B. Q.; Holland, P. L. Nitrogenase-Relevant Reactivity of a Synthetic Iron-Sulfur-Carbon Site. *J. Am. Chem. Soc.* **2019**, 141 (33), 13148–13157.
https://doi.org/10.1021/JACS.9B05353/SUPPL_FILE/JA9B05353_SI_002.CIF.
- (29) Thompson, N. B.; Green, M. T.; Peters, J. C. Nitrogen Fixation via a Terminal Fe(IV) Nitride. *J. Am. Chem. Soc.* **2017**, 139 (43), 15312–15315.
https://doi.org/10.1021/JACS.7B09364/SUPPL_FILE/JA7B09364_SI_002.PDF.
- (30) Schild, D. J.; Peters, J. C. Light Enhanced Fe-Mediated Nitrogen Fixation: Mechanistic Insights Regarding H₂ Elimination, HER, and NH₃ Generation. *ACS Catal.* **2019**, 9 (5), 4286–4295.
https://doi.org/10.1021/ACSCATAL.9B00523/SUPPL_FILE/CS9B00523_SI_003.MOL.
- (31) Chalkley, M. J.; Peters, J. C. Relating N–H Bond Strengths to the Overpotential for Catalytic Nitrogen Fixation. *Eur. J. Inorg. Chem.* **2020**, 2020 (15–16), 1353–1357.
<https://doi.org/10.1002/EJIC.202000232>.

- (32) Bezdek, M. J.; Pappas, I.; Chirik, P. J. Determining and Understanding N-H Bond Strengths in Synthetic Nitrogen Fixation Cycles. *Top. Organomet. Chem.* **2017**, *60*, 1–21. https://doi.org/10.1007/3418_2016_8/COVER/.
- (33) Nitrogen Fixation. **2011**, *766*. <https://doi.org/10.1007/978-1-61779-194-9>.
- (34) Speelman, A. L.; Čorić, I.; van Stappen, C.; Debeer, S.; Mercado, B. Q.; Holland, P. L. Nitrogenase-Relevant Reactivity of a Synthetic Iron-Sulfur-Carbon Site. *J. Am. Chem. Soc.* **2019**, *141* (33), 13148–13157. <https://doi.org/10.1021/JACS.9B05353>.
- (35) Horitani, M.; Grubel, K.; McWilliams, S. F.; Stubbert, B. D.; Mercado, B. Q.; Yu, Y.; Gurubasavaraj, P. M.; Lees, N. S.; Holland, P. L.; Hoffman, B. M. ENDOR Characterization of an Iron–Alkene Complex Provides Insight into a Corresponding Organometallic Intermediate of Nitrogenase. *Chem. Sci.* **2017**, *8* (9), 5941–5948. <https://doi.org/10.1039/C7SC01602F>.
- (36) Knobloch, D. J.; Semproni, S. P.; Lobkovsky, E.; Chirik, P. J. Studies into the Mechanism of CO-Induced N₂ Cleavage Promoted by an Ansa-Hafnocene Complex and C-C Bond Formation from an Observed Intermediate. *J. Am. Chem. Soc.* **2012**, *134* (7), 3377–3386. https://doi.org/10.1021/JA208562D/SUPPL_FILE/JA208562D_SI_002.CIF.
- (37) Yandulov, D. v.; Schrock, R. R. Catalytic Reduction of Dinitrogen to Ammonia at a Single Molybdenum Center. *Science (1979)* **2003**, *301* (5629), 76–78. <https://doi.org/10.1126/SCIENCE.1085326>.
- (38) Tsai, Y. C.; Cummins, C. C. Base-Catalyzed Dinitrogen Cleavage by Molybdenum Amides. *Inorg. Chim. Acta* **2003**, *345*, 63–69. [https://doi.org/10.1016/S0020-1693\(02\)01344-0](https://doi.org/10.1016/S0020-1693(02)01344-0).
- (39) Bruch, Q. J.; Malakar, S.; Goldman, A. S.; Miller, A. J. M. Mechanisms of Electrochemical N₂ Splitting by a Molybdenum Pincer Complex. *Inorg. Chem.* **2022**, *61* (4), 2307–2318. https://doi.org/10.1021/ACS.INORGCHEM.1C03698/SUPPL_FILE/IC1C03698_SI_002.ZIP.
- (40) Yamout, L. S.; Ataya, M.; Hasanayn, F.; Holland, P. L.; Miller, A. J. M.; Goldman, A. S. Understanding Terminal versus Bridging End-on N₂ Coordination in Transition Metal Complexes. *J. Am. Chem. Soc.* **2021**, *143* (26), 9744–9757. <https://doi.org/10.1021/JACS.1C01146>.

- (41) van Alten, R. S.; Wätjen, F.; Demeshko, S.; Miller, A. J. M.; Würtele, C.; Siewert, I.; Schneider, S. (Electro-)Chemical Splitting of Dinitrogen with a Rhenium Pincer Complex. *Eur. J. Inorg. Chem.* **2020**, 2020 (15–16), 1402–1410.
<https://doi.org/10.1002/EJIC.201901278>.
- (42) Creutz, S. E.; Peters, J. C. Catalytic Reduction of N₂ to NH₃ by an Fe-N₂ Complex Featuring a C-Atom Anchor. *J. Am. Chem. Soc.* **2014**, 136 (3), 1105–1115.
https://doi.org/10.1021/JA4114962/SUPPL_FILE/JA4114962_SI_002.CIF.
- (43) Gynane, B. M. J. S.; Jeffery, J.; Lappert, M. F. Bonding in [Zr(q-CsH_n),(N₂)(R)] [R = Me, Si, CH]. *J.C.S. Chem. Comm.* **1978**, No. 34, 34–36.
- (44) Cusanelli, A.; Sutton, D. End-to-End Rotation of Rhenium-Bound Dinitrogen. *J. Organomet. Chem.* **1996**, No. 20, 1457–1464.
- (45) Fomitchev, D. v.; Bagley, K. A.; Coppens, P. The First Crystallographic Evidence for Side-on Coordination of N₂ to a Single Metal Center in a Photoinduced Metastable State. *J. Am. Chem. Soc.* **2000**, 122 (3), 532–533.
<https://doi.org/10.1021/ja993623p>.
- (46) Bercaw, E. John, Rosenberg, Edward, Roberts, D. J. ‘H, ¹³C, and ¹⁵N Nuclear Magnetic Resonance Studies of a Dinitrogen Complex of Permethyltitanocene. Evidence for Molecular Nitrogen Coordinated “Edge-on” to a Transition Metal. *J. Am. Chem. Soc.* **1974**, 9–25.
- (47) Frisch, M. J. ; et. al. Gaussian 16, Revision A.03. Gaussian Inc.: Wallingford, CT 2009.
- (48) Fields, K.; Barngrover, B. M.; Gary, J. B. Computational Investigation of the Preferred Binding Modes of N₂O in Group 8 Metal Complexes. *Inorg. Chem.* **2020**, 59 (24), 18314–18318. <https://doi.org/10.1021/acs.inorgchem.0c02903>.
- (49) Lombardi, B. M. P.; Gendy, C.; Gelfand, B. S.; Bernard, G. M.; Wasylissen, R. E.; Tuononen, H. M.; Roesler, R. Side-on Coordination in Isostructural Nitrous Oxide and Carbon Dioxide Complexes of Nickel. *Angew. Chem., Int. Ed.* **2021**, 60 (13), 7077–7081. <https://doi.org/10.1002/ANIE.202011301>.
- (50) Mokhtarzadeh, C. C.; Chan, C.; Moore, C. E.; Rheingold, A. L.; Figueroa, J. S. Side-On Coordination of Nitrous Oxide to a Mononuclear Cobalt Center. *J. Am. Chem. Soc.* **2019**, 141 (38), 15003–15007. <https://doi.org/10.1021/JACS.9B08241>.

(51) Pun, D.; Lobkovsky, E.; Chirik, P. J. Indenyl Zirconium Dinitrogen Chemistry: N₂ Coordination to an Isolated Zirconium Sandwich and Synthesis of Side-on, End-on Dinitrogen Compounds. *J. Am. Chem. Soc.* **2008**, *130* (18), 6047–6054.
https://doi.org/10.1021/JA801021W/SUPPL_FILE/JA801021W-FILE005.CIF.

Chapter 4

Conclusions and Future Prospects

4.1 Conclusions

The activation of N₂O and N₂ has long caught the considerable interest of inorganic chemists.^{1,2} Despite being a potentially effective oxygen-atom transfer agent for transition metals, N₂O is a notably poor ligand, and the coordination chemistry of just a few terminal, end-on κ -N, and side-bound η^2 -NN complexes has been characterized.¹ Despite the fact that molecular nitrogen is abundant in the atmosphere as N₂, many organisms cannot access it, making it a limited resource that frequently restricts primary production in many ecosystems.² This work used computational chemistry to examine the binding of these two small molecules to various transition metal complexes in an effort to understand the binding properties of N₂O and N₂.

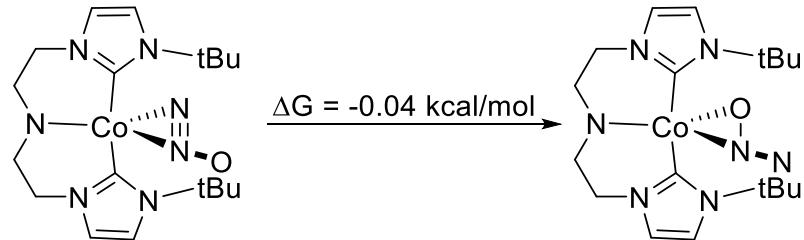
This study demonstrates that the N₂O unit prefers to connect across both nitrogens in a bent configuration to low valent metal centers. The strong π -donor properties of a metal with a low valent electron configuration can alter the preference for η^2 -NN over κ -N as seen in Table 4.1.

Table 4.1 Trans ligand modification effect on terminal nitrogen vs oxygen binding

Ligand (L)	Energy (ΔG in kcal/mol)
PMe ₃	-6.6
PH ₃	-6.0
CMNe	-2.9
CO	+3.3

The binding between the nitrogen and oxygen atoms of the N₂O can be enhanced by using a pseudo-square planar geometry, similar to the Figueroa lab system, and making the metal center more electron-rich as seen in Scheme 4.1.

Scheme 4.1 Binding between the nitrogen and oxygen atoms of the N₂O

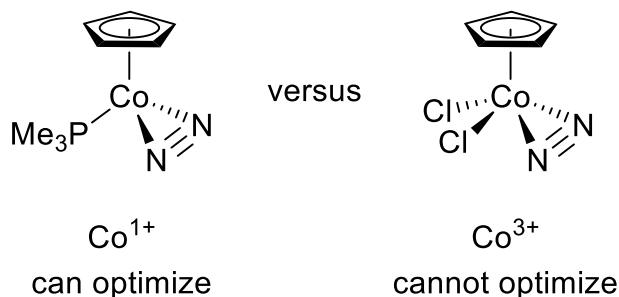


One may conclude that η²-NN is a better π-acceptor than κ-N and requires a suitable low valent π-donor metal that is electron-rich in order to be the favored isomer. These

discoveries help to explain the well-known binding characteristics of N₂O and N₂, and they pave the way for more investigation into the chemistry of these incredible small molecules.

This study showed that the N₂ unit prefers to bind via the terminal nitrogen in a linear arrangement. It was demonstrated, nonetheless, that the strong π -donor characteristics of a low valent metal might change the choice for η^2 -NN. It should also be noted that the weak π -donor characteristics of a high valent metal will not bind N₂ in side-bound coordination as seen in Scheme 4.2.

Scheme 4.2 Electronic effect on N₂ binding



4.2 Future Prospects

While this work offers significant insight into the potential binding of N₂O and N₂, many questions and research directions remain to be studied. The use of N₂O is desirable as it is a relatively “green oxidant”.¹ However, understanding of its binding, specifically in a side-bound configuration is fairly limited as few complexes have been characterized. As such, future work could continue to investigate the η^2 -NO binding mode of N₂O in order to find a system in which it is favored far more than the terminal-bound isomers. Likely metals to

investigate could be low oxidation state early metals because the metal needs to be a good electron rich π -donor.

Terminal and side-on coordination of N₂ to transition metal complexes offer possibilities for distinct pathways in ammonia synthesis and N₂ functionalization. Additionally, further investigation into the activation of N₂ through transition metal binding would be worthwhile. One could start, similar to N₂O, with low oxidation state early metals because the metal needs to be a good electron rich π -donor.

Furthermore, future work could consist of a computational examination of the transfer of an oxygen atom from N₂O to a metal core, with the systems from chapter 2. This study on how the preferred binding mechanism relates to the kinetics of O-atom transfer could be an important step toward making this potentially green oxidant more widely used.

4.3 References

- (1) Mokhtarzadeh, C. C.; Chan, C.; Moore, C. E.; Rheingold, A. L.; Figueroa, J. S. Side-On Coordination of Nitrous Oxide to a Mononuclear Cobalt Center. *J. Am. Chem. Soc.* **2019**, *141* (38), 15003–15007. <https://doi.org/10.1021/JACS.9B08241>.
- (2) Zhang, X.; Ward, B. B.; Sigman, D. M. Global Nitrogen Cycle: Critical Enzymes, Organisms, and Processes for Nitrogen Budgets and Dynamics. *Chem. Rev.* **2020**, *120* (12), 5308–5351.
https://doi.org/10.1021/ACS.CHEMREV.9B00613/ASSET/IMAGES/LARGE/CR9B_00613_0002.JPG.

Appendix

XYZ Coordinates – Calculated Structures

[Ni(C₁₀N₄H₁₆)₂(N₂O)]

κ-N

6	-2.140938000	-2.045558000	-0.818745000
6	-0.597413000	-0.516911000	-0.111296000
1	-2.535948000	-2.838825000	-1.434993000
6	4.604443000	-0.996899000	-0.199216000
6	4.104148000	-1.802056000	0.771188000
6	2.457791000	-0.338163000	0.175108000
7	3.593986000	-0.118061000	-0.554914000
1	5.576652000	-0.976256000	-0.666783000
1	4.554884000	-2.613072000	1.322204000
7	2.792707000	-1.397442000	0.979695000
6	1.918110000	-1.959256000	1.994820000
6	3.757699000	0.987737000	-1.497664000
1	3.007927000	0.919337000	-2.290250000
1	3.647545000	1.949267000	-0.974583000
1	4.752983000	0.914638000	-1.943306000
7	-1.746334000	-0.445961000	0.636716000

7	-0.866605000	-1.525188000	-0.999916000
6	-2.693897000	-1.366921000	0.218205000
1	-3.660759000	-1.461646000	0.688586000
6	-1.933207000	0.519467000	1.710771000
1	-2.891546000	1.033998000	1.592680000
1	-1.902470000	0.030792000	2.690830000
1	-1.119745000	1.244542000	1.635992000
6	0.045119000	-1.942241000	-2.053796000
28	0.884452000	0.645459000	0.081653000
1	1.065266000	-1.730712000	-1.731456000
1	-0.067393000	-3.015627000	-2.232546000
1	-0.155249000	-1.400475000	-2.984684000
1	1.981725000	-1.389901000	2.928697000
1	0.889733000	-1.929645000	1.631805000
1	2.207423000	-2.996140000	2.187892000
7	1.225795000	2.311947000	0.185097000
7	1.725627000	3.408243000	0.033361000
8	2.950878000	3.725566000	-0.058440000

κ-O

NA

η-NN

6	-2.418633000	-0.761678000	-1.190271000
6	-0.889392000	0.776777000	-0.472671000
1	-2.796740000	-1.570501000	-1.796635000
6	4.293745000	0.153009000	-0.255833000
6	3.695678000	-0.565299000	0.727348000
6	2.135567000	0.876090000	-0.114636000
7	3.336986000	1.021400000	-0.754774000
1	5.301618000	0.122105000	-0.640068000
1	4.080073000	-1.336999000	1.376372000
7	2.382707000	-0.120347000	0.794704000
6	1.421289000	-0.587084000	1.781128000
6	3.587023000	2.006968000	-1.806951000
1	3.052716000	1.731913000	-2.721737000
1	3.260872000	2.995979000	-1.469296000
1	4.659495000	2.022536000	-2.015643000
7	-2.063647000	0.886143000	0.222717000
7	-1.131618000	-0.260197000	-1.334544000
6	-3.004052000	-0.040734000	-0.202022000
1	-3.990066000	-0.106270000	0.232031000
6	-2.300729000	1.875213000	1.266600000
1	-3.253564000	2.381838000	1.088247000

1	-2.323182000	1.400162000	2.254055000
1	-1.494633000	2.606869000	1.217404000
6	-0.186008000	-0.717168000	-2.341531000
28	0.603344000	1.941251000	-0.377514000
1	0.825826000	-0.589575000	-1.955318000
1	-0.366071000	-1.774609000	-2.555168000
1	-0.289117000	-0.140873000	-3.267271000
1	1.447622000	0.037737000	2.680557000
1	0.421371000	-0.544142000	1.349309000
1	1.657261000	-1.619090000	2.055635000
7	-0.133317000	3.670750000	-0.568086000
7	1.099538000	3.663962000	-0.507858000
8	2.024420000	4.513930000	-0.538336000

η -NO (constrained)

6	-2.966131000	0.017230000	0.575884000
6	-1.334130000	1.342441000	-0.319307000
1	-3.414151000	-0.646660000	1.299542000
6	3.826180000	0.836456000	0.282525000
6	3.379119000	-0.062461000	-0.629785000
6	1.733705000	1.495415000	-0.337207000
7	2.819685000	1.773954000	0.447310000

1	4.759612000	0.892044000	0.821096000
1	3.847338000	-0.938360000	-1.051620000
7	2.102395000	0.346306000	-0.990424000
6	1.294833000	-0.309151000	-2.006113000
6	2.920281000	2.954548000	1.306090000
1	2.674626000	3.846349000	0.719066000
1	2.242783000	2.864291000	2.161164000
1	3.946435000	3.025833000	1.674654000
7	-2.458577000	1.346406000	-1.100070000
7	-1.672568000	0.503398000	0.708457000
6	-3.460261000	0.547132000	-0.571231000
1	-4.419782000	0.425121000	-1.050481000
6	-2.582665000	2.107236000	-2.338445000
1	-2.535960000	1.442083000	-3.208083000
1	-3.535189000	2.644796000	-2.348102000
1	-1.766598000	2.829231000	-2.372348000
6	-0.809634000	0.215033000	1.844344000
28	0.187915000	2.513680000	-0.535026000
1	0.226693000	0.348768000	1.534182000
1	-1.022757000	0.889947000	2.680480000
1	-0.964772000	-0.817989000	2.169635000

1	1.579499000	-1.363137000	-2.069158000
1	0.243136000	-0.230272000	-1.727201000
1	1.438687000	0.160003000	-2.985458000
7	0.726715000	4.249617000	-0.783700000
7	1.585728000	5.075605000	-0.797114000
8	-0.512094000	4.204249000	-0.944310000

[Ni(PMe₃)₂(N₂O)]

κ-N			
28	2.497864000	0.050376000	-0.276226000
15	1.354148000	1.854317000	-0.123853000
15	1.313071000	-1.712046000	0.043613000
6	0.869197000	2.371838000	1.588249000
1	0.276560000	3.294620000	1.585627000
1	0.288280000	1.577009000	2.066829000
1	1.772944000	2.534854000	2.184251000
6	2.125038000	3.416198000	-0.759297000
1	2.361619000	3.294133000	-1.821227000
1	1.468501000	4.287495000	-0.638736000
1	3.064337000	3.600065000	-0.228471000
6	-0.268593000	1.913855000	-1.021024000
1	-0.930717000	1.123023000	-0.653185000

1	-0.772607000	2.880755000	-0.902210000
1	-0.094850000	1.738170000	-2.088021000
6	-0.025119000	-1.676899000	1.328255000
1	-0.758014000	-0.899931000	1.086373000
1	-0.546088000	-2.638350000	1.414984000
1	0.418245000	-1.430077000	2.299004000
6	2.230498000	-3.227417000	0.588837000
1	1.571455000	-4.097160000	0.703727000
1	3.011164000	-3.465855000	-0.140504000
1	2.720057000	-3.026272000	1.547442000
6	0.400866000	-2.352074000	-1.436993000
1	1.118818000	-2.587348000	-2.229270000
1	-0.178308000	-3.254044000	-1.204915000
1	-0.277490000	-1.581117000	-1.816223000
7	4.186574000	0.007649000	-0.632770000
7	5.321333000	-0.217653000	-0.842475000
8	6.447484000	0.213447000	-1.088758000

κ-O

28	2.450765000	0.133874000	-0.089502000
15	1.380653000	1.927466000	-0.148844000
15	1.461742000	-1.698798000	0.066043000

6	0.803635000	2.570258000	1.497671000
1	0.139366000	3.437240000	1.390937000
1	0.272143000	1.780529000	2.038399000
1	1.670848000	2.863289000	2.098741000
6	2.062802000	3.497258000	-0.888849000
1	2.323618000	3.313791000	-1.936990000
1	1.354920000	4.335914000	-0.845879000
1	2.978876000	3.783154000	-0.360617000
6	-0.234004000	1.873102000	-1.073324000
1	-0.877958000	1.094850000	-0.650569000
1	-0.767374000	2.831972000	-1.038442000
1	-0.042305000	1.616402000	-2.120891000
6	0.240289000	-1.801482000	1.465921000
1	-0.485217000	-0.985391000	1.386992000
1	-0.298701000	-2.757471000	1.474225000
1	0.768875000	-1.688837000	2.418607000
6	2.331057000	-3.330601000	0.327937000
1	1.638053000	-4.178993000	0.408627000
1	3.014099000	-3.517181000	-0.508338000
1	2.929100000	-3.280944000	1.244736000
6	0.390924000	-2.173009000	-1.378227000

1	1.024703000	-2.345206000	-2.254698000
1	-0.190309000	-3.082754000	-1.180488000
1	-0.296359000	-1.355757000	-1.617627000
8	4.395950000	0.199487000	0.091521000
7	5.360573000	-0.295806000	-0.438946000
7	6.274457000	-0.756218000	-0.943251000

η-NN

28	5.394728000	0.209776000	-0.375478000
15	4.240093000	2.034484000	-0.065491000
15	4.219556000	-1.550714000	0.074841000
6	3.843270000	2.497586000	1.680972000
1	3.304283000	3.450332000	1.741259000
1	3.237024000	1.716562000	2.150725000
1	4.778128000	2.584723000	2.244214000
6	5.110556000	3.543598000	-0.669476000
1	5.293889000	3.446873000	-1.743842000
1	4.534412000	4.456929000	-0.480756000
1	6.085060000	3.617654000	-0.178344000
6	2.605713000	2.190129000	-0.919792000
1	1.925322000	1.398883000	-0.588279000
1	2.133822000	3.162050000	-0.733669000

1	2.758298000	2.073912000	-1.997920000
6	2.801707000	-1.483166000	1.261281000
1	2.020243000	-0.814252000	0.885169000
1	2.362400000	-2.473331000	1.430073000
1	3.155358000	-1.089990000	2.220255000
6	5.276667000	-2.864680000	0.820012000
1	4.761288000	-3.830092000	0.882654000
1	6.185833000	-2.957692000	0.216972000
1	5.572633000	-2.553638000	1.827710000
6	3.477839000	-2.391459000	-1.393870000
1	4.272823000	-2.620109000	-2.110538000
1	2.968611000	-3.322148000	-1.117674000
1	2.760968000	-1.723000000	-1.881650000
7	7.077441000	0.813517000	-0.947534000
7	7.001109000	-0.413360000	-0.880566000
8	7.704974000	-1.422198000	-1.079341000

η -NO (constrained)

28	8.371732000	0.255442000	-0.326971000
15	7.162060000	2.102090000	-0.051057000
15	7.188766000	-1.475534000	0.122631000
6	6.734142000	2.614740000	1.674250000

1	6.215353000	3.580188000	1.702695000
1	6.099889000	1.857504000	2.146063000
1	7.657100000	2.693143000	2.258197000
6	8.081448000	3.578870000	-0.667978000
1	8.289764000	3.454176000	-1.735117000
1	7.521932000	4.509069000	-0.515632000
1	9.044555000	3.646296000	-0.152943000
6	5.553613000	2.270700000	-0.950981000
1	4.848237000	1.502993000	-0.616005000
1	5.096875000	3.256618000	-0.804185000
1	5.730800000	2.121107000	-2.021240000
6	5.795260000	-1.363051000	1.332927000
1	5.035706000	-0.658781000	0.976784000
1	5.319169000	-2.336411000	1.498664000
1	6.180813000	-0.995039000	2.289567000
6	8.210721000	-2.832850000	0.830025000
1	7.676724000	-3.789510000	0.851175000
1	9.123936000	-2.920400000	0.228842000
1	8.503371000	-2.564347000	1.850785000
6	6.394324000	-2.265694000	-1.346310000
1	7.170696000	-2.515377000	-2.076343000

1	5.854260000	-3.180118000	-1.074628000
1	5.696829000	-1.564417000	-1.815679000
7	9.978914000	-0.460750000	-0.807618000
7	10.668441000	-1.402402000	-0.983941000
8	10.046196000	0.834403000	-0.897358000

[Ni(PH₃)₂(N₂O)]

κ-N

28	-0.643771000	0.126938000	0.071479000
15	-1.739599000	1.923641000	-0.031528000
15	-1.779718000	-1.644962000	0.168721000
1	-1.227232000	-2.959405000	0.213292000
1	-2.672373000	-1.858514000	1.254897000
1	-2.715701000	-1.952463000	-0.856803000
1	-2.669811000	2.135330000	-1.086137000
1	-1.157552000	3.222003000	-0.133806000
1	-2.625742000	2.277639000	1.022827000
7	1.123182000	0.104083000	0.016617000
7	2.217631000	0.072642000	-0.323227000
8	3.385552000	0.042355000	-0.629056000

κ-O

28	-0.667513000	-0.096803000	-0.042343000
----	--------------	--------------	--------------

15	-1.469449000	1.802129000	-0.043900000
15	-1.897455000	-1.743833000	0.145820000
1	-1.530772000	-3.129046000	0.180832000
1	-2.743189000	-1.844086000	1.287567000
1	-2.927869000	-1.944950000	-0.816759000
1	-2.439005000	2.138202000	-1.032109000
1	-0.788811000	3.060235000	-0.156982000
1	-2.243932000	2.222998000	1.075297000
7	2.189534000	0.360076000	-0.313672000
8	1.239427000	-0.383769000	-0.231058000
7	3.100412000	1.033914000	-0.393178000

η -NN

28	-0.995579000	0.309165000	0.051600000
15	-2.196648000	2.122489000	-0.092930000
15	-2.256365000	-1.432417000	0.134774000
1	-1.590611000	-2.676936000	0.230489000
1	-3.173608000	-1.642769000	1.194785000
1	-3.124785000	-1.780036000	-0.930211000
1	-3.052290000	2.376040000	-1.194720000
1	-1.475394000	3.337284000	-0.152994000
1	-3.104296000	2.511551000	0.924476000

7	0.747456000	0.968827000	0.049463000
7	0.710937000	-0.255961000	0.127889000
8	1.444133000	-1.246405000	0.209164000

η -NO (constrained)

28	-1.062841000	0.250814000	0.053524000
15	-2.344848000	2.059593000	-0.095864000
15	-2.265627000	-1.501736000	0.154835000
1	-1.527741000	-2.707650000	0.253399000
1	-3.164050000	-1.743512000	1.224520000
1	-3.125941000	-1.893948000	-0.901517000
1	-3.214030000	2.291140000	-1.192203000
1	-1.657989000	3.295666000	-0.169791000
1	-3.251906000	2.438853000	0.926126000
7	0.629340000	-0.446430000	0.132094000
7	1.313148000	-1.395442000	0.209151000
8	0.692654000	0.849092000	0.044832000

[Ni(CNMe)₂(N₂O)]

1	-3.067067000	-2.777803000	2.256621000
1	-1.752096000	-2.439794000	3.408420000
1	-3.252306000	-1.488580000	3.470945000

6	-1.311107000	-0.618611000	-4.280974000
1	-0.341261000	-0.884778000	-4.716541000
1	-1.944727000	-1.513601000	-4.265233000
1	-1.783671000	0.133062000	-4.923712000
7	0.474954000	1.890555000	0.291828000
7	1.494935000	2.336944000	0.555396000
8	2.557917000	2.834834000	0.836288000
28	-0.829782000	0.723694000	-0.159989000
6	-1.584156000	-0.259477000	1.115158000
6	-1.064526000	0.317235000	-1.875041000
7	-1.986149000	-1.046151000	1.901616000
7	-1.137990000	-0.105166000	-2.977381000
6	-2.540999000	-1.983440000	2.799700000

κ-O

28	-1.717791000	4.417905000	0.133446000
6	-1.985515000	2.704503000	0.089629000
6	-0.183359000	5.240067000	0.221466000
7	-2.030050000	1.511989000	0.167337000
7	0.921591000	5.652241000	0.375104000
6	-1.453570000	0.212815000	0.188678000
1	-1.762174000	-0.321559000	1.094210000

1	-1.797744000	-0.365919000	-0.676233000
1	-0.357112000	0.261477000	0.165596000
6	2.327482000	5.773906000	0.467110000
1	2.706599000	6.413626000	-0.338541000
1	2.607613000	6.230594000	1.423359000
1	2.814668000	4.793022000	0.392877000
7	-4.507570000	5.263066000	0.357709000
8	-3.356975000	5.623260000	0.334705000
7	-5.588002000	4.921427000	0.382738000

η -NN

1	-5.321995000	2.211817000	-3.505499000
1	-5.594407000	1.091246000	-2.147499000
1	-5.254447000	2.811821000	-1.827974000
6	0.690674000	-1.941789000	-4.821855000
1	0.188080000	-2.887846000	-4.596306000
1	0.515950000	-1.688523000	-5.872490000
1	1.766217000	-2.060785000	-4.657512000
7	-0.199162000	2.546689000	-0.878365000
7	0.764266000	1.897774000	-1.264627000
8	1.984067000	1.787154000	-1.097441000
28	-0.688618000	1.265391000	-2.169017000

6	-2.484301000	1.486497000	-2.361347000
6	-0.182592000	-0.057499000	-3.277820000
7	-3.637718000	1.682421000	-2.435307000
7	0.186746000	-0.915398000	-3.986995000
6	-5.024899000	1.964373000	-2.481447000

η -NO (constrained)

28	-1.843051000	0.716617000	0.303002000
6	-2.630034000	-0.915539000	0.680651000
6	-0.126677000	0.572949000	0.748644000
7	-3.102591000	-1.988679000	0.637712000
7	1.015479000	0.467413000	1.000688000
6	-3.726738000	-3.260573000	0.620462000
1	-4.607885000	-3.233584000	-0.028075000
1	-3.028672000	-4.014763000	0.243799000
1	-4.037100000	-3.536663000	1.633268000
6	2.398494000	0.414278000	1.295190000
1	2.971871000	0.862046000	0.477034000
1	2.604777000	0.968745000	2.216569000
1	2.717539000	-0.624809000	1.427511000
7	-2.005615000	2.373252000	-0.566134000
7	-1.503401000	3.359274000	-0.948528000

8 -3.155283000 1.768035000 -0.521700000

[Ni(CO)₂(N₂O)]

κ-N

28 0.133079000 0.709473000 0.131772000

6 -0.560090000 2.279021000 0.471436000

8 -0.896253000 3.371531000 0.622835000

6 1.868883000 0.583154000 -0.042559000

8 3.008957000 0.644989000 -0.203532000

7 -0.952614000 -0.705349000 -0.330872000

7 -1.593055000 -1.446127000 -0.913304000

8 -2.269338000 -2.237406000 -1.498408000

κ-O

28 -0.002559000 1.019250000 0.032347000

6 -0.394210000 2.634920000 0.482144000

8 -0.407340000 3.742371000 0.816276000

6 1.635729000 0.498375000 0.033698000

8 2.784880000 0.380747000 0.110923000

7 -1.420949000 -1.413637000 -0.832022000

8 -1.507360000 -0.257682000 -0.492309000

7 -1.361636000 -2.495293000 -1.151856000

η-NN

28	-0.384295000	0.050650000	0.295663000
6	-1.003355000	1.733444000	0.415660000
8	-1.397100000	2.803786000	0.492050000
6	1.370090000	0.139711000	0.054996000
8	2.500913000	0.232620000	-0.095083000
7	-1.916687000	-1.043554000	0.494809000
7	-0.931908000	-1.733598000	0.342151000
8	-0.500603000	-2.874548000	0.260638000

η -NO

8	1.896787000	-0.120699000	0.027297000
7	-1.624903000	-1.883798000	0.295462000
7	-1.318630000	-2.989415000	0.434820000
8	-2.594567000	-1.024917000	0.136384000
28	-0.989641000	-0.003871000	0.193691000
6	-1.375962000	1.711310000	0.620752000
8	-1.470399000	2.723934000	1.147964000
6	0.750206000	-0.128250000	0.089923000

[Co(η^5 -Cp*)(CNMe)(N₂O)]

κ -N

6	-0.077040000	2.318317000	-0.162358000
6	-1.070686000	3.326605000	-0.111411000

6	-2.357152000	2.658256000	-0.111363000
6	-2.144952000	1.234454000	-0.053381000
6	-0.742059000	1.026126000	-0.145842000
27	-1.376003000	2.131262000	-1.811666000
7	-0.300720000	1.410906000	-3.068295000
7	0.439373000	1.009898000	-3.837446000
8	1.172641000	0.504416000	-4.656614000
6	-2.466275000	2.922183000	-2.941177000
7	-3.147093000	3.574507000	-3.678241000
6	-4.319990000	3.611439000	-4.498058000
1	-4.867971000	2.661040000	-4.463444000
1	-4.982082000	4.415696000	-4.160859000
1	-4.038707000	3.820898000	-5.535281000
6	-3.684991000	3.349454000	0.027008000
1	-4.501227000	2.739671000	-0.372722000
1	-3.909740000	3.551699000	1.083261000
1	-3.699215000	4.306919000	-0.502537000
6	-0.856313000	4.809846000	-0.012292000
1	-1.621226000	5.366308000	-0.562898000
1	-0.894141000	5.147697000	1.033705000
1	0.116301000	5.106139000	-0.417401000

6	1.411994000	2.519432000	-0.157270000
1	1.684981000	3.524321000	-0.493772000
1	1.827039000	2.384870000	0.851638000
1	1.923369000	1.805292000	-0.812100000
6	-0.035915000	-0.300628000	-0.119378000
1	0.862358000	-0.297715000	-0.746062000
1	0.284470000	-0.555961000	0.900348000
1	-0.681518000	-1.110393000	-0.473409000
6	-3.203004000	0.183885000	0.130037000
1	-2.909646000	-0.770443000	-0.318777000
1	-3.398630000	-0.002535000	1.196584000
1	-4.151954000	0.480839000	-0.327175000

κ-O

6	-0.071626000	2.271821000	-0.067164000
6	-1.023980000	3.320165000	-0.038440000
6	-2.336191000	2.703245000	-0.084577000
6	-2.179513000	1.266046000	-0.017440000
6	-0.786450000	1.005353000	-0.065645000
27	-1.345900000	2.150877000	-1.730132000
7	-0.038324000	1.366127000	-4.196503000
8	-0.022247000	1.303081000	-2.992207000

7	-0.028523000	1.398923000	-5.328154000
6	-2.320837000	2.917457000	-2.948759000
7	-2.910075000	3.575056000	-3.780685000
6	-4.084433000	3.368338000	-4.596887000
1	-4.526996000	2.373904000	-4.449204000
1	-4.834114000	4.129642000	-4.355401000
1	-3.822153000	3.487994000	-5.653552000
6	-3.640553000	3.443462000	0.009808000
1	-4.463202000	2.865824000	-0.423314000
1	-3.898139000	3.652468000	1.057600000
1	-3.598700000	4.400923000	-0.518474000
6	-0.754869000	4.792843000	0.077849000
1	-1.475460000	5.380645000	-0.499123000
1	-0.820057000	5.126694000	1.124015000
1	0.243680000	5.052884000	-0.287639000
6	1.424773000	2.408724000	-0.033286000
1	1.745553000	3.412136000	-0.330473000
1	1.819782000	2.221939000	0.975640000
1	1.915134000	1.695739000	-0.706019000
6	-0.124975000	-0.344676000	-0.041300000
1	0.736667000	-0.388998000	-0.717252000

1	0.244094000	-0.588197000	0.964994000
1	-0.816464000	-1.139371000	-0.338553000
6	-3.285444000	0.262676000	0.145151000
1	-3.010888000	-0.715346000	-0.263051000
1	-3.536385000	0.115980000	1.206531000
1	-4.199677000	0.582713000	-0.364071000

η-NN

6	-0.071626000	2.271821000	-0.067164000
6	-1.023980000	3.320165000	-0.038440000
6	-2.336191000	2.703245000	-0.084577000
6	-2.179513000	1.266046000	-0.017440000
6	-0.786450000	1.005353000	-0.065645000
27	-1.345900000	2.150877000	-1.730132000
7	-0.038324000	1.366127000	-4.196503000
8	-0.022247000	1.303081000	-2.992207000
7	-0.028523000	1.398923000	-5.328154000
6	-2.320837000	2.917457000	-2.948759000
7	-2.910075000	3.575056000	-3.780685000
6	-4.084433000	3.368338000	-4.596887000
1	-4.526996000	2.373904000	-4.449204000
1	-4.834114000	4.129642000	-4.355401000

1	-3.822153000	3.487994000	-5.653552000
6	-3.640553000	3.443462000	0.009808000
1	-4.463202000	2.865824000	-0.423314000
1	-3.898139000	3.652468000	1.057600000
1	-3.598700000	4.400923000	-0.518474000
6	-0.754869000	4.792843000	0.077849000
1	-1.475460000	5.380645000	-0.499123000
1	-0.820057000	5.126694000	1.124015000
1	0.243680000	5.052884000	-0.287639000
6	1.424773000	2.408724000	-0.033286000
1	1.745553000	3.412136000	-0.330473000
1	1.819782000	2.221939000	0.975640000
1	1.915134000	1.695739000	-0.706019000
6	-0.124975000	-0.344676000	-0.041300000
1	0.736667000	-0.388998000	-0.717252000
1	0.244094000	-0.588197000	0.964994000
1	-0.816464000	-1.139371000	-0.338553000
6	-3.285444000	0.262676000	0.145151000
1	-3.010888000	-0.715346000	-0.263051000
1	-3.536385000	0.115980000	1.206531000
1	-4.199677000	0.582713000	-0.364071000

η-NO

6	-0.206415000	2.205115000	-0.293455000
6	-1.256052000	3.170844000	-0.327489000
6	-2.494454000	2.514154000	0.084340000
6	-2.207281000	1.148116000	0.291923000
6	-0.806532000	0.938717000	0.001689000
27	-1.695078000	1.701598000	-1.714658000
7	-0.633644000	0.781237000	-2.947822000
7	0.369447000	0.667593000	-3.532603000
8	-1.786635000	0.055307000	-2.719890000
6	-2.588840000	2.677794000	-2.945360000
7	-3.192731000	3.299870000	-3.735808000
6	-3.902933000	3.991085000	-4.747761000
1	-4.969935000	3.751530000	-4.691455000
1	-3.773165000	5.071023000	-4.625591000
1	-3.523419000	3.696714000	-5.731229000
6	-3.803485000	3.214785000	0.311874000
1	-4.644741000	2.515663000	0.296023000
1	-3.811949000	3.721560000	1.286818000
1	-3.995155000	3.976377000	-0.451151000
6	-1.102123000	4.649440000	-0.551800000

1	-1.987258000	5.079586000	-1.031274000
1	-0.958729000	5.177636000	0.400954000
1	-0.242067000	4.873768000	-1.189245000
6	1.239646000	2.427102000	-0.633467000
1	1.477209000	3.493501000	-0.684449000
1	1.891847000	1.979825000	0.125708000
1	1.497078000	1.978641000	-1.603152000
6	-0.102934000	-0.385215000	0.084281000
1	0.910198000	-0.322643000	-0.321498000
1	-0.031500000	-0.730817000	1.124035000
1	-0.639423000	-1.147921000	-0.490126000
6	-3.159907000	0.055798000	0.680131000
1	-3.085206000	-0.792876000	-0.009186000
1	-2.939783000	-0.319140000	1.688633000
1	-4.198114000	0.399820000	0.673971000

[Co(η^5 -Cp)(CNMe)(N₂O)]

κ-N			
6	-0.093369000	2.479841000	-0.126377000
6	-1.182112000	3.379455000	-0.057993000
6	-2.373031000	2.583820000	-0.059454000
6	-2.009422000	1.192812000	-0.011744000

6	-0.602174000	1.131470000	-0.082658000
1	0.953347000	2.749725000	-0.188328000
1	-1.130620000	4.459294000	-0.041850000
1	-3.385121000	2.966136000	-0.032016000
1	-2.693298000	0.357006000	0.042708000
1	0.001168000	0.233098000	-0.103189000
27	-1.359610000	2.146679000	-1.772495000
7	-0.150475000	1.768099000	-3.056423000
7	0.624982000	1.489695000	-3.841953000
8	1.459713000	1.226704000	-4.674874000
6	-2.694157000	2.448849000	-2.884057000
7	-3.636769000	2.561003000	-3.601345000
6	-4.515865000	3.309937000	-4.436248000
1	-4.499212000	2.902598000	-5.452377000
1	-5.540710000	3.235619000	-4.058195000
1	-4.225925000	4.367696000	-4.472514000

κ-O

6	-0.136658000	2.433255000	0.070499000
6	-1.248850000	3.301506000	0.090653000
6	-2.431587000	2.483503000	-0.009520000
6	-2.042490000	1.102160000	0.001117000

6	-0.628952000	1.080607000	-0.006762000
1	0.906019000	2.721967000	0.104913000
1	-1.227073000	4.380914000	0.153498000
1	-3.448165000	2.853650000	-0.023483000
1	-2.703431000	0.246469000	-0.007181000
1	-0.008228000	0.194149000	-0.056641000
27	-1.322377000	2.182878000	-1.657110000
7	0.060456000	1.644459000	-4.138658000
8	0.079083000	1.536488000	-2.937740000
7	0.076949000	1.722719000	-5.267774000
6	-2.372350000	2.883818000	-2.853747000
7	-3.042311000	3.483304000	-3.660498000
6	-4.190259000	3.226335000	-4.493470000
1	-4.552700000	2.194705000	-4.393123000
1	-4.997663000	3.914711000	-4.222302000
1	-3.932847000	3.414327000	-5.541117000

η-NN

6	-0.418612000	2.885047000	-0.997074000
6	-1.464705000	3.835662000	-1.043295000
6	-2.704325000	3.157151000	-0.727623000
6	-2.410532000	1.796867000	-0.508598000

6	-1.008386000	1.614215000	-0.723844000
1	0.626221000	3.070130000	-1.207206000
1	-1.360420000	4.891763000	-1.253964000
1	-3.679228000	3.622277000	-0.668146000
1	-3.124935000	1.012870000	-0.296398000
1	-0.482326000	0.669602000	-0.678595000
27	-1.837529000	2.379974000	-2.484840000
7	-1.952998000	0.731013000	-3.546228000
7	-0.922911000	1.334959000	-3.745181000
8	0.146450000	1.293414000	-4.361531000
6	-2.659971000	3.397779000	-3.726398000
7	-3.206738000	4.049414000	-4.532901000
6	-3.841246000	4.800407000	-5.552426000
1	-3.116111000	5.064915000	-6.328873000
1	-4.647845000	4.212614000	-6.002074000
1	-4.262227000	5.718202000	-5.130467000

η-NO

6	-0.218644000	2.449515000	-0.423533000
6	-1.413723000	3.211422000	-0.372034000
6	-2.463278000	2.370903000	0.163547000
6	-1.921809000	1.093163000	0.373523000

6	-0.544709000	1.121928000	-0.030867000
1	0.739609000	2.788196000	-0.794509000
1	-1.514434000	4.256833000	-0.632309000
1	-3.486140000	2.675038000	0.341649000
1	-2.461371000	0.216020000	0.705702000
1	0.131562000	0.277448000	-0.021683000
27	-1.694392000	1.641904000	-1.704783000
7	-0.645515000	0.841550000	-3.060222000
7	0.272213000	0.887075000	-3.773695000
8	-1.635711000	-0.012290000	-2.680916000
6	-2.812148000	2.441203000	-2.876694000
7	-3.558046000	2.937808000	-3.631253000
6	-4.442487000	3.499591000	-4.584699000
1	-5.396970000	2.963579000	-4.573102000
1	-4.621572000	4.553431000	-4.350442000
1	-4.003439000	3.426104000	-5.584502000

[Co(η^5 -Cp)(PMe₃)(N₂O)]

κ-N			
6	0.132163000	1.490694000	-0.220354000
6	-0.336195000	2.750451000	0.286965000
6	-1.748236000	2.707170000	0.370961000

6	-2.152881000	1.428089000	-0.139056000
6	-0.988665000	0.655061000	-0.453262000
1	1.168056000	1.235310000	-0.401539000
1	0.293281000	3.592455000	0.545074000
1	-2.400640000	3.484255000	0.744933000
1	-3.174462000	1.073091000	-0.201460000
1	-0.973636000	-0.364387000	-0.813580000
27	-1.124517000	2.403428000	-1.620540000
15	-2.767377000	2.463014000	-3.033363000
7	0.008673000	3.177957000	-2.709994000
7	0.750375000	3.863418000	-3.295045000
8	1.650194000	3.995510000	-4.124713000
6	-4.306100000	3.321002000	-2.462562000
1	-5.097172000	3.285978000	-3.220793000
1	-4.676606000	2.854960000	-1.544226000
1	-4.071821000	4.366658000	-2.240064000
6	-2.439973000	3.320101000	-4.634893000
1	-1.613849000	2.829115000	-5.157365000
1	-3.325975000	3.309826000	-5.279352000
1	-2.150712000	4.358138000	-4.445081000
6	-3.430599000	0.832118000	-3.607209000

1	-3.755574000	0.237636000	-2.747523000
1	-4.276737000	0.954195000	-4.293748000
1	-2.633720000	0.281692000	-4.116817000

κ-O

6	0.171251000	1.930410000	0.067940000
6	-0.492095000	3.179709000	-0.035579000
6	-1.889443000	2.901485000	0.122991000
6	-2.064936000	1.502596000	0.439437000
6	-0.788322000	0.903323000	0.389710000
1	1.236893000	1.768252000	-0.046751000
1	-0.035931000	4.149688000	-0.178597000
1	-2.681680000	3.640618000	0.106902000
1	-2.999720000	1.019584000	0.691239000
1	-0.557567000	-0.139695000	0.566384000
27	-1.248421000	1.812542000	-1.450253000
15	-2.805020000	2.350724000	-2.817150000
7	0.355934000	1.003477000	-3.690835000
8	-0.154415000	0.661885000	-2.649041000
7	0.839805000	1.239700000	-4.689830000
6	-4.537024000	2.027068000	-2.231582000
1	-5.288538000	2.381027000	-2.947780000

1	-4.669586000	0.952396000	-2.073310000
1	-4.699665000	2.530269000	-1.273246000
6	-2.960539000	4.130632000	-3.332222000
1	-2.058456000	4.431254000	-3.874775000
1	-3.837602000	4.311288000	-3.966440000
1	-3.034756000	4.757915000	-2.437938000
6	-2.851965000	1.517480000	-4.478095000
1	-2.859340000	0.430271000	-4.348292000
1	-3.746373000	1.807216000	-5.041560000
1	-1.974849000	1.791212000	-5.074065000

η-NN

6	-0.194517000	2.651509000	-0.460765000
6	-1.501707000	3.210892000	-0.266835000
6	-2.374380000	2.199505000	0.236907000
6	-1.618979000	0.998097000	0.290731000
6	-0.280638000	1.276241000	-0.120663000
1	0.684033000	3.179132000	-0.805804000
1	-1.766972000	4.249431000	-0.423430000
1	-3.414940000	2.321907000	0.503696000
1	-2.001093000	0.020784000	0.555291000
1	0.515200000	0.548023000	-0.207460000

27	-1.619923000	1.723531000	-1.691736000
15	-2.276014000	3.052404000	-3.309605000
7	-2.290576000	0.101451000	-2.564670000
7	-1.143448000	0.377945000	-2.861336000
8	-0.195806000	-0.032481000	-3.553246000
6	-1.127131000	4.431905000	-3.744395000
1	-0.955681000	5.071365000	-2.872789000
1	-0.165286000	4.011143000	-4.053501000
1	-1.526022000	5.045920000	-4.559874000
6	-3.884090000	3.923208000	-3.045976000
1	-4.147901000	4.552171000	-3.903947000
1	-4.671850000	3.178821000	-2.894216000
1	-3.831452000	4.550579000	-2.150699000
6	-2.530016000	2.205242000	-4.923753000
1	-1.607287000	1.693592000	-5.212450000
1	-3.311785000	1.448760000	-4.812856000
1	-2.816154000	2.917766000	-5.704700000

η-NO

6	-0.111594000	2.541751000	-0.350930000
6	-1.390242000	3.145631000	-0.228314000
6	-2.303833000	2.170899000	0.323941000

6	-1.592973000	0.968007000	0.471225000
6	-0.245427000	1.177089000	0.030485000
1	0.780933000	3.011225000	-0.743037000
1	-1.622053000	4.181755000	-0.440670000
1	-3.344933000	2.338408000	0.564822000
1	-2.006688000	0.019280000	0.787678000
1	0.530783000	0.424662000	-0.005916000
27	-1.511120000	1.575199000	-1.590347000
15	-2.861415000	2.528323000	-3.043997000
7	-0.532291000	0.843911000	-2.968411000
7	0.312097000	0.881767000	-3.771471000
8	-1.504669000	-0.085671000	-2.529003000
6	-4.384560000	3.340646000	-2.389027000
1	-4.118246000	4.119069000	-1.667553000
1	-4.969986000	3.794726000	-3.196350000
1	-5.004174000	2.596652000	-1.878791000
6	-3.520113000	1.371895000	-4.313393000
1	-4.177513000	1.893483000	-5.017731000
1	-2.687229000	0.918099000	-4.855694000
1	-4.070938000	0.567402000	-3.819343000
6	-2.068571000	3.858480000	-4.046724000

1	-2.743215000	4.246314000	-4.818278000
1	-1.760319000	4.682521000	-3.395272000
1	-1.172760000	3.446293000	-4.521945000

[Co(η^5 -Cp)(PH₃)(N₂O)]

κ-N			
6	0.037099000	2.061459000	-0.001231000
6	-0.719276000	3.276183000	0.044009000
6	-2.083063000	2.912006000	-0.048934000
6	-2.170314000	1.467709000	-0.083320000
6	-0.858846000	0.943215000	-0.014288000
1	1.118094000	1.995954000	0.004752000
1	-0.318985000	4.278355000	0.114481000
1	-2.927436000	3.589319000	-0.050650000
1	-3.088755000	0.896173000	-0.116991000
1	-0.578186000	-0.101040000	0.000102000
27	-1.095239000	2.169042000	-1.711318000
1	-2.902586000	1.398114000	-4.171266000
1	-3.994395000	2.448572000	-2.690541000
1	-2.738789000	3.499201000	-4.036663000
15	-2.663141000	2.375655000	-3.167437000
7	0.213898000	2.060855000	-2.904669000

7	1.124791000	2.159799000	-3.603230000
8	2.030883000	1.865463000	-4.362051000

$\kappa\text{-O}$

6	-0.015762000	2.375164000	0.160805000
6	-1.009690000	3.374634000	0.067106000
6	-2.266566000	2.688822000	-0.084956000
6	-2.054499000	1.269677000	0.035963000
6	-0.658721000	1.079721000	0.141847000
1	1.050880000	2.539697000	0.251674000
1	-0.863445000	4.445437000	0.106768000
1	-3.233108000	3.169796000	-0.176397000
1	-2.819469000	0.505307000	0.048518000
1	-0.145893000	0.128693000	0.216585000
27	-1.076476000	2.120312000	-1.606486000
1	-1.921822000	2.586059000	-4.539941000
1	-3.648441000	2.244835000	-3.378057000
1	-2.719776000	4.130549000	-3.345331000
15	-2.335113000	2.766625000	-3.187713000
7	0.655661000	1.299375000	-3.841457000
8	0.432158000	1.389023000	-2.660206000
7	0.905065000	1.194700000	-4.945037000

η-NN

6	-0.111233000	2.446441000	-0.292399000
6	-1.323385000	3.193571000	-0.244997000
6	-2.396775000	2.325416000	0.165477000
6	-1.849358000	1.032425000	0.303494000
6	-0.448132000	1.095953000	0.007965000
1	0.868290000	2.825531000	-0.549214000
1	-1.413180000	4.256832000	-0.432338000
1	-3.427827000	2.612997000	0.319165000
1	-2.398784000	0.128340000	0.530689000
1	0.232275000	0.254502000	0.003902000
27	-1.510584000	1.709500000	-1.663477000
1	-2.468159000	2.414083000	-4.506305000
1	-3.495791000	3.676921000	-3.118275000
1	-1.506793000	4.147527000	-3.694541000
15	-2.246299000	3.022821000	-3.251847000
7	-1.987071000	0.084280000	-2.647991000
7	-0.847664000	0.451996000	-2.855782000
8	0.186993000	0.140713000	-3.459082000

η-NO

6	0.153826000	2.239560000	-0.223368000
---	-------------	-------------	--------------

6	-0.782992000	3.301837000	-0.244490000
6	-2.053879000	2.791077000	0.225459000
6	-1.900112000	1.417109000	0.477562000
6	-0.546933000	1.057650000	0.158642000
1	1.189434000	2.294523000	-0.532133000
1	-0.574807000	4.329919000	-0.510995000
1	-2.958898000	3.370884000	0.353442000
1	-2.677043000	0.732096000	0.790485000
1	-0.129133000	0.061324000	0.216381000
27	-1.399864000	1.815668000	-1.559307000
1	-2.084184000	2.873952000	-4.421432000
1	-3.862395000	2.728559000	-3.252780000
1	-2.551619000	4.391395000	-2.992821000
15	-2.481996000	2.980547000	-3.069026000
8	-1.746389000	0.233201000	-2.530502000
7	-0.534191000	0.726666000	-2.905080000
7	0.358632000	0.452494000	-3.600604000

[Co(η^5 -Cp)(CO)(N₂O)]

κ -N

6	-0.064953000	2.386073000	-0.009504000
6	-1.096342000	3.346363000	0.014675000

6	-2.331176000	2.634944000	-0.171935000
6	-2.061337000	1.223532000	-0.191325000
6	-0.660634000	1.075670000	-0.136947000
1	0.996697000	2.585140000	0.060766000
1	-0.984543000	4.417028000	0.114917000
1	-3.314486000	3.085649000	-0.210062000
1	-2.795060000	0.433538000	-0.271495000
1	-0.113622000	0.142461000	-0.176751000
27	-1.171488000	2.266005000	-1.799663000
7	0.120060000	1.790620000	-2.980695000
7	0.947984000	1.481125000	-3.695650000
8	1.826290000	1.152681000	-4.449561000
6	-2.286549000	2.889971000	-2.970099000
8	-3.063764000	3.318105000	-3.714482000

κ-Ω

6	-0.120110000	2.351463000	0.182866000
6	-1.112757000	3.343010000	0.055884000
6	-2.344035000	2.661602000	-0.250816000
6	-2.121607000	1.240461000	-0.188169000
6	-0.742190000	1.055105000	0.031028000
1	0.933161000	2.520179000	0.368934000

1	-0.978431000	4.412303000	0.143460000
1	-3.300663000	3.139922000	-0.416013000
1	-2.869315000	0.469575000	-0.313606000
1	-0.225139000	0.105203000	0.085172000
27	-1.036784000	2.205589000	-1.709016000
7	0.760923000	1.568877000	-3.889155000
8	0.565860000	1.539833000	-2.698578000
7	0.988282000	1.574755000	-4.997429000
6	-1.949746000	2.812630000	-3.036266000
8	-2.611139000	3.240788000	-3.893174000

η-NN

6	-0.436746000	2.950379000	-1.064778000
6	-1.543699000	3.831626000	-1.079163000
6	-2.720692000	3.086639000	-0.676680000
6	-2.330217000	1.754099000	-0.441814000
6	-0.931495000	1.654610000	-0.729630000
1	0.581827000	3.194287000	-1.334579000
1	-1.519086000	4.886181000	-1.320127000
1	-3.717953000	3.493446000	-0.575347000
1	-2.980344000	0.932895000	-0.172200000
1	-0.345677000	0.745379000	-0.691794000

27	-1.900196000	2.319050000	-2.454645000
7	-1.967433000	0.620570000	-3.447321000
7	-0.977323000	1.252276000	-3.721915000
8	0.070859000	1.259555000	-4.361656000
6	-2.817149000	3.250293000	-3.634706000
8	-3.440967000	3.885289000	-4.362080000

η-NO

6	-0.137221000	2.240481000	-0.215817000
6	-1.151518000	3.229921000	-0.308479000
6	-2.430129000	2.596322000	-0.056049000
6	-2.198183000	1.223026000	0.113823000
6	-0.786556000	0.990772000	-0.028409000
1	0.923025000	2.395469000	-0.364604000
1	-0.995933000	4.288015000	-0.473228000
1	-3.388482000	3.097719000	-0.034599000
1	-2.950359000	0.458135000	0.254277000
1	-0.302760000	0.023829000	0.016122000
27	-1.467607000	1.846490000	-1.812269000
7	-0.329939000	0.854633000	-3.031001000
7	0.714661000	0.750264000	-3.523799000
8	-1.515982000	0.260408000	-2.910227000

6	-2.106864000	2.939933000	-3.033807000
8	-2.567415000	3.684374000	-3.779186000

[Co(η^5 -Cp)py(N₂O)]

κ-N			
6	-0.112042000	2.814938000	-0.109584000
6	-1.322855000	3.548891000	-0.080292000
6	-2.391874000	2.593993000	0.023529000
6	-1.846116000	1.284234000	0.132478000
6	-0.436779000	1.421471000	-0.011786000
1	0.884802000	3.222468000	-0.213373000
1	-1.425271000	4.625676000	-0.114993000
1	-3.447623000	2.836533000	0.048179000
1	-2.398229000	0.364718000	0.270049000
1	0.281451000	0.611997000	-0.038639000
27	-1.324578000	2.209627000	-1.708416000
7	-0.230645000	1.635681000	-2.958138000
7	0.463944000	1.052180000	-3.699882000
8	1.405311000	1.077960000	-4.493035000
7	-2.698622000	2.687175000	-2.972212000
6	-3.388567000	3.847627000	-2.876662000
6	-3.036742000	1.849527000	-3.980043000

6	-4.411045000	4.197467000	-3.749085000
1	-3.088886000	4.509650000	-2.073945000
6	-4.039239000	2.134646000	-4.898897000
1	-2.469727000	0.928977000	-4.039197000
6	-4.750028000	3.329193000	-4.787029000
1	-4.922357000	5.145420000	-3.615623000
1	-4.256583000	1.419323000	-5.685526000
1	-5.540506000	3.577845000	-5.488567000

κ-O

6	-0.198368000	2.243019000	0.159723000
6	-1.316367000	3.083007000	0.404551000
6	-2.472431000	2.331196000	0.017066000
6	-2.072584000	0.992140000	-0.340770000
6	-0.660392000	0.948173000	-0.265702000
1	0.841801000	2.518277000	0.288549000
1	-1.301025000	4.088068000	0.803635000
1	-3.494065000	2.691227000	0.052743000
1	-2.733165000	0.169944000	-0.585565000
1	-0.029211000	0.094931000	-0.480340000
27	-1.279559000	2.448498000	-1.603117000
7	0.421212000	2.303840000	-3.982662000

8	0.302798000	2.545468000	-2.804758000
7	0.579522000	2.116553000	-5.092081000
7	-2.479460000	3.016584000	-2.919096000
6	-3.735216000	2.489782000	-3.034036000
6	-2.165190000	3.998318000	-3.816765000
6	-4.650468000	2.893765000	-3.991636000
1	-3.977159000	1.707714000	-2.326085000
6	-3.020570000	4.427826000	-4.819153000
1	-1.196412000	4.466630000	-3.681951000
6	-4.299971000	3.876223000	-4.922887000
1	-5.629045000	2.423479000	-4.013049000
1	-2.687770000	5.209695000	-5.495288000
1	-4.995234000	4.202174000	-5.689600000

η-NN

6	-0.542964000	3.391957000	-1.116383000
6	-1.890574000	3.840649000	-0.895820000
6	-2.674170000	2.757933000	-0.419818000
6	-1.831958000	1.608060000	-0.428252000
6	-0.517969000	2.004094000	-0.824962000
1	0.295529000	4.000529000	-1.428002000
1	-2.247589000	4.850795000	-1.056860000

1	-3.721971000	2.789071000	-0.154705000
1	-2.137130000	0.596057000	-0.196841000
1	0.331611000	1.342947000	-0.935979000
27	-1.892981000	2.359800000	-2.368811000
7	-2.628503000	0.782767000	-3.249238000
7	-1.461701000	0.992588000	-3.529462000
8	-0.523870000	0.526995000	-4.192711000
7	-2.426061000	3.522655000	-3.843207000
6	-1.706406000	4.615143000	-4.175166000
6	-3.552420000	3.270774000	-4.542885000
6	-2.084180000	5.486878000	-5.189193000
1	-0.798173000	4.775246000	-3.606848000
6	-3.993589000	4.094338000	-5.573582000
1	-4.082023000	2.366917000	-4.266779000
6	-3.252608000	5.227072000	-5.905309000
1	-1.462243000	6.347220000	-5.413594000
1	-4.906952000	3.841803000	-6.102092000
1	-3.572579000	5.887406000	-6.705904000

η-NO

NA

[Co(η⁵-Cp)py(p-NMe)₂(N₂O)]

κ -N

6	-0.135729000	2.903356000	-0.041238000
6	-1.455361000	3.414694000	0.052361000
6	-2.339023000	2.284397000	0.096290000
6	-1.573871000	1.083999000	0.105997000
6	-0.211586000	1.472134000	-0.035889000
1	0.773237000	3.483677000	-0.128805000
1	-1.743058000	4.456968000	0.094965000
1	-3.420359000	2.337470000	0.139785000
1	-1.957563000	0.075817000	0.181867000
1	0.635188000	0.803723000	-0.124668000
27	-1.250336000	2.214142000	-1.666826000
7	-0.119231000	1.976138000	-2.978423000
7	0.620030000	1.585509000	-3.808803000
8	1.472701000	1.920082000	-4.640601000
7	-2.745893000	2.468117000	-2.878144000
6	-3.608721000	3.501095000	-2.766155000
6	-2.987654000	1.599185000	-3.884176000
6	-4.695099000	3.698486000	-3.597748000
1	-3.400490000	4.204918000	-1.968806000
6	-4.042651000	1.714558000	-4.770016000

1	-2.284296000	0.780559000	-3.979590000
6	-4.956497000	2.789113000	-4.651417000
1	-5.322609000	4.564192000	-3.428372000
1	-4.145131000	0.967672000	-5.546804000
7	-6.020874000	2.940460000	-5.500161000
6	-6.894777000	4.097609000	-5.371907000
1	-7.692165000	4.027393000	-6.112862000
1	-7.360688000	4.137924000	-4.379002000
1	-6.355955000	5.041176000	-5.537626000
6	-6.201605000	2.016512000	-6.611124000
1	-5.353170000	2.042508000	-7.308990000
1	-6.323796000	0.984879000	-6.257125000
1	-7.102674000	2.290872000	-7.161189000

κ-O

NA

η-NN

6	-0.797917000	3.511710000	-0.921590000
6	-2.222496000	3.609550000	-0.774122000
6	-2.751848000	2.333336000	-0.446370000
6	-1.660246000	1.418241000	-0.476417000
6	-0.459491000	2.144578000	-0.740598000

1	-0.113451000	4.325410000	-1.121030000
1	-2.801560000	4.518096000	-0.887735000
1	-3.790083000	2.095557000	-0.260062000
1	-1.726572000	0.345743000	-0.348628000
1	0.528365000	1.713134000	-0.836990000
27	-1.783192000	2.318260000	-2.355668000
7	-2.014179000	0.709979000	-3.433054000
7	-0.938572000	1.257048000	-3.601941000
8	0.120413000	1.133482000	-4.237335000
7	-2.548322000	3.446435000	-3.754908000
6	-2.108010000	4.699614000	-3.992282000
6	-3.565648000	3.008348000	-4.527245000
6	-2.643505000	5.537522000	-4.951846000
1	-1.277010000	5.033120000	-3.381228000
6	-4.165169000	3.771425000	-5.511828000
1	-3.880827000	1.986950000	-4.349184000
6	-3.719050000	5.092255000	-5.759400000
1	-2.213474000	6.523304000	-5.073852000
1	-4.971094000	3.330078000	-6.083911000
7	-4.286371000	5.886003000	-6.719094000
6	-3.744202000	7.211882000	-6.981808000

1	-4.340985000	7.694905000	-7.756518000
1	-3.782388000	7.843163000	-6.085007000
1	-2.702493000	7.168902000	-7.329095000
6	-5.342597000	5.355718000	-7.570729000
1	-6.207956000	5.032982000	-6.978297000
1	-5.676933000	6.137541000	-8.253839000
1	-4.997151000	4.501768000	-8.169647000

η -NO

NA



κ-N			
6	0.117234000	2.454012000	-0.249454000
6	-0.455147000	3.742124000	-0.431233000
6	-1.852524000	3.598753000	-0.197352000
6	-2.120843000	2.252685000	0.237507000
6	-0.899880000	1.544248000	0.199645000
1	1.157404000	2.196951000	-0.406716000
1	0.061377000	4.645078000	-0.725942000
1	-2.586848000	4.390000000	-0.285266000
1	-3.081792000	1.858745000	0.542498000
1	-0.750891000	0.502152000	0.449860000

27	-1.347110000	2.310844000	-1.717114000
7	-0.395046000	1.272019000	-2.853507000
7	0.252907000	0.510510000	-3.404191000
8	0.968989000	-0.235802000	-4.023321000
7	-2.740221000	2.655647000	-2.932176000
6	-4.019363000	2.898730000	-2.510326000
6	-2.542775000	2.695741000	-4.285669000
6	-5.072819000	3.172062000	-3.359381000
1	-4.182399000	2.849802000	-1.442857000
6	-3.542210000	2.945406000	-5.204613000
1	-1.527646000	2.530614000	-4.620807000
6	-4.833804000	3.193540000	-4.734905000
1	-6.067768000	3.353490000	-2.973294000
1	-3.331141000	2.965329000	-6.266176000
7	-5.912186000	3.470833000	-5.662750000
8	-5.640970000	3.480509000	-6.868512000
8	-7.034776000	3.679508000	-5.189273000

κ-O

6	1.293616000	-0.333188000	-4.979509000
6	1.716987000	1.017507000	-4.965327000
6	2.077696000	1.351906000	-3.608387000

6	1.926805000	0.191883000	-2.792676000
6	1.376953000	-0.817723000	-3.630894000
1	0.966115000	-0.898943000	-5.842002000
1	1.778720000	1.677051000	-5.821620000
1	2.446223000	2.313413000	-3.273026000
1	2.165880000	0.101776000	-1.742024000
1	1.097858000	-1.815039000	-3.308799000
27	0.124275000	0.841613000	-3.674857000
7	-1.811482000	-1.335338000	-3.459221000
8	-1.611219000	-0.146811000	-3.433617000
7	-2.038174000	-2.445489000	-3.490427000
7	-0.740537000	2.476588000	-3.518651000
6	-1.940634000	2.634572000	-2.866370000
6	-0.203832000	3.628171000	-4.044569000
6	-2.602620000	3.836056000	-2.744549000
1	-2.360314000	1.742303000	-2.421568000
6	-0.794484000	4.868806000	-3.952847000
1	0.733848000	3.513461000	-4.569598000
6	-2.023536000	4.985777000	-3.294213000
1	-3.546759000	3.899467000	-2.218660000
1	-0.326344000	5.740795000	-4.391697000

7	-2.677918000	6.263406000	-3.181160000
8	-2.117028000	7.246036000	-3.686820000
8	-3.763980000	6.304347000	-2.585404000

η-NN

6	-0.491622000	3.423074000	-1.222266000
6	-1.818699000	3.895974000	-0.928790000
6	-2.575948000	2.838254000	-0.365425000
6	-1.745552000	1.679671000	-0.389466000
6	-0.455604000	2.048739000	-0.881499000
1	0.329988000	4.009945000	-1.611267000
1	-2.176618000	4.904628000	-1.096682000
1	-3.605346000	2.886680000	-0.038258000
1	-2.044066000	0.679914000	-0.102938000
1	0.380712000	1.376665000	-1.022700000
27	-1.935035000	2.361708000	-2.342282000
7	-2.801851000	0.769287000	-3.067558000
7	-1.643833000	0.885017000	-3.421790000
8	-0.763735000	0.328568000	-4.088143000
7	-2.477579000	3.454630000	-3.850036000
6	-1.804212000	4.582208000	-4.180381000
6	-3.571091000	3.136463000	-4.582378000

6	-2.188957000	5.422124000	-5.214537000
1	-0.921223000	4.799796000	-3.594036000
6	-4.025206000	3.914750000	-5.639039000
1	-4.069614000	2.215900000	-4.304965000
6	-3.319539000	5.072134000	-5.945033000
1	-1.626888000	6.315058000	-5.455744000
1	-4.904467000	3.634695000	-6.204828000
7	-3.767507000	5.933128000	-7.051603000
8	-3.103535000	6.941757000	-7.282774000
8	-4.774285000	5.584083000	-7.663599000

η-NO

6	0.407591000	2.190583000	-0.306625000
6	-0.538463000	3.233266000	-0.146158000
6	-1.770514000	2.635861000	0.309266000
6	-1.583947000	1.244893000	0.415997000
6	-0.250716000	0.959667000	-0.024394000
1	1.420101000	2.298262000	-0.672980000
1	-0.350469000	4.290133000	-0.283905000
1	-2.689471000	3.170011000	0.517261000
1	-2.332695000	0.515680000	0.694176000
1	0.181338000	-0.026085000	-0.131438000

27	-1.220928000	1.883782000	-1.609143000
7	-0.289088000	1.335245000	-3.109097000
8	-1.281339000	0.371276000	-2.764262000
7	0.534855000	1.463147000	-3.916844000
7	-2.508543000	2.857133000	-2.695532000
6	-3.230761000	2.198789000	-3.633680000
6	-2.749898000	4.178326000	-2.524090000
6	-4.206249000	2.818708000	-4.403937000
1	-2.974822000	1.152435000	-3.752225000
6	-3.704149000	4.876727000	-3.248107000
1	-2.145717000	4.685055000	-1.783496000
6	-4.438529000	4.172742000	-4.196796000
1	-4.769540000	2.265857000	-5.144770000
1	-3.871067000	5.934222000	-3.089387000
7	-5.463743000	4.871809000	-4.988696000
8	-5.626406000	6.069927000	-4.764568000
8	-6.085910000	4.208244000	-5.814798000

[Ir(η^5 -Cp)(CNMe)(N₂O)]

κ -N

6	-0.093967000	2.589535000	0.067010000
6	-1.264010000	3.379523000	0.085722000

6	-2.388995000	2.467922000	0.091287000
6	-1.878770000	1.119902000	0.246581000
6	-0.473380000	1.197856000	0.175192000
1	0.921238000	2.958820000	0.003311000
1	-1.315596000	4.458107000	0.043001000
1	-3.428718000	2.755782000	0.173535000
1	-2.471981000	0.221389000	0.339929000
1	0.212728000	0.361382000	0.198115000
77	-1.447239000	2.080628000	-1.851968000
7	-0.227007000	1.659526000	-3.352526000
7	0.496203000	1.402765000	-4.191780000
8	1.273259000	1.136577000	-5.076791000
6	-2.849350000	2.331669000	-3.043497000
7	-3.780778000	2.419806000	-3.800721000
6	-4.445136000	3.421695000	-4.593940000
1	-4.439269000	3.118401000	-5.645745000
1	-5.489642000	3.505930000	-4.276693000
1	-3.962250000	4.402320000	-4.501405000

κ-O

6	-0.170134000	2.674510000	0.247111000
6	-1.405013000	3.343292000	0.265901000

6	-2.438325000	2.325138000	0.104756000
6	-1.792293000	1.016609000	0.187770000
6	-0.405650000	1.244397000	0.211741000
1	0.805000000	3.144435000	0.257614000
1	-1.563669000	4.410990000	0.317185000
1	-3.504036000	2.498854000	0.179188000
1	-2.292827000	0.058784000	0.184967000
1	0.364770000	0.484531000	0.202596000
7	-0.049844000	1.214483000	-4.313529000
8	0.155562000	2.053041000	-3.468634000
7	-0.206529000	0.444067000	-5.124039000
6	-2.718021000	2.388740000	-3.069201000
7	-3.536709000	2.430710000	-3.970355000
6	-4.117624000	3.555334000	-4.677458000
1	-3.939869000	3.442134000	-5.752227000
1	-5.201967000	3.559303000	-4.522085000
1	-3.704951000	4.516579000	-4.345296000
77	-1.441304000	2.195998000	-1.764455000

η-NN

6	-0.482830000	3.162765000	-0.975043000
6	-1.750093000	3.834716000	-0.863193000

6	-2.722394000	2.898248000	-0.340271000
6	-2.076025000	1.645601000	-0.229336000
6	-0.708111000	1.794251000	-0.634561000
1	0.453918000	3.605592000	-1.281289000
1	-1.915652000	4.893279000	-1.013710000
1	-3.756919000	3.112669000	-0.113385000
1	-2.545326000	0.714185000	0.059189000
1	0.032430000	1.005788000	-0.653633000
77	-1.985819000	2.384074000	-2.520998000
7	-2.497559000	0.619421000	-3.671103000
7	-1.378585000	1.003786000	-3.955532000
8	-0.399374000	0.720382000	-4.642395000
6	-2.714814000	3.473841000	-3.880393000
7	-3.185952000	4.154231000	-4.718043000
6	-3.590750000	4.967016000	-5.804217000
1	-2.722324000	5.271669000	-6.399063000
1	-4.283751000	4.413674000	-6.446214000
1	-4.098444000	5.863089000	-5.432472000

η-NO

6	-0.268540000	2.598996000	-0.406976000
6	-1.557163000	3.209339000	-0.209293000

6	-2.389147000	2.280290000	0.536102000
6	-1.666755000	1.085969000	0.667922000
6	-0.368130000	1.254735000	0.061866000
1	0.595939000	3.062674000	-0.861088000
1	-1.800666000	4.242453000	-0.418782000
1	-3.405887000	2.457923000	0.857768000
1	-2.040841000	0.161759000	1.089415000
1	0.418182000	0.512505000	0.038352000
77	-1.811614000	1.629589000	-1.734211000
8	-1.970753000	-0.205792000	-2.828568000
7	-0.024257000	0.383379000	-4.036680000
7	-0.901892000	0.478155000	-3.283138000
6	-2.913060000	2.473689000	-3.019818000
7	-3.605756000	3.003193000	-3.810506000
6	-4.350782000	3.634212000	-4.835080000
1	-5.019271000	2.909665000	-5.311465000
1	-4.954524000	4.445272000	-4.414743000
1	-3.678948000	4.049041000	-5.594648000

[Rh(η^5 -Cp)(CNMe)(N₂O)]

κ -N

6	-0.057319000	2.497305000	0.163245000
---	--------------	-------------	-------------

6	-1.167598000	3.353025000	0.176587000
6	-2.340077000	2.512863000	0.076871000
6	-1.929294000	1.132473000	0.218139000
6	-0.528898000	1.123804000	0.193221000
1	0.982878000	2.798465000	0.164513000
1	-1.154585000	4.434136000	0.175613000
1	-3.363551000	2.864497000	0.117318000
1	-2.583890000	0.272614000	0.252265000
1	0.106867000	0.247523000	0.217165000
45	-1.347633000	2.139013000	-1.852000000
7	-0.083328000	1.684445000	-3.369429000
7	0.640124000	1.420410000	-4.205509000
8	1.408336000	1.141453000	-5.089746000
6	-2.752064000	2.582496000	-3.046210000
7	-3.644780000	2.834091000	-3.777981000
6	-4.613088000	3.380498000	-4.654862000
1	-4.607213000	2.838668000	-5.606551000
1	-5.611539000	3.292231000	-4.213661000
1	-4.406009000	4.439527000	-4.850370000

κ-O

6	-0.039780000	2.244592000	0.255359000
---	--------------	-------------	-------------

6	-0.975289000	3.281949000	0.213005000
6	-2.282950000	2.660799000	0.073962000
6	-2.125473000	1.227582000	0.249815000
6	-0.749820000	0.976215000	0.268850000
1	1.037175000	2.354257000	0.290004000
1	-0.772077000	4.343901000	0.211713000
1	-3.227139000	3.190936000	0.091762000
1	-2.921735000	0.496673000	0.276679000
1	-0.279125000	0.001632000	0.312113000
45	-1.364916000	2.116689000	-1.795800000
7	0.388533000	2.013627000	-4.352281000
8	0.065412000	1.268250000	-3.465273000
7	0.707196000	2.688662000	-5.201488000
6	-2.566946000	2.788177000	-3.050444000
7	-3.289699000	3.275859000	-3.875431000
6	-4.459511000	3.053152000	-4.675606000
1	-4.864030000	2.044409000	-4.526479000
1	-5.228368000	3.787766000	-4.414613000
1	-4.214892000	3.186796000	-5.734292000

η-NN

6	-0.445608000	3.060338000	-0.896766000
---	--------------	-------------	--------------

6	-1.648773000	3.828269000	-0.879023000
6	-2.727854000	2.984283000	-0.391722000
6	-2.207755000	1.694928000	-0.229960000
6	-0.811832000	1.720962000	-0.595013000
1	0.538075000	3.413843000	-1.172639000
1	-1.720543000	4.894043000	-1.054785000
1	-3.748778000	3.297827000	-0.221488000
1	-2.763800000	0.811220000	0.055066000
1	-0.141437000	0.871456000	-0.563777000
45	-1.931108000	2.365935000	-2.543730000
7	-2.239187000	0.589795000	-3.763280000
7	-1.158902000	1.043160000	-4.009575000
8	-0.114076000	0.925367000	-4.638076000
6	-2.700302000	3.479013000	-3.908946000
7	-3.173059000	4.155988000	-4.743604000
6	-3.723952000	4.944112000	-5.784217000
1	-2.964976000	5.140709000	-6.548440000
1	-4.566727000	4.419026000	-6.244996000
1	-4.077780000	5.898001000	-5.380691000

η-NO

6	-0.265143000	2.589512000	-0.356299000
---	--------------	-------------	--------------

6	-1.564111000	3.189134000	-0.233718000
6	-2.418551000	2.278827000	0.518257000
6	-1.706852000	1.094952000	0.692552000
6	-0.390530000	1.260074000	0.104946000
1	0.607933000	3.039330000	-0.808592000
1	-1.805277000	4.218113000	-0.468999000
1	-3.439168000	2.473774000	0.818731000
1	-2.079882000	0.181717000	1.138726000
1	0.391241000	0.510898000	0.107529000
45	-1.839282000	1.596545000	-1.726017000
8	-1.787750000	-0.254077000	-2.922660000
7	0.108586000	0.494314000	-4.083019000
7	-0.800791000	0.402465000	-3.380691000
6	-2.921415000	2.474867000	-3.033962000
7	-3.595958000	3.018194000	-3.830324000
6	-4.392601000	3.648364000	-4.816667000
1	-5.238745000	3.006828000	-5.084387000
1	-4.777912000	4.598944000	-4.433858000
1	-3.797657000	3.844270000	-5.715118000

[Co(PMe₃)₂(N₂O)Cl]

κ-N

27	0.019077000	0.971266000	-0.103737000
17	-2.234571000	0.972509000	-0.098511000
15	-0.070658000	-1.265603000	-0.106052000
15	-0.068312000	3.208293000	-0.105789000
7	1.736872000	0.970407000	-0.163260000
7	2.878847000	0.969826000	0.072156000
8	4.059521000	0.969277000	-0.269910000
6	-0.940679000	3.932620000	-1.558950000
1	-1.941971000	3.498707000	-1.620066000
1	-1.014367000	5.023421000	-1.481563000
1	-0.396217000	3.676590000	-2.473589000
6	-0.946305000	3.941477000	1.339878000
1	-1.016926000	5.032050000	1.256672000
1	-1.949050000	3.510777000	1.398875000
1	-0.407007000	3.689402000	2.258771000
6	1.536143000	4.121589000	-0.102536000
1	2.115203000	3.857614000	0.788040000
1	2.126162000	3.847816000	-0.982707000
1	1.377157000	5.205623000	-0.108789000
6	-0.945198000	-1.989670000	-1.558000000
1	-1.020750000	-3.080267000	-1.479563000

1	-1.945775000	-1.554059000	-1.618786000
1	-0.400935000	-1.735438000	-2.473259000
6	-0.948103000	-1.997216000	1.340788000
1	-1.950305000	-1.565364000	1.400607000
1	-1.020049000	-3.087744000	1.258102000
1	-0.407606000	-1.745391000	2.259046000
6	1.532824000	-2.180598000	-0.103642000
1	2.122805000	-1.907386000	-0.984011000
1	2.112482000	-1.917286000	0.786743000
1	1.372702000	-3.264464000	-0.109901000

κ-O

NA

η-NN

27	2.844993000	0.291472000	0.084508000
17	1.103553000	-0.933013000	-0.630246000
15	1.444323000	1.250215000	1.599775000
15	4.129110000	-0.783969000	-1.455599000
7	4.312756000	0.788565000	1.043573000
8	5.043255000	0.378709000	1.959036000
7	4.143814000	1.716082000	0.276934000
6	4.163287000	-2.615444000	-1.276410000

1	3.138067000	-2.993057000	-1.309714000
1	4.755888000	-3.085591000	-2.069187000
1	4.597248000	-2.875636000	-0.305620000
6	3.594184000	-0.513464000	-3.196364000
1	4.219180000	-1.074258000	-3.900399000
1	2.552437000	-0.828783000	-3.299142000
1	3.659069000	0.553154000	-3.434369000
6	5.909184000	-0.311486000	-1.463441000
1	5.999849000	0.758172000	-1.673121000
1	6.338980000	-0.498082000	-0.474462000
1	6.470687000	-0.877925000	-2.214108000
6	0.042643000	2.174097000	0.843893000
1	-0.521334000	1.495178000	0.198742000
1	-0.621957000	2.591752000	1.608502000
1	0.435571000	2.990500000	0.229375000
6	0.631799000	0.061582000	2.746315000
1	-0.067172000	0.568166000	3.421244000
1	0.097084000	-0.688133000	2.157426000
1	1.396851000	-0.446571000	3.342069000
6	2.226177000	2.480136000	2.725774000
1	3.044158000	2.002485000	3.273849000

1	2.649665000	3.296543000	2.133750000
1	1.501908000	2.886669000	3.439784000
η-NO			
27	2.863525000	0.197721000	0.163472000
17	1.171095000	-1.180304000	-0.405560000
15	1.457054000	1.244726000	1.609286000
15	4.137723000	-0.786958000	-1.440576000
7	4.278444000	0.414516000	1.262574000
7	4.980243000	0.058812000	2.113347000
8	4.150492000	1.559659000	0.387455000
6	4.246652000	-2.621672000	-1.366420000
1	3.235923000	-3.036970000	-1.386161000
1	4.829838000	-3.024032000	-2.202188000
1	4.722417000	-2.916315000	-0.425469000
6	3.510125000	-0.438588000	-3.136731000
1	4.119373000	-0.935895000	-3.899786000
1	2.476188000	-0.786805000	-3.211283000
1	3.528199000	0.641146000	-3.316924000
6	5.890769000	-0.233350000	-1.494726000
1	5.918469000	0.856685000	-1.565537000
1	6.391481000	-0.519933000	-0.565097000

1	6.425117000	-0.677172000	-2.341599000
6	0.097785000	2.131865000	0.738996000
1	-0.455952000	1.417308000	0.123735000
1	-0.586404000	2.611447000	1.447981000
1	0.524419000	2.898594000	0.084202000
6	0.590217000	0.157097000	2.813062000
1	-0.108491000	0.724638000	3.437708000
1	0.050166000	-0.618243000	2.263790000
1	1.330490000	-0.328482000	3.457087000
6	2.235057000	2.543332000	2.653700000
1	2.979032000	2.083999000	3.311398000
1	2.755333000	3.255480000	2.008693000
1	1.490478000	3.065761000	3.263937000

[Co(PH₃)₂(N₂O)Cl]

κ-N			
27	0.054392000	0.971255000	-0.107864000
17	-2.185955000	0.971249000	-0.107694000
15	-0.094164000	-1.224692000	-0.107973000
1	1.062679000	-2.046727000	-0.108369000
1	-0.772403000	-1.842765000	0.965276000
1	-0.773051000	-1.842594000	-1.180905000

15	-0.094173000	3.167201000	-0.107899000
1	1.062665000	3.989241000	-0.106543000
1	-0.771455000	3.785235000	-1.181788000
1	-0.774023000	3.785134000	0.964392000
7	1.801590000	0.971259000	-0.108253000
7	2.943067000	0.971253000	-0.107053000
8	4.151792000	0.971211000	-0.107206000

κ-O

NA

η-NN

27	0.272378000	0.967190000	-0.168560000
17	-1.846231000	0.942413000	0.507523000
15	0.188388000	-1.290168000	-0.117982000
1	1.354566000	-2.038568000	-0.399406000
1	-0.218641000	-1.921027000	1.074572000
1	-0.699844000	-1.907329000	-1.022696000
15	0.117574000	3.221240000	-0.175862000
1	1.259850000	3.998385000	-0.476710000
1	-0.789099000	3.786801000	-1.095889000
1	-0.309721000	3.869421000	1.000199000
7	2.101323000	0.997218000	-0.064327000

7	1.870917000	0.978191000	-1.255372000
8	2.986465000	1.022493000	0.792628000

η -NO

27	-1.627068000	-1.829945000	-1.258606000
17	-2.870952000	-0.064626000	-1.801707000
15	-2.567676000	-2.951885000	-2.975633000
15	-0.941845000	-0.711750000	0.577346000
1	-3.886615000	-3.403328000	-2.756830000
1	-2.744252000	-2.310402000	-4.217431000
1	-1.985439000	-4.167087000	-3.399242000
1	-0.590226000	0.648788000	0.477106000
1	-1.881746000	-0.632904000	1.626918000
1	0.162778000	-1.205430000	1.306233000
7	0.001406000	-2.609185000	-1.512662000
8	-0.774163000	-3.406249000	-0.655960000
7	1.052577000	-2.597645000	-2.000508000

[Co(CNMe)₂(N₂O)Cl]

27	0.212497000	0.970937000	0.472719000
17	-2.022510000	0.971696000	0.511015000
6	0.095686000	-0.850292000	0.598652000

6	0.096921000	2.792019000	0.601618000
7	1.983040000	0.970486000	0.290753000
7	3.081621000	0.970120000	-0.008765000
8	4.269261000	0.969475000	-0.222523000
7	-0.055529000	3.942556000	0.756159000
7	-0.057573000	-2.000973000	0.751316000
6	-0.391060000	5.305679000	0.939544000
1	-1.475362000	5.403731000	1.050112000
1	0.098954000	5.694668000	1.837792000
1	-0.063688000	5.891487000	0.074664000
6	-0.394018000	-3.364168000	0.932473000
1	-1.478544000	-3.461865000	1.041146000
1	-0.065484000	-3.949068000	0.067421000
1	0.094339000	-3.754473000	1.831055000

κ-Ω

NA

η-NN

27	2.828345000	0.870235000	-0.314997000
17	1.056469000	-0.126364000	-1.209740000
6	1.659412000	1.658631000	0.900420000
6	3.855025000	0.002815000	-1.603041000

7	4.303727000	1.116604000	0.816051000
8	4.869702000	0.504297000	1.717015000
7	4.272439000	2.107393000	0.133537000
7	0.886194000	2.133562000	1.631264000
7	4.451301000	-0.555046000	-2.434184000
6	-0.104766000	2.670313000	2.491327000
1	-1.090631000	2.318151000	2.174565000
1	0.084627000	2.344841000	3.518332000
1	-0.077395000	3.762966000	2.448792000
6	5.116667000	-1.267428000	-3.463561000
1	5.815174000	-1.984167000	-3.022073000
1	4.380427000	-1.803554000	-4.069208000
1	5.670110000	-0.568543000	-4.097366000

η-NO

27	2.827851000	0.731959000	-0.223431000
17	1.105981000	-0.494328000	-0.923495000
6	1.656656000	1.580322000	0.949320000
6	3.852021000	-0.073707000	-1.553356000
7	4.224533000	0.679465000	1.036554000
7	4.729196000	0.069482000	1.882652000
8	4.301765000	1.826990000	0.345936000

7	0.880382000	2.109548000	1.638206000
7	4.442940000	-0.574937000	-2.423322000
6	-0.112414000	2.702704000	2.458351000
1	-1.093595000	2.300508000	2.190476000
1	0.097422000	2.478762000	3.508252000
1	-0.110980000	3.786944000	2.313972000
6	5.106037000	-1.230255000	-3.491487000
1	5.882852000	-1.888664000	-3.092068000
1	4.383485000	-1.823860000	-4.058756000
1	5.566242000	-0.488610000	-4.150719000

[Co(CO)₂(N₂O)Cl]

κ-N			
27	0.014461000	0.971202000	-0.004102000
17	-2.092752000	0.973105000	-0.601396000
6	-0.159900000	-0.751940000	0.477618000
6	-0.157735000	2.693635000	0.480875000
7	1.841228000	0.970151000	-0.016237000
7	2.960942000	0.970270000	-0.199156000
8	4.140262000	0.970566000	-0.373991000
8	-0.350198000	3.715196000	0.962109000
8	-0.353416000	-1.774233000	0.956706000

κ -O

27	0.085324000	0.762574000	0.238954000
17	-1.882163000	0.965326000	1.160963000
6	-0.387900000	-0.783396000	-0.528139000
6	0.127670000	2.491025000	-0.175848000
8	2.041523000	0.365789000	0.160762000
7	2.944723000	1.165036000	0.041539000
7	3.819063000	1.868803000	-0.058166000
8	-0.804765000	-1.629251000	-1.183509000
8	-0.003264000	3.531848000	-0.649374000

 η -NN

27	2.804875000	0.524838000	-0.106244000
17	1.087863000	-0.609659000	-0.858982000
6	1.629126000	1.272615000	1.092735000
6	3.794867000	-0.367865000	-1.371687000
7	4.339589000	0.949524000	0.959955000
8	4.949110000	0.402794000	1.858622000
7	4.186006000	1.871616000	0.211770000
8	4.358889000	-0.943184000	-2.174150000
8	0.859452000	1.708535000	1.807489000

 η -NO

27	2.814518000	0.425073000	-0.030937000
17	1.191058000	-0.971576000	-0.527032000
6	1.619484000	1.259512000	1.084091000
6	3.776963000	-0.375791000	-1.372535000
7	4.324310000	0.595257000	1.181628000
7	4.881216000	0.051934000	2.032152000
8	4.240617000	1.650109000	0.406224000
8	0.823911000	1.775332000	1.712578000
8	4.306405000	-0.864273000	-2.252906000

[Co(C₄N₂H₅)₂(NC₄H₈)(N₂O)]

κ-N			
6	2.427590000	-2.186447000	0.855929000
6	1.407993000	-0.077524000	1.690586000
1	1.818689000	0.938773000	1.692726000
27	0.878962000	-2.930446000	-0.051012000
7	2.520395000	-1.010838000	1.535999000
6	-2.368888000	-2.738607000	0.604097000
1	-2.230542000	-2.897497000	1.681123000
7	-1.907652000	-3.918980000	-0.119594000
1	-3.438560000	-2.622811000	0.399872000
6	-0.613172000	-4.095923000	-0.513561000

7	-0.634840000	-5.338756000	-1.099966000
6	-2.706902000	-5.003666000	-0.431316000
6	-1.902486000	-5.898206000	-1.055805000
1	-2.116433000	-6.867751000	-1.478253000
6	0.486084000	-6.029307000	-1.733489000
1	0.713734000	-5.596016000	-2.709285000
1	1.372817000	-5.951392000	-1.106983000
1	0.211424000	-7.080326000	-1.854098000
1	-3.758021000	-5.038265000	-0.189628000
7	3.642574000	-2.780271000	1.081467000
6	4.453126000	-1.987243000	1.878950000
6	3.741930000	-0.870772000	2.172138000
1	5.453455000	-2.282820000	2.155284000
1	3.992182000	-0.001718000	2.761024000
1	0.927584000	-0.263641000	2.659807000
6	4.069563000	-4.083518000	0.579625000
1	3.221390000	-4.766572000	0.591779000
1	4.441953000	-4.008836000	-0.444384000
1	4.857259000	-4.464297000	1.235146000
7	-0.211542000	-1.560046000	0.552555000
6	0.382314000	-0.230005000	0.578317000

6	-1.610275000	-1.495751000	0.174013000
1	0.847044000	0.057655000	-0.380811000
1	-0.389895000	0.529383000	0.782514000
1	-1.773889000	-1.340683000	-0.910799000
1	-2.101637000	-0.648312000	0.682055000
7	1.875249000	-3.486669000	-1.365454000
7	2.546382000	-3.500937000	-2.359300000
8	2.697345000	-2.745187000	-3.353153000

κ-O

NA

η-NN

6	2.442198000	-2.152986000	0.947665000
6	1.208086000	-0.386202000	2.110282000
1	1.465759000	0.602006000	2.505490000
27	0.831313000	-3.121469000	0.400599000
7	2.452686000	-1.057277000	1.749232000
6	-2.386028000	-2.680852000	0.511037000
1	-2.237676000	-2.827335000	1.586628000
7	-1.956612000	-3.876149000	-0.207843000
1	-3.453922000	-2.540162000	0.313612000
6	-0.638250000	-4.123282000	-0.418282000

7	-0.643130000	-5.215311000	-1.240664000
6	-2.771060000	-4.786441000	-0.855783000
6	-1.937887000	-5.637616000	-1.507050000
1	-2.145972000	-6.488054000	-2.137666000
6	0.528355000	-5.857900000	-1.831050000
1	1.288801000	-5.102724000	-2.020040000
1	0.947302000	-6.606432000	-1.156313000
1	0.226288000	-6.323444000	-2.773503000
1	-3.848050000	-4.744757000	-0.803473000
7	3.767401000	-2.392060000	0.712368000
6	4.568826000	-1.473562000	1.375592000
6	3.735604000	-0.624631000	2.029770000
1	5.645651000	-1.500538000	1.312813000
1	3.937881000	0.234540000	2.650487000
1	0.695589000	-0.960553000	2.889893000
6	4.317269000	-3.430184000	-0.155788000
1	4.440333000	-4.369904000	0.385406000
1	3.631770000	-3.598734000	-0.984027000
1	5.281620000	-3.083333000	-0.537769000
7	-0.173894000	-1.541220000	0.388775000
6	0.317082000	-0.258995000	0.876601000

6	-1.583790000	-1.474957000	0.026918000
1	0.882221000	0.309133000	0.113745000
1	-0.521923000	0.389069000	1.174611000
1	-1.749024000	-1.379705000	-1.062839000
1	-2.059461000	-0.592176000	0.481991000
7	1.894764000	-4.850403000	0.496726000
7	1.253931000	-4.527360000	1.474824000
8	0.988480000	-4.913008000	2.628742000

η -NO

6	2.446707000	-2.137544000	0.907789000
6	1.220411000	-0.390280000	2.113085000
1	1.481077000	0.593756000	2.516597000
27	0.835663000	-3.093116000	0.349857000
7	2.461654000	-1.049415000	1.718235000
6	-2.387785000	-2.693095000	0.507070000
1	-2.237563000	-2.865707000	1.578686000
7	-1.948167000	-3.868429000	-0.238960000
1	-3.457560000	-2.559514000	0.315058000
6	-0.629787000	-4.105348000	-0.455881000
7	-0.626019000	-5.208460000	-1.261161000
6	-2.756560000	-4.796451000	-0.869329000

6	-1.917630000	-5.646370000	-1.515403000
1	-2.120509000	-6.506396000	-2.134586000
6	0.552308000	-5.868275000	-1.819215000
1	1.295271000	-5.115896000	-2.078164000
1	0.994479000	-6.550574000	-1.092885000
1	0.246460000	-6.409032000	-2.719379000
1	-3.833599000	-4.765301000	-0.811530000
7	3.769225000	-2.394773000	0.684397000
6	4.575561000	-1.490948000	1.360917000
6	3.747520000	-0.636421000	2.014970000
1	5.652470000	-1.530717000	1.307281000
1	3.955109000	0.215519000	2.643798000
1	0.730721000	-0.978194000	2.897467000
6	4.313302000	-3.458192000	-0.157155000
1	4.347584000	-4.402675000	0.386334000
1	3.673902000	-3.588002000	-1.028657000
1	5.316073000	-3.161041000	-0.477130000
7	-0.190810000	-1.520440000	0.396958000
6	0.297873000	-0.247856000	0.904490000
6	-1.603199000	-1.463815000	0.053723000
1	0.838296000	0.347919000	0.143478000

1	-0.540562000	0.383888000	1.239195000
1	-1.784609000	-1.336435000	-1.031099000
1	-2.086474000	-0.601982000	0.541256000
8	1.878550000	-4.762045000	0.405380000
7	1.156855000	-4.425324000	1.547716000
7	0.976297000	-4.877987000	2.608514000

[Co(C₆N₂H₁₁)₂(NC₄H₈)(N₂O)]

κ-N			
6	2.574284000	-2.185951000	0.845504000
6	1.176236000	-0.264735000	1.623859000
1	1.440048000	0.791724000	1.750161000
27	0.985220000	-3.303423000	0.321292000
7	2.432108000	-1.014024000	1.529075000
6	-1.852887000	-3.273948000	1.440458000
1	-1.198601000	-3.550571000	2.277381000
7	-1.704027000	-4.293729000	0.405655000
1	-2.894542000	-3.281692000	1.777023000
6	-0.522765000	-4.430673000	-0.236079000
7	-0.763150000	-5.387880000	-1.186452000
6	-2.674421000	-5.132401000	-0.103107000
6	-2.082597000	-5.830478000	-1.104541000

1	-2.489361000	-6.588083000	-1.750771000
6	0.249803000	-5.917668000	-2.149938000
1	-3.679541000	-5.166264000	0.287597000
7	3.927527000	-2.435333000	0.919727000
6	4.567790000	-1.459190000	1.675860000
6	3.626677000	-0.563691000	2.053672000
1	5.626019000	-1.467187000	1.869457000
1	3.701993000	0.346504000	2.628447000
1	0.618730000	-0.592235000	2.508904000
6	4.697578000	-3.581148000	0.330248000
7	-0.161332000	-1.846629000	0.262260000
6	0.358323000	-0.492971000	0.352409000
6	-1.492560000	-1.907845000	0.863174000
1	0.998258000	-0.262035000	-0.509265000
1	-0.464102000	0.238148000	0.333635000
1	-2.269724000	-1.617533000	0.130877000
1	-1.586003000	-1.197140000	1.703934000
6	1.259804000	-6.777940000	-1.368758000
1	0.759129000	-7.654156000	-0.941945000
1	2.055097000	-7.128332000	-2.036554000
1	1.704777000	-6.217322000	-0.544413000

6	-0.436740000	-6.787127000	-3.216098000
1	0.318200000	-7.105062000	-3.941973000
1	-0.879994000	-7.693317000	-2.789886000
1	-1.210221000	-6.233012000	-3.759004000
6	0.919317000	-4.723424000	-2.851054000
1	1.329295000	-4.020954000	-2.122102000
1	1.730147000	-5.073097000	-3.499182000
1	0.191110000	-4.180441000	-3.463548000
6	4.823771000	-4.694199000	1.386014000
1	3.837901000	-5.056105000	1.683927000
1	5.405359000	-5.529525000	0.978982000
1	5.343044000	-4.322643000	2.276993000
6	3.983444000	-4.087133000	-0.925115000
1	4.537103000	-4.936905000	-1.338302000
1	2.975798000	-4.418159000	-0.683087000
1	3.924772000	-3.304337000	-1.689361000
6	6.099926000	-3.090527000	-0.085954000
1	6.742215000	-2.861074000	0.770194000
1	6.594845000	-3.888335000	-0.648662000
1	6.041166000	-2.206173000	-0.729787000
7	1.494004000	-4.550834000	1.457939000

7	1.185000000	-5.174689000	2.451238000
8	0.322728000	-4.890491000	3.344451000

κ-O

NA

η-NN

6	2.489315000	-2.127448000	0.685239000
6	1.048299000	-0.192023000	1.373517000
1	1.326709000	0.858643000	1.517597000
27	0.860017000	-3.264547000	0.179654000
7	2.296388000	-0.963568000	1.376536000
6	-2.007058000	-3.037373000	1.186486000
1	-1.366958000	-3.245967000	2.048856000
7	-1.874176000	-4.139087000	0.238307000
1	-3.050217000	-2.995996000	1.515758000
6	-0.692571000	-4.337289000	-0.392187000
7	-0.971867000	-5.288305000	-1.338817000
6	-2.880221000	-4.934364000	-0.272508000
6	-2.310714000	-5.668866000	-1.259719000
1	-2.746104000	-6.414960000	-1.900711000
6	-0.009499000	-5.868817000	-2.330789000
1	-3.890182000	-4.915618000	0.106857000

7	3.849914000	-2.338298000	0.797565000
6	4.433172000	-1.360289000	1.596350000
6	3.457988000	-0.494104000	1.951969000
1	5.482271000	-1.345188000	1.832491000
1	3.489879000	0.408062000	2.543207000
1	0.417792000	-0.501397000	2.214803000
6	4.722017000	-3.373063000	0.131732000
7	-0.243750000	-1.748884000	-0.036979000
6	0.322267000	-0.415482000	0.044245000
6	-1.601545000	-1.732676000	0.503232000
1	1.039435000	-0.240746000	-0.768363000
1	-0.457436000	0.355386000	-0.061491000
1	-2.354038000	-1.498742000	-0.274959000
1	-1.716580000	-0.950676000	1.276297000
7	1.810400000	-4.986982000	0.753360000
7	1.169879000	-4.365419000	1.575435000
8	0.840323000	-4.410032000	2.780425000
6	0.813630000	-6.971930000	-1.642378000
1	0.153299000	-7.771365000	-1.286126000
1	1.520841000	-7.409426000	-2.356916000
1	1.366167000	-6.574219000	-0.788409000

6	-0.784943000	-6.470438000	-3.516928000
1	-0.063432000	-6.774451000	-4.281731000
1	-1.352057000	-7.364809000	-3.237699000
1	-1.467898000	-5.742355000	-3.968048000
6	0.881307000	-4.736550000	-2.859722000
1	1.375182000	-4.221754000	-2.035144000
1	1.647516000	-5.146909000	-3.526026000
1	0.288749000	-4.001100000	-3.414720000
6	4.959965000	-4.539822000	1.106220000
1	4.020947000	-5.049179000	1.332707000
1	5.652983000	-5.260493000	0.656314000
1	5.407016000	-4.178525000	2.039785000
6	4.071147000	-3.849380000	-1.169952000
1	4.737603000	-4.570078000	-1.656394000
1	3.122248000	-4.336199000	-0.968135000
1	3.910504000	-3.010614000	-1.856843000
6	6.075390000	-2.723937000	-0.236292000
1	6.700723000	-2.508958000	0.635930000
1	6.635285000	-3.427131000	-0.860637000
1	5.936458000	-1.799436000	-0.807132000

η-NO

6	2.505081000	-2.103868000	0.694039000
6	1.049349000	-0.164232000	1.344387000
1	1.314814000	0.892420000	1.465999000
27	0.889705000	-3.254279000	0.189420000
7	2.306822000	-0.919642000	1.347718000
6	-1.992954000	-3.074294000	1.188282000
1	-1.370674000	-3.296218000	2.060084000
7	-1.837209000	-4.155729000	0.221242000
1	-3.041165000	-3.046368000	1.501670000
6	-0.645133000	-4.346566000	-0.392547000
7	-0.903638000	-5.300533000	-1.341494000
6	-2.830090000	-4.957988000	-0.304688000
6	-2.241294000	-5.688313000	-1.283461000
1	-2.663036000	-6.435448000	-1.932120000
6	0.083284000	-5.881666000	-2.309122000
1	-3.845754000	-4.945334000	0.059348000
7	3.866389000	-2.305246000	0.811071000
6	4.446609000	-1.295578000	1.571102000
6	3.467861000	-0.423280000	1.901047000
1	5.496374000	-1.264470000	1.801342000
1	3.496924000	0.499847000	2.459093000

1	0.434006000	-0.465496000	2.199928000
6	4.733950000	-3.384652000	0.212353000
7	-0.242737000	-1.761110000	-0.025088000
6	0.310132000	-0.421208000	0.029291000
6	-1.585756000	-1.748539000	0.549970000
1	1.014889000	-0.251867000	-0.795426000
1	-0.482154000	0.336172000	-0.080191000
1	-2.346252000	-1.471040000	-0.205382000
1	-1.678726000	-0.997973000	1.358117000
8	1.854522000	-4.911486000	0.692941000
7	1.115319000	-4.263501000	1.676733000
7	0.878343000	-4.431088000	2.809837000
6	0.935941000	-6.931648000	-1.575416000
1	0.302828000	-7.757035000	-1.228553000
1	1.688397000	-7.344879000	-2.257440000
1	1.431357000	-6.488835000	-0.708201000
6	-0.663096000	-6.548732000	-3.478253000
1	0.075373000	-6.867151000	-4.220634000
1	-1.212484000	-7.443908000	-3.168056000
1	-1.356658000	-5.856820000	-3.968744000
6	0.938760000	-4.738373000	-2.873999000

1	1.424814000	-4.184381000	-2.070666000
1	1.712217000	-5.144779000	-3.534335000
1	0.321477000	-4.036844000	-3.445861000
6	4.882106000	-4.531184000	1.227086000
1	3.908118000	-4.982399000	1.424238000
1	5.557101000	-5.295786000	0.824446000
1	5.311222000	-4.161843000	2.165999000
6	4.126676000	-3.871183000	-1.107147000
1	4.796187000	-4.614633000	-1.553471000
1	3.161254000	-4.335568000	-0.933112000
1	4.012554000	-3.042207000	-1.815247000
6	6.125030000	-2.793133000	-0.106776000
1	6.716031000	-2.577703000	0.789091000
1	6.687095000	-3.533906000	-0.683708000
1	6.049832000	-1.881064000	-0.709045000

[N₂]

7	-0.319166000	0.170807000	0.000000000
7	0.786135000	0.170807000	0.000000000

[Ni(CNMe)₂(N₂)]

κ-N

1	-3.067619000	-2.796576000	2.352720000
---	--------------	--------------	-------------

1	-1.777659000	-2.374548000	3.506156000
1	-3.286025000	-1.433661000	3.477587000
6	-1.309765000	-0.581469000	-4.328026000
1	-0.335717000	-0.778595000	-4.789807000
1	-1.904278000	-1.502249000	-4.355695000
1	-1.824217000	0.185314000	-4.918122000
7	0.530281000	1.774414000	0.278894000
7	1.395373000	2.435666000	0.556655000
28	-0.756409000	0.632125000	-0.164808000
6	-1.558949000	-0.333741000	1.104676000
6	-1.039478000	0.242196000	-1.884225000
7	-1.993123000	-1.082405000	1.908697000
7	-1.140227000	-0.136802000	-2.998525000
6	-2.558136000	-1.965582000	2.854638000

η-NN

1	-5.276784000	2.124601000	-3.573808000
1	-5.555319000	1.006984000	-2.217517000
1	-5.329496000	2.745654000	-1.905393000
6	0.618279000	-1.919686000	-4.816958000
1	0.086812000	-2.849330000	-4.585524000
1	0.418246000	-1.653333000	-5.860652000

1	1.693897000	-2.089918000	-4.697170000
7	-0.052391000	2.550887000	-0.842384000
7	0.858268000	1.932556000	-1.210387000
28	-0.697648000	1.290499000	-2.149369000
6	-2.456954000	1.561183000	-2.288977000
6	-0.135616000	-0.015777000	-3.229381000
7	-3.620058000	1.700450000	-2.415191000
7	0.185998000	-0.883867000	-3.958125000
6	-5.013439000	1.905285000	-2.533224000

[Ni(CO)₂(N₂)]

κ-N			
28	0.128013000	0.727116000	0.049862000
6	-0.560578000	2.290369000	0.460522000
8	-0.924584000	3.360950000	0.676862000
6	1.876279000	0.588940000	-0.055120000
8	3.022801000	0.604796000	-0.158484000
7	-0.938543000	-0.650178000	-0.446600000
7	-1.594481000	-1.485301000	-0.791266000

η-NN			
28	-0.379237000	0.117093000	0.297372000
6	-1.035240000	1.745710000	0.419761000

8	-1.361110000	2.844725000	0.488127000
6	1.363273000	0.120211000	0.056606000
8	2.495874000	0.234273000	-0.095683000
7	-1.893658000	-1.118783000	0.488345000
7	-0.952246000	-1.760170000	0.345718000

[Ni(C₁₀N₄H₁₆)(N₂)]

κ-N

6	-2.349724000	-1.693255000	-1.220263000
6	-0.659911000	-0.495157000	-0.226771000
1	-2.837568000	-2.247710000	-2.007734000
6	4.780324000	-1.150869000	-0.005701000
6	4.371247000	-1.613214000	1.200613000
6	2.619656000	-0.431546000	0.298661000
7	3.715906000	-0.432290000	-0.537355000
1	5.719588000	-1.267479000	-0.525078000
1	4.888833000	-2.200323000	1.944207000
7	3.063418000	-1.175479000	1.370058000
6	2.245918000	-1.428527000	2.539185000
6	3.735479000	0.247774000	-1.817435000
1	3.596306000	-0.453437000	-2.650132000
1	2.917425000	0.971381000	-1.816881000

1	4.685200000	0.775516000	-1.952853000
7	-1.753241000	-0.616236000	0.604239000
7	-1.066492000	-1.176948000	-1.352846000
6	-2.780064000	-1.346029000	0.016884000
1	-3.711176000	-1.550253000	0.523734000
6	-1.805723000	-0.037602000	1.932359000
1	-2.779915000	0.431689000	2.104312000
1	-1.632480000	-0.792760000	2.709637000
1	-1.023783000	0.722712000	1.989806000
6	-0.238896000	-1.297698000	-2.535932000
28	0.957542000	0.412245000	0.069589000
1	0.782247000	-1.052341000	-2.234721000
1	-0.272816000	-2.321374000	-2.924440000
1	-0.557559000	-0.603970000	-3.323336000
1	2.528680000	-0.779640000	3.376931000
1	1.212432000	-1.216330000	2.255389000
1	2.335468000	-2.475116000	2.850009000
7	0.912707000	2.160590000	0.136705000
7	0.883482000	3.292148000	0.180317000

η-NN

6	-2.445756000	-0.714753000	-1.284826000
---	--------------	--------------	--------------

6	-0.923526000	0.816244000	-0.505332000
1	-2.817879000	-1.499368000	-1.926099000
6	4.301892000	0.079805000	-0.284518000
6	3.741272000	-0.547631000	0.778172000
6	2.163462000	0.886735000	-0.071449000
7	3.338123000	0.942778000	-0.787645000
1	5.284138000	-0.013786000	-0.722523000
1	4.141425000	-1.287692000	1.454566000
7	2.446528000	-0.057329000	0.889257000
6	1.508250000	-0.452200000	1.924805000
6	3.548453000	1.827819000	-1.920815000
1	3.692236000	1.254654000	-2.844444000
1	2.663484000	2.457000000	-2.010457000
1	4.424292000	2.463224000	-1.753028000
7	-2.098045000	0.858230000	0.212021000
7	-1.171294000	-0.179892000	-1.422192000
6	-3.028419000	-0.062991000	-0.248972000
1	-4.005381000	-0.175266000	0.196408000
6	-2.340641000	1.784528000	1.305043000
1	-3.237703000	2.381271000	1.109638000
1	-2.465635000	1.247945000	2.253043000

1	-1.478093000	2.447351000	1.366583000
6	-0.220349000	-0.583559000	-2.442653000
28	0.599531000	1.900046000	-0.313577000
1	0.786821000	-0.493968000	-2.031957000
1	-0.410433000	-1.622719000	-2.727659000
1	-0.298417000	0.050840000	-3.333042000
1	1.551347000	0.229432000	2.782027000
1	0.500539000	-0.430383000	1.506278000
1	1.744298000	-1.465588000	2.263126000
7	-0.023317000	3.684787000	-0.403656000
7	1.150760000	3.710574000	-0.307841000

[Ni(PH₃)₂(N₂)]

κ-N			
28	-0.637634000	0.122652000	-0.004112000
15	-1.743755000	1.926980000	-0.043699000
15	-1.784025000	-1.649505000	0.157007000
1	-1.217690000	-2.955935000	0.202127000
1	-2.640629000	-1.851078000	1.273217000
1	-2.748982000	-1.970630000	-0.835971000
1	-2.702610000	2.156470000	-1.067519000
1	-1.148201000	3.217207000	-0.144510000

1	-2.594136000	2.271479000	1.041889000
7	1.101292000	0.097725000	-0.099778000
7	2.225683000	0.081568000	-0.162318000

η -NN

28	-1.016684000	0.339390000	0.049245000
15	-2.182667000	2.117417000	-0.092399000
15	-2.231995000	-1.408928000	0.133844000
1	-1.634605000	-2.693479000	0.230652000
1	-3.164921000	-1.607617000	1.186292000
1	-3.116029000	-1.744096000	-0.926166000
1	-3.057726000	2.339285000	-1.188979000
1	-1.549502000	3.386701000	-0.158144000
1	-3.108871000	2.474280000	0.923595000
7	0.784000000	0.899068000	0.055049000
7	0.767815000	-0.264785000	0.129633000

[Ni(PMe₃)₂(N₂)]

		κ-N	
28	2.444886000	0.038579000	-0.258946000
15	1.357183000	1.876515000	-0.139250000
15	1.324090000	-1.754729000	0.063738000
6	0.852717000	2.406297000	1.565957000

1	0.280297000	3.341945000	1.553622000
1	0.246482000	1.623746000	2.033215000
1	1.747961000	2.550306000	2.179770000
6	2.155705000	3.437416000	-0.752968000
1	2.401963000	3.323694000	-1.814015000
1	1.511726000	4.318128000	-0.632202000
1	3.091845000	3.605436000	-0.210474000
6	-0.255568000	1.978403000	-1.054242000
1	-0.935975000	1.199478000	-0.694677000
1	-0.740741000	2.955634000	-0.938271000
1	-0.076302000	1.801278000	-2.120225000
6	-0.008328000	-1.744598000	1.358384000
1	-0.750759000	-0.975622000	1.119800000
1	-0.516547000	-2.713105000	1.446663000
1	0.435208000	-1.492977000	2.327969000
6	2.233792000	-3.288619000	0.581753000
1	1.569588000	-4.154468000	0.699715000
1	3.000703000	-3.527500000	-0.162177000
1	2.740642000	-3.101755000	1.534492000
6	0.392240000	-2.383005000	-1.412181000
1	1.097132000	-2.606049000	-2.219622000

1	-0.178288000	-3.290854000	-1.180915000
1	-0.296949000	-1.611519000	-1.770205000
7	4.145121000	-0.004094000	-0.621168000
7	5.248959000	-0.029848000	-0.861752000

$\eta\text{-NN}$

28	5.471816000	0.244961000	-0.158130000
15	4.271011000	2.023985000	-0.075999000
15	4.298526000	-1.533653000	0.116041000
6	3.765329000	2.604604000	1.609896000
1	3.149677000	3.511312000	1.568774000
1	3.204096000	1.816405000	2.122171000
1	4.663618000	2.814656000	2.199748000
6	5.035084000	3.560422000	-0.769759000
1	5.252836000	3.405591000	-1.831557000
1	4.384341000	4.437218000	-0.664614000
1	5.985109000	3.754695000	-0.262683000
6	2.653061000	2.027708000	-0.982960000
1	2.003290000	1.233490000	-0.599996000
1	2.127540000	2.986222000	-0.893245000
1	2.838904000	1.827754000	-2.043746000
6	2.974736000	-1.518545000	1.415131000

1	2.241985000	-0.733861000	1.198362000
1	2.450464000	-2.479224000	1.487318000
1	3.430495000	-1.294174000	2.385586000
6	5.226859000	-3.054410000	0.617067000
1	4.577160000	-3.934129000	0.702195000
1	6.013644000	-3.257596000	-0.115792000
1	5.711975000	-2.877263000	1.582556000
6	3.371193000	-2.148518000	-1.365985000
1	4.084358000	-2.372369000	-2.166123000
1	2.790977000	-3.052919000	-1.146363000
1	2.692752000	-1.369786000	-1.729243000
7	7.238757000	0.828274000	-0.461905000
7	7.249482000	-0.341475000	-0.385486000

[Co(η^5 -Cp*)](CNMe)(N₂)]

κ-N			
6	-0.072529000	2.319197000	-0.159871000
6	-1.060180000	3.333187000	-0.090989000
6	-2.348106000	2.672340000	-0.099162000
6	-2.144912000	1.246063000	-0.057334000
6	-0.743893000	1.028639000	-0.152500000
27	-1.362999000	2.146636000	-1.808500000

7	-0.278611000	1.481449000	-3.033450000
7	0.454184000	1.044645000	-3.766151000
6	-2.489583000	2.892168000	-2.940718000
7	-3.201279000	3.500302000	-3.679481000
6	-4.345984000	3.563899000	-4.529790000
1	-4.901837000	2.617635000	-4.524985000
1	-5.010983000	4.368132000	-4.198494000
1	-4.032754000	3.785298000	-5.555287000
6	-3.673392000	3.368175000	0.041409000
1	-4.489937000	2.770090000	-0.374975000
1	-3.905025000	3.552765000	1.099320000
1	-3.678638000	4.335050000	-0.471018000
6	-0.835693000	4.813978000	0.021869000
1	-1.603774000	5.381313000	-0.513155000
1	-0.859176000	5.141046000	1.071702000
1	0.133849000	5.108829000	-0.391395000
6	1.418057000	2.509639000	-0.158709000
1	1.695529000	3.521666000	-0.468737000
1	1.836416000	2.344132000	0.844037000
1	1.918723000	1.811043000	-0.837602000
6	-0.042158000	-0.300484000	-0.138462000

1	0.830190000	-0.307942000	-0.800405000
1	0.314621000	-0.545994000	0.871473000
1	-0.704446000	-1.110111000	-0.459970000
6	-3.211454000	0.201517000	0.110381000
1	-2.921764000	-0.750808000	-0.344816000
1	-3.415369000	0.007722000	1.173922000
1	-4.155455000	0.509069000	-0.350192000

η-NN

6	-0.201862000	2.192422000	-0.302669000
6	-1.243053000	3.155830000	-0.324582000
6	-2.498313000	2.474078000	-0.016659000
6	-2.215735000	1.098985000	0.197369000
6	-0.811389000	0.912651000	-0.051001000
27	-1.675570000	1.721874000	-1.761285000
7	-0.695538000	0.634475000	-3.099177000
6	-2.544483000	2.725602000	-2.955163000
7	-3.090249000	3.440181000	-3.721055000
6	-3.953201000	4.106496000	-4.623928000
1	-4.944451000	3.637074000	-4.633448000
1	-4.063018000	5.156569000	-4.333225000
1	-3.536454000	4.070290000	-5.636061000

7	-1.673669000	0.067669000	-2.858618000
6	-3.819126000	3.163292000	0.181301000
1	-4.657655000	2.468054000	0.076206000
1	-3.882779000	3.610434000	1.183355000
1	-3.967914000	3.969064000	-0.544718000
6	-1.094731000	4.641674000	-0.493450000
1	-1.925597000	5.069421000	-1.063815000
1	-1.071630000	5.148603000	0.481573000
1	-0.170750000	4.897727000	-1.020723000
6	1.266342000	2.437650000	-0.502217000
1	1.450482000	3.389648000	-1.009368000
1	1.799887000	2.467343000	0.458564000
1	1.728989000	1.651680000	-1.108319000
6	-0.076338000	-0.393938000	0.050722000
1	0.800370000	-0.416485000	-0.604447000
1	0.274451000	-0.568572000	1.077061000
1	-0.714186000	-1.239106000	-0.227236000
6	-3.183410000	0.027108000	0.610190000
1	-2.988862000	-0.916203000	0.089257000
1	-3.118261000	-0.175247000	1.688946000
1	-4.217378000	0.311723000	0.392235000

[Co(η^5 -Cp)(CNMe)(N₂)]

κ -N

6	-0.086412000	2.479188000	-0.130028000
6	-1.180195000	3.370908000	-0.046964000
6	-2.366419000	2.568042000	-0.038093000
6	-1.996575000	1.180561000	0.002413000
6	-0.589614000	1.127489000	-0.089866000
1	0.958224000	2.753949000	-0.201255000
1	-1.136288000	4.451097000	-0.031231000
1	-3.379999000	2.945850000	-0.001697000
1	-2.675134000	0.340608000	0.058898000
1	0.018092000	0.232605000	-0.125500000
27	-1.352015000	2.141935000	-1.766585000
7	-0.184543000	1.742458000	-3.029084000
7	0.586627000	1.475121000	-3.800807000
6	-2.679931000	2.499905000	-2.880743000
7	-3.606022000	2.668959000	-3.599352000
6	-4.524139000	3.319882000	-4.464591000
1	-4.492938000	2.857233000	-5.456460000
1	-5.541760000	3.221280000	-4.072646000
1	-4.283334000	4.385762000	-4.562919000

η-NN

6	-0.421015000	2.901867000	-1.001206000
6	-1.474819000	3.841404000	-1.068066000
6	-2.713699000	3.156225000	-0.758517000
6	-2.413228000	1.799106000	-0.510436000
6	-1.009358000	1.630705000	-0.721492000
1	0.627518000	3.093647000	-1.184396000
1	-1.379771000	4.897568000	-1.283266000
1	-3.690663000	3.618283000	-0.706932000
1	-3.118340000	1.016650000	-0.264405000
1	-0.474436000	0.691243000	-0.660043000
27	-1.855770000	2.377918000	-2.484651000
7	-1.909860000	0.697375000	-3.542827000
7	-0.906140000	1.212579000	-3.782546000
6	-2.666559000	3.371209000	-3.729784000
7	-3.206465000	4.042096000	-4.532309000
6	-3.838236000	4.817608000	-5.531353000
1	-3.119917000	5.092384000	-6.311900000
1	-4.658415000	4.253341000	-5.988970000
1	-4.245650000	5.733652000	-5.091031000

[Co(η⁵-Cp)(CO)(N₂)]

κ -N

6	-0.062374000	2.386134000	-0.018165000
6	-1.095189000	3.345326000	0.013745000
6	-2.329790000	2.633914000	-0.167374000
6	-2.059790000	1.223520000	-0.192163000
6	-0.658427000	1.075032000	-0.145676000
1	0.999506000	2.585449000	0.045479000
1	-0.983495000	4.416261000	0.111123000
1	-3.313225000	3.084490000	-0.203181000
1	-2.793647000	0.433960000	-0.275245000
1	-0.111006000	0.142524000	-0.192006000
27	-1.164691000	2.263063000	-1.801566000
7	0.100722000	1.797155000	-2.957654000
7	0.922906000	1.488849000	-3.653385000
6	-2.278714000	2.887011000	-2.977532000
8	-3.051999000	3.313537000	-3.724648000

 η -NN

6	-0.434322000	2.956498000	-1.060620000
6	-1.541587000	3.835146000	-1.079532000
6	-2.722137000	3.087104000	-0.688747000
6	-2.332886000	1.753022000	-0.441003000

6	-0.935571000	1.660609000	-0.730769000
1	0.589271000	3.201874000	-1.308168000
1	-1.519517000	4.891171000	-1.314032000
1	-3.718578000	3.496317000	-0.588441000
1	-2.977720000	0.936480000	-0.146201000
1	-0.347280000	0.752353000	-0.692118000
27	-1.920914000	2.334346000	-2.446115000
7	-1.931798000	0.614556000	-3.471992000
7	-0.975774000	1.173998000	-3.775243000
6	-2.832851000	3.257242000	-3.615983000
8	-3.445485000	3.906305000	-4.347814000

[Co(η^5 -Cp)(PH₃)(N₂)]

κ-N			
6	0.056532000	2.321211000	-0.048574000
6	-0.894397000	3.365887000	-0.074570000
6	-2.188498000	2.744921000	-0.067270000
6	-2.035032000	1.325571000	0.082628000
6	-0.647388000	1.062092000	0.048535000
1	1.132617000	2.433005000	-0.083071000
1	-0.693316000	4.427529000	-0.116426000
1	-3.134548000	3.272635000	-0.082583000

1	-2.832456000	0.601546000	0.178448000
1	-0.178604000	0.087700000	0.097907000
27	-1.165638000	2.047214000	-1.700536000
1	-2.290237000	2.450832000	-4.532936000
1	-3.912320000	2.253819000	-3.177312000
1	-2.882279000	4.093142000	-3.324959000
15	-2.567562000	2.715355000	-3.169225000
7	-0.069523000	1.345056000	-2.870904000
7	0.657521000	0.883056000	-3.596344000

η-NN

6	-0.153565000	2.518367000	-0.335051000
6	-1.426094000	3.178890000	-0.254480000
6	-2.417569000	2.248828000	0.209212000
6	-1.776167000	0.992610000	0.305100000
6	-0.387392000	1.157817000	-0.029268000
1	0.790880000	2.972892000	-0.600814000
1	-1.593242000	4.235666000	-0.426377000
1	-3.455297000	2.467785000	0.419916000
1	-2.246910000	0.053777000	0.567485000
1	0.346290000	0.362298000	-0.056315000
27	-1.587214000	1.710201000	-1.650924000

1	-2.383258000	2.479407000	-4.531742000
1	-3.317309000	3.847497000	-3.191068000
1	-1.279739000	4.095714000	-3.688619000
15	-2.142632000	3.052499000	-3.259838000
7	-2.209570000	0.083779000	-2.566324000
7	-1.091683000	0.221109000	-2.837368000

[Co(η^5 -Cp)(PMe₃)(N₂)]

κ-N			
6	0.127576000	1.478267000	-0.260538000
6	-0.325617000	2.732139000	0.288966000
6	-1.734331000	2.690609000	0.402625000
6	-2.153410000	1.430431000	-0.141009000
6	-0.998232000	0.653902000	-0.489393000
1	1.160410000	1.220121000	-0.456970000
1	0.314996000	3.558956000	0.568150000
1	-2.376815000	3.459390000	0.809569000
1	-3.178328000	1.083275000	-0.194086000
1	-0.995940000	-0.360502000	-0.864040000
27	-1.138098000	2.441402000	-1.611016000
15	-2.762487000	2.477753000	-3.027387000
7	-0.017334000	3.293643000	-2.641056000

7	0.743065000	3.840877000	-3.269988000
6	-4.315036000	3.333370000	-2.486946000
1	-5.102357000	3.273444000	-3.247886000
1	-4.684035000	2.883129000	-1.560066000
1	-4.092549000	4.385723000	-2.285080000
6	-2.438218000	3.302971000	-4.648468000
1	-1.605242000	2.807950000	-5.156668000
1	-3.321052000	3.269260000	-5.296510000
1	-2.158438000	4.347756000	-4.483127000
6	-3.418091000	0.834590000	-3.579633000
1	-3.739050000	0.251294000	-2.710899000
1	-4.265521000	0.941459000	-4.267337000
1	-2.618765000	0.280217000	-4.081097000

η-NN

6	-0.175566000	2.639076000	-0.418735000
6	-1.487384000	3.210498000	-0.300697000
6	-2.396285000	2.216897000	0.195837000
6	-1.662010000	1.010247000	0.286966000
6	-0.302640000	1.268884000	-0.088389000
1	0.727526000	3.156550000	-0.712045000
1	-1.732181000	4.252545000	-0.468924000

1	-3.439124000	2.363967000	0.441207000
1	-2.058385000	0.042855000	0.568403000
1	0.484184000	0.526615000	-0.134154000
27	-1.591355000	1.730652000	-1.695363000
15	-2.271802000	3.039115000	-3.298321000
7	-2.137360000	0.064789000	-2.573734000
7	-1.035044000	0.267741000	-2.876621000
6	-1.146826000	4.432556000	-3.768379000
1	-0.952783000	5.065280000	-2.896511000
1	-0.190313000	4.022713000	-4.107539000
1	-1.573352000	5.051764000	-4.566304000
6	-3.873954000	3.926037000	-3.025006000
1	-4.131385000	4.576219000	-3.869415000
1	-4.671902000	3.190055000	-2.885271000
1	-3.813254000	4.534505000	-2.117145000
6	-2.564307000	2.221232000	-4.927152000
1	-1.640336000	1.748774000	-5.273477000
1	-3.320365000	1.438378000	-4.815787000
1	-2.902405000	2.943377000	-5.678279000

[Co(η^5 -Cp)py(N₂)]

κ -N

6	-0.062364000	2.670255000	-0.196270000
6	-1.176338000	3.538890000	-0.192656000
6	-2.342228000	2.724303000	0.019374000
6	-1.944802000	1.374623000	0.247308000
6	-0.538659000	1.333594000	0.047103000
1	0.971159000	2.948857000	-0.356169000
1	-1.158230000	4.613841000	-0.317488000
1	-3.362224000	3.088677000	0.053882000
1	-2.591380000	0.544169000	0.495227000
1	0.085440000	0.449921000	0.091522000
27	-1.394312000	2.051377000	-1.695022000
7	-0.395745000	1.229475000	-2.876405000
7	0.282336000	0.702146000	-3.608033000
7	-2.726640000	2.555612000	-2.972847000
6	-3.313481000	3.778806000	-2.940877000
6	-3.152042000	1.702321000	-3.938207000
6	-4.308101000	4.171219000	-3.825939000
1	-2.949544000	4.454325000	-2.177473000
6	-4.128508000	2.030461000	-4.869519000
1	-2.681526000	0.727139000	-3.949193000
6	-4.731852000	3.287113000	-4.818913000

1	-4.732036000	5.166598000	-3.738034000
1	-4.413792000	1.296508000	-5.616509000
1	-5.502546000	3.568918000	-5.529681000

η-NN

6	-0.409911000	3.192603000	-1.267874000
6	-1.665175000	3.841309000	-0.994945000
6	-2.555184000	2.914599000	-0.377811000
6	-1.890177000	1.657765000	-0.382430000
6	-0.568150000	1.834894000	-0.906328000
1	0.484292000	3.657548000	-1.662011000
1	-1.883453000	4.886250000	-1.181964000
1	-3.550681000	3.122383000	-0.010063000
1	-2.307227000	0.716020000	-0.048660000
1	0.166112000	1.049080000	-1.029549000
27	-2.026733000	2.337854000	-2.349498000
7	-2.988075000	0.804764000	-3.087157000
7	-1.899283000	0.762540000	-3.491200000
7	-2.532241000	3.481777000	-3.826004000
6	-1.739729000	4.505814000	-4.220184000
6	-3.702956000	3.319189000	-4.483946000
6	-2.079228000	5.378322000	-5.246023000

1	-0.801692000	4.606813000	-3.688649000
6	-4.105657000	4.142066000	-5.529827000
1	-4.320598000	2.493332000	-4.152505000
6	-3.286046000	5.198908000	-5.922703000
1	-1.395873000	6.178446000	-5.511918000
1	-5.056737000	3.953287000	-6.017284000
1	-3.576465000	5.861030000	-6.732833000

[Co(η^5 -Cp)py(p-NMe)₂(N₂)]

κ-N			
6	0.166623000	2.555159000	-0.344154000
6	-0.786245000	3.600868000	-0.355610000
6	-2.034038000	3.025936000	0.063467000
6	-1.834089000	1.663605000	0.437936000
6	-0.482498000	1.358822000	0.125434000
1	1.207298000	2.630099000	-0.632642000
1	-0.610193000	4.636213000	-0.616917000
1	-2.975232000	3.558883000	0.133380000
1	-2.573265000	0.994832000	0.857242000
1	-0.002239000	0.395043000	0.238972000
27	-1.391107000	2.002530000	-1.620281000
7	-0.651247000	0.959263000	-2.809274000

7	-0.149396000	0.281585000	-3.561433000
7	-2.833033000	2.517135000	-2.796698000
6	-3.203471000	3.808424000	-2.957187000
6	-3.552206000	1.609379000	-3.496042000
6	-4.247720000	4.224640000	-3.761665000
1	-2.615555000	4.536014000	-2.410678000
6	-4.603820000	1.929315000	-4.335177000
1	-3.255224000	0.574690000	-3.371632000
6	-5.003084000	3.277411000	-4.493730000
1	-4.457695000	5.284982000	-3.823179000
1	-5.108618000	1.124627000	-4.854533000
7	-6.049703000	3.641679000	-5.302150000
6	-6.376456000	5.048585000	-5.477720000
1	-7.250420000	5.134947000	-6.124895000
1	-6.619920000	5.524364000	-4.519051000
1	-5.551814000	5.610261000	-5.939651000
6	-6.740671000	2.634044000	-6.092753000
1	-7.543723000	3.110497000	-6.656823000
1	-6.068232000	2.138103000	-6.807074000
1	-7.189970000	1.863044000	-5.453570000

η-NN

6	-0.626728000	3.411282000	-1.017728000
6	-2.026590000	3.681469000	-0.841619000
6	-2.694398000	2.494497000	-0.413346000
6	-1.723198000	1.457404000	-0.443044000
6	-0.455512000	2.022150000	-0.799202000
1	0.144644000	4.130346000	-1.260249000
1	-2.497045000	4.649790000	-0.967048000
1	-3.737776000	2.402726000	-0.143942000
1	-1.902760000	0.409379000	-0.238615000
1	0.467890000	1.466747000	-0.904382000
27	-1.847731000	2.299302000	-2.351572000
7	-2.213178000	0.674011000	-3.363267000
7	-1.139721000	1.027135000	-3.642091000
7	-2.602830000	3.425686000	-3.747499000
6	-2.088397000	4.640497000	-4.038697000
6	-3.688538000	3.055572000	-4.460904000
6	-2.606999000	5.495115000	-4.993522000
1	-1.209734000	4.926985000	-3.472405000
6	-4.274206000	3.835597000	-5.441801000
1	-4.090997000	2.075635000	-4.231557000
6	-3.745738000	5.113087000	-5.743475000

1	-2.115582000	6.446058000	-5.155466000
1	-5.139935000	3.444445000	-5.960899000
7	-4.298056000	5.926222000	-6.698488000
6	-5.431526000	5.459199000	-7.483435000
1	-6.292636000	5.225311000	-6.844082000
1	-5.733245000	6.244128000	-8.178247000
1	-5.184177000	4.561924000	-8.067723000
6	-3.684075000	7.209679000	-7.005047000
1	-2.660643000	7.095154000	-7.389190000
1	-4.277323000	7.717256000	-7.766847000
1	-3.647432000	7.857551000	-6.119691000

[Co(η^5 -Cp)py(p-NO)₂(N₂)]

κ-N			
6	0.072810000	2.179465000	-0.250997000
6	-0.298635000	3.506575000	-0.598769000
6	-1.688708000	3.623512000	-0.315282000
6	-2.145851000	2.413230000	0.313565000
6	-1.056333000	1.516171000	0.343821000
1	1.051142000	1.737568000	-0.391797000
1	0.337659000	4.265204000	-1.033002000
1	-2.292940000	4.503186000	-0.500273000

1	-3.142906000	2.225053000	0.690377000
1	-1.062811000	0.505862000	0.730829000
27	-1.445726000	2.063280000	-1.652932000
7	-0.732610000	0.768167000	-2.619664000
7	-0.243811000	-0.083354000	-3.167934000
7	-2.800439000	2.476272000	-2.899937000
6	-4.005474000	2.984570000	-2.502106000
6	-2.635886000	2.310770000	-4.246587000
6	-5.021216000	3.329638000	-3.372039000
1	-4.143468000	3.095359000	-1.435699000
6	-3.604415000	2.615580000	-5.183709000
1	-1.678084000	1.922922000	-4.565935000
6	-4.818478000	3.140232000	-4.739433000
1	-5.959519000	3.726190000	-3.005741000
1	-3.423946000	2.465082000	-6.240386000
7	-5.861985000	3.484706000	-5.689990000
8	-6.916680000	3.941434000	-5.238231000
8	-5.626544000	3.298789000	-6.887604000

η-NN

6	-0.818069000	3.244438000	-0.644468000
6	-2.183146000	3.611977000	-0.808700000

6	-2.996868000	2.422632000	-0.803421000
6	-2.119051000	1.320425000	-0.720864000
6	-0.778959000	1.822763000	-0.656322000
1	0.026945000	3.915363000	-0.573838000
1	-2.554403000	4.627524000	-0.875382000
1	-4.077452000	2.382223000	-0.848548000
1	-2.397586000	0.274944000	-0.743152000
1	0.115115000	1.213193000	-0.623804000
27	-1.641285000	2.478188000	-2.435289000
7	-1.228705000	1.012717000	-3.680640000
7	-0.284336000	1.677270000	-3.618155000
7	-2.353532000	3.598764000	-3.806261000
6	-1.653888000	3.912739000	-4.931037000
6	-3.589706000	4.157652000	-3.682879000
6	-2.139241000	4.741058000	-5.928964000
1	-0.666685000	3.478055000	-5.015595000
6	-4.146338000	5.006479000	-4.622605000
1	-4.145552000	3.888131000	-2.795414000
6	-3.405008000	5.298562000	-5.765800000
1	-1.547035000	4.963742000	-6.807272000
1	-5.135147000	5.424671000	-4.484747000

7	-3.953878000	6.186097000	-6.787107000
8	-3.253923000	6.415605000	-7.775091000
8	-5.080551000	6.646837000	-6.591881000

[Ir(η^5 -Cp)(CNMe)(N₂)]

κ-N			
6	-0.623417000	0.952156000	0.064891000
6	-1.648540000	1.918751000	0.177376000
6	-2.904343000	1.219246000	0.035703000
6	-2.633990000	-0.201290000	0.003891000
6	-1.232237000	-0.356038000	-0.032915000
1	0.440529000	1.149032000	0.074317000
1	-1.519481000	2.986610000	0.281323000
1	-3.885270000	1.667234000	0.123276000
1	-3.371684000	-0.989326000	-0.048307000
1	-0.696197000	-1.292206000	-0.116307000
77	-1.922292000	0.931709000	-1.937404000
7	-0.744338000	0.529337000	-3.410571000
7	-0.033704000	0.278271000	-4.244430000
6	-3.205713000	1.576308000	-3.129523000
7	-4.068075000	1.921837000	-3.887367000
6	-4.568064000	3.072248000	-4.588170000

1	-4.558952000	2.876342000	-5.664914000
1	-5.604379000	3.260768000	-4.289969000
1	-3.966277000	3.965381000	-4.381109000

η-NN

6	-0.558621000	3.301271000	-1.089939000
6	-1.898119000	3.821496000	-0.882593000
6	-2.697428000	2.781745000	-0.256752000
6	-1.903748000	1.614799000	-0.231382000
6	-0.595913000	1.930646000	-0.747168000
1	0.289122000	3.849540000	-1.474924000
1	-2.187636000	4.860058000	-0.975850000
1	-3.719075000	2.877945000	0.081622000
1	-2.223749000	0.636684000	0.103537000
1	0.219899000	1.227691000	-0.854918000
77	-2.180799000	2.414254000	-2.496377000
7	-2.785779000	0.573322000	-3.590589000
7	-1.745207000	0.831022000	-4.006392000
6	-2.975190000	3.464323000	-3.809576000
7	-3.556308000	4.120389000	-4.626168000
6	-3.378098000	4.960653000	-5.775527000
1	-2.318793000	5.062704000	-6.040154000

1	-3.919403000	4.536472000	-6.626995000
1	-3.794155000	5.952881000	-5.573122000

[Rh(η^5 -Cp)(CNMe)(N₂)]

	κ-N		
6	-0.048025000	2.473421000	0.158771000
6	-1.147454000	3.343990000	0.202568000
6	-2.330753000	2.522837000	0.105046000
6	-1.940441000	1.136280000	0.212848000
6	-0.539159000	1.106607000	0.167742000
1	0.996163000	2.759799000	0.152842000
1	-1.119245000	4.424713000	0.215442000
1	-3.348876000	2.888753000	0.152386000
1	-2.607187000	0.285425000	0.234402000
1	0.083683000	0.220949000	0.167491000
45	-1.331174000	2.155872000	-1.846325000
7	-0.111641000	1.719944000	-3.333765000
7	0.598056000	1.465968000	-4.161237000
6	-2.728864000	2.640571000	-3.050348000
7	-3.603436000	2.935044000	-3.783131000
6	-4.616515000	3.358683000	-4.675218000
1	-4.597326000	2.747248000	-5.583575000

1	-5.599560000	3.253407000	-4.204209000
1	-4.469247000	4.408572000	-4.953365000
η-NN			
6	-0.523334000	3.222110000	-0.973450000
6	-1.850522000	3.795337000	-0.897819000
6	-2.741158000	2.819776000	-0.304837000
6	-2.017361000	1.621874000	-0.210906000
6	-0.651228000	1.869432000	-0.623841000
1	0.372556000	3.726302000	-1.307791000
1	-2.087321000	4.840470000	-1.054600000
1	-3.781727000	2.973042000	-0.055100000
1	-2.402404000	0.667031000	0.124675000
1	0.137069000	1.127229000	-0.642469000
45	-2.075379000	2.347549000	-2.518143000
7	-2.675899000	0.485484000	-3.648791000
7	-1.602610000	0.682496000	-3.970089000
6	-2.703026000	3.488344000	-3.904197000
7	-3.094345000	4.196082000	-4.761175000
6	-3.539520000	5.021171000	-5.820802000
1	-2.776307000	5.080777000	-6.604715000
1	-4.460699000	4.617165000	-6.254573000

1 -3.741082000 6.031237000 -5.448963000

[Co(PH₃)₂Cl(N₂)]

κ-N

27 0.051770000 0.971254000 -0.107797000
17 -2.189051000 0.971249000 -0.107883000
15 -0.085506000 -1.225934000 -0.107939000
1 1.079859000 -2.032696000 -0.108129000
1 -0.761471000 -1.842447000 0.966852000
1 -0.761721000 -1.842280000 -1.182667000
15 -0.085515000 3.168442000 -0.107864000
1 1.079846000 3.975209000 -0.106629000
1 -0.760428000 3.784888000 -1.183367000
1 -0.762786000 3.784849000 0.966151000
7 1.763388000 0.971258000 -0.107814000
7 2.890784000 0.971257000 -0.107587000

η-NN

27 0.299674000 0.967286000 -0.189382000
17 -1.857397000 0.940731000 0.380729000
15 0.174699000 -1.277997000 -0.118107000
1 1.310629000 -2.074958000 -0.404591000
1 -0.205855000 -1.877226000 1.101123000

1	-0.753203000	-1.912631000	-0.970545000
15	0.104278000	3.208629000	-0.175655000
1	1.214743000	4.033258000	-0.482910000
1	-0.842708000	3.791691000	-1.043799000
1	-0.295481000	3.826728000	1.027906000
7	2.224205000	0.998647000	-0.099294000
7	1.927876000	0.979608000	-1.219985000

[Co(PMe₃)₂Cl(N₂)]

κ-N			
27	0.040432000	0.971286000	-0.102759000
17	-2.225782000	0.972489000	-0.115072000
15	-0.063383000	-1.255499000	-0.102731000
15	-0.061000000	3.198168000	-0.103077000
7	1.741111000	0.970373000	-0.093567000
7	2.872664000	0.969836000	-0.087467000
6	-0.934931000	3.920292000	-1.556814000
1	-1.931180000	3.476020000	-1.623400000
1	-1.019750000	5.009977000	-1.474169000
1	-0.384343000	3.674444000	-2.470588000
6	-0.950463000	3.919491000	1.341635000
1	-1.034033000	5.009266000	1.258899000

1	-1.947487000	3.475446000	1.397127000
1	-0.409851000	3.672779000	2.261116000
6	1.532226000	4.131807000	-0.094160000
1	2.115901000	3.868260000	0.793282000
1	2.125535000	3.868856000	-0.975366000
1	1.359267000	5.213688000	-0.094737000
6	-0.938026000	-1.976904000	-1.556396000
1	-1.024085000	-3.066476000	-1.473557000
1	-1.933769000	-1.531517000	-1.623130000
1	-0.387096000	-1.731847000	-2.470176000
6	-0.953688000	-1.975625000	1.342062000
1	-1.950229000	-1.530486000	1.397450000
1	-1.038437000	-3.065322000	1.259494000
1	-0.412835000	-1.729355000	2.261519000
6	1.528826000	-2.190867000	-0.093569000
1	2.122498000	-1.928684000	-0.974759000
1	2.112712000	-1.927837000	0.793888000
1	1.354686000	-3.272559000	-0.094012000

η-NN

27	2.890640000	0.303502000	0.116527000
17	1.127445000	-0.876814000	-0.647371000

15	1.459623000	1.253230000	1.596410000
15	4.133403000	-0.772746000	-1.444609000
7	4.438921000	0.840161000	1.120663000
7	4.206933000	1.683448000	0.353681000
6	4.153759000	-2.607996000	-1.277454000
1	3.122895000	-2.971270000	-1.289113000
1	4.721359000	-3.081867000	-2.086459000
1	4.605719000	-2.881640000	-0.318452000
6	3.577558000	-0.509765000	-3.181948000
1	4.175182000	-1.093882000	-3.891088000
1	2.526507000	-0.799618000	-3.259930000
1	3.662831000	0.552035000	-3.435211000
6	5.927713000	-0.345373000	-1.516413000
1	6.045878000	0.723932000	-1.717183000
1	6.398021000	-0.559148000	-0.551650000
1	6.443262000	-0.913969000	-2.298072000
6	0.049492000	2.157713000	0.828314000
1	-0.483554000	1.473485000	0.163137000
1	-0.640118000	2.549582000	1.584682000
1	0.433158000	2.991329000	0.231049000
6	0.632447000	0.062326000	2.734003000

1	-0.089043000	0.563088000	3.389681000
1	0.121088000	-0.695420000	2.134734000
1	1.386995000	-0.436451000	3.351020000
6	2.169856000	2.503572000	2.753735000
1	2.966302000	2.046667000	3.349194000
1	2.608610000	3.328395000	2.183805000
1	1.405131000	2.903666000	3.428436000

[Co(CO)₂Cl(N₂)]

κ-N			
27	0.087111000	0.971093000	0.046472000
17	-1.994192000	0.973384000	-0.640422000
6	-0.110507000	-0.775739000	0.447401000
6	-0.108853000	2.717543000	0.450005000
7	1.860401000	0.970391000	0.211999000
7	2.976155000	0.970200000	0.217802000
8	-0.321725000	3.766936000	0.851042000
8	-0.324316000	-1.825487000	0.847016000

η-NN			
27	2.807859000	0.519946000	-0.099790000
17	1.098990000	-0.635379000	-0.833207000
6	1.629866000	1.266328000	1.080893000

6	3.782120000	-0.364885000	-1.367765000
7	4.415015000	1.114354000	0.915854000
7	4.180634000	1.939652000	0.160998000
8	4.342090000	-0.944192000	-2.173438000
8	0.854136000	1.699337000	1.794412000

[Co(CNMe)2Cl(N2)]

κ-N

27	0.226988000	0.970642000	0.749114000
17	-2.005246000	0.971234000	0.888445000
6	0.109214000	-0.856117000	0.756775000
6	0.110341000	2.797627000	0.760149000
7	1.965129000	0.970218000	0.638249000
7	3.085230000	0.969917000	0.566141000
7	-0.035210000	3.957952000	0.772696000
7	-0.037118000	-2.016369000	0.767540000
6	-0.345750000	5.339265000	0.796045000
1	-1.429862000	5.468989000	0.867774000
1	0.132169000	5.814579000	1.658364000
1	0.014924000	5.818246000	-0.119611000
6	-0.348683000	-3.397480000	0.788885000
1	-1.432922000	-3.526504000	0.859972000

1	0.012016000	-3.875504000	-0.127263000
1	0.128512000	-3.874311000	1.650766000

η -NN

27	2.881178000	0.903027000	-0.290582000
17	1.092974000	0.036192000	-1.287084000
6	1.705587000	1.681356000	0.905308000
6	3.886294000	0.041474000	-1.581575000
7	4.432702000	1.174101000	0.891225000
7	4.359571000	2.101469000	0.215374000
7	0.908028000	2.141089000	1.624249000
7	4.457736000	-0.529009000	-2.425550000
6	-0.132379000	2.651083000	2.438641000
1	-1.096542000	2.294140000	2.064427000
1	0.002505000	2.312110000	3.470338000
1	-0.119302000	3.745095000	2.417269000
6	5.057937000	-1.253418000	-3.484223000
1	5.741086000	-2.006399000	-3.079472000
1	4.280773000	-1.750114000	-4.072751000
1	5.619611000	-0.571541000	-4.130010000

[Ti(η^5 -Cp*)₂(N₂)] singlet

κ -N

6	-0.993559000	3.392063000	-2.638938000
6	-2.102514000	4.279052000	-2.833750000
6	-3.310222000	3.534478000	-2.608133000
6	-2.929203000	2.178837000	-2.289526000
6	-1.508843000	2.100095000	-2.331497000
22	-2.317154000	2.717126000	-4.540107000
7	-5.099653000	1.344012000	-4.874704000
7	-4.084343000	1.840293000	-4.747541000
6	-4.701715000	4.099323000	-2.503227000
1	-5.456922000	3.404680000	-2.886798000
1	-4.965452000	4.322633000	-1.458908000
1	-4.801791000	5.034210000	-3.065218000
6	-2.021705000	5.769073000	-3.044171000
1	-2.840887000	6.140705000	-3.669681000
1	-2.079706000	6.309224000	-2.088546000
1	-1.083822000	6.064601000	-3.525159000
6	0.454243000	3.785146000	-2.537677000
1	0.671989000	4.714145000	-3.069430000
1	0.728196000	3.945044000	-1.485015000
1	1.126447000	3.012660000	-2.926291000
6	-0.684002000	0.881540000	-2.011084000

1	0.272185000	0.883109000	-2.544425000
1	-0.454923000	0.828189000	-0.937482000
1	-1.206884000	-0.044348000	-2.273645000
6	-3.851636000	1.099818000	-1.794534000
1	-3.439231000	0.100116000	-1.966615000
1	-4.017654000	1.201794000	-0.712577000
1	-4.831441000	1.135635000	-2.278721000
1	-4.673031000	2.118303000	-7.326973000
1	-3.807570000	2.799825000	-8.707797000
6	-3.975699000	2.906150000	-7.626436000
1	-4.480790000	3.865289000	-7.470588000
6	-2.673444000	2.834154000	-6.877886000
6	-1.946577000	1.625539000	-6.559360000
6	-0.663725000	2.017842000	-6.052230000
6	-0.589778000	3.446576000	-6.082423000
6	-1.834850000	3.943975000	-6.561882000
6	-2.169580000	5.392405000	-6.803948000
1	-3.235244000	5.596530000	-6.651505000
1	-1.932399000	5.686382000	-7.836018000
1	-1.608543000	6.058760000	-6.141395000
1	0.418146000	5.248043000	-5.404983000

6	0.643242000	4.277245000	-5.859067000
1	1.375365000	3.773821000	-5.223319000
1	1.137538000	4.481888000	-6.819468000
6	0.463218000	1.079201000	-5.705781000
1	0.099226000	0.156185000	-5.240951000
1	1.025396000	0.784203000	-6.603004000
1	1.178086000	1.534820000	-5.012900000
6	-2.376517000	0.227237000	-6.912939000
1	-3.443798000	0.065266000	-6.727782000
1	-2.194219000	0.011664000	-7.976043000
1	-1.826437000	-0.522993000	-6.334743000

η -NN

6	-1.015583000	3.227333000	-2.654228000
6	-2.031624000	4.213793000	-2.881523000
6	-3.300798000	3.604119000	-2.599640000
6	-3.059119000	2.269123000	-2.169283000
6	-1.664886000	2.018717000	-2.244075000
22	-2.310619000	2.573931000	-4.543662000
7	-3.520396000	0.839970000	-4.605407000
7	-4.238014000	1.750465000	-4.800605000
6	-4.639008000	4.293183000	-2.634956000

1	-5.437466000	3.611850000	-2.945632000
1	-4.911255000	4.692606000	-1.647646000
1	-4.640779000	5.137317000	-3.333156000
6	-1.819203000	5.692593000	-3.073750000
1	-2.583471000	6.143200000	-3.715386000
1	-1.864712000	6.216539000	-2.108251000
1	-0.844139000	5.918145000	-3.513080000
6	0.464825000	3.474461000	-2.536001000
1	0.769964000	4.399762000	-3.028578000
1	0.748726000	3.564667000	-1.477694000
1	1.064886000	2.659191000	-2.954759000
6	-0.973173000	0.739262000	-1.856045000
1	-0.034666000	0.600805000	-2.404271000
1	-0.722504000	0.729387000	-0.785577000
1	-1.601912000	-0.135572000	-2.050365000
6	-4.098175000	1.292634000	-1.698656000
1	-3.860685000	0.266154000	-1.994830000
1	-4.170912000	1.313011000	-0.602451000
1	-5.088264000	1.527056000	-2.099169000
1	-4.508491000	1.813566000	-7.316221000
1	-3.779188000	2.630682000	-8.705279000

6	-3.921368000	2.686957000	-7.616758000
1	-4.531568000	3.572377000	-7.411822000
6	-2.597097000	2.747624000	-6.907900000
6	-1.726426000	1.635652000	-6.666635000
6	-0.511857000	2.142528000	-6.113932000
6	-0.613636000	3.564750000	-6.046247000
6	-1.912829000	3.934525000	-6.504747000
6	-2.418416000	5.340281000	-6.697819000
1	-3.488409000	5.426834000	-6.477020000
1	-2.281007000	5.667003000	-7.738204000
1	-1.890033000	6.056524000	-6.062369000
1	0.209527000	5.438498000	-5.304886000
6	0.524901000	4.515889000	-5.800746000
1	1.320597000	4.067401000	-5.201590000
1	0.974476000	4.810664000	-6.759913000
6	0.710538000	1.318912000	-5.802754000
1	0.449516000	0.351654000	-5.358970000
1	1.286349000	1.106685000	-6.714442000
1	1.384668000	1.829786000	-5.108832000
6	-1.987018000	0.213518000	-7.077846000
1	-3.026655000	-0.080573000	-6.903429000

1	-1.779345000	0.073168000	-8.148083000
1	-1.355517000	-0.490529000	-6.526743000

[Ti(η^5 -Cp*)₂(N₂)] triplet

	κ-N		
6	-1.024090000	3.366662000	-2.629383000
6	-2.116508000	4.286480000	-2.751815000
6	-3.322586000	3.561388000	-2.531110000
6	-2.973768000	2.190600000	-2.278727000
6	-1.557034000	2.068795000	-2.359568000
22	-2.327525000	2.714133000	-4.546763000
7	-5.190789000	1.376949000	-4.940380000
7	-4.172853000	1.852539000	-4.800290000
6	-4.705414000	4.150063000	-2.444035000
1	-5.474961000	3.447612000	-2.780688000
1	-4.953666000	4.429383000	-1.409878000
1	-4.799630000	5.055304000	-3.053272000
6	-2.004545000	5.779192000	-2.916200000
1	-2.866369000	6.196953000	-3.447297000
1	-1.949902000	6.285093000	-1.941410000
1	-1.107079000	6.063807000	-3.475038000
6	0.431648000	3.730201000	-2.517350000

1	0.662050000	4.683121000	-2.999317000
1	0.714674000	3.828373000	-1.459377000
1	1.089059000	2.968941000	-2.950749000
6	-0.755473000	0.836052000	-2.035239000
1	0.192581000	0.809330000	-2.582270000
1	-0.511196000	0.787107000	-0.963813000
1	-1.302086000	-0.080305000	-2.282100000
6	-3.915839000	1.104805000	-1.833721000
1	-3.593899000	0.115755000	-2.177430000
1	-3.968434000	1.065674000	-0.736542000
1	-4.934376000	1.262951000	-2.199446000
1	-4.513521000	1.941942000	-7.552391000
1	-3.738372000	2.949756000	-8.775319000
6	-3.919746000	2.849154000	-7.695928000
1	-4.540682000	3.698833000	-7.391775000
6	-2.617395000	2.804714000	-6.943322000
6	-1.858815000	1.623859000	-6.635428000
6	-0.614682000	2.035995000	-6.077551000
6	-0.601437000	3.468916000	-6.047613000
6	-1.847935000	3.940935000	-6.562029000
6	-2.203716000	5.379772000	-6.828070000

1	-3.287048000	5.539852000	-6.814860000
1	-1.841937000	5.705732000	-7.814433000
1	-1.763809000	6.054102000	-6.086088000
1	0.340301000	5.286505000	-5.315648000
6	0.601031000	4.337527000	-5.796110000
1	1.350115000	3.843335000	-5.173117000
1	1.089730000	4.588289000	-6.748668000
6	0.541406000	1.131859000	-5.739950000
1	0.203598000	0.154583000	-5.379135000
1	1.175471000	0.949426000	-6.619637000
1	1.185170000	1.561290000	-4.965396000
6	-2.253806000	0.209806000	-6.968149000
1	-3.334201000	0.052329000	-6.885175000
1	-1.968024000	-0.050324000	-7.997604000
1	-1.765871000	-0.513583000	-6.306217000

η-NN

6	-1.010714000	3.331514000	-2.664429000
6	-2.077997000	4.278322000	-2.781829000
6	-3.299606000	3.592058000	-2.519808000
6	-2.986534000	2.223752000	-2.245776000
6	-1.577305000	2.054907000	-2.360095000

22	-2.369001000	2.668753000	-4.554979000
7	-4.145967000	1.215269000	-4.756429000
7	-4.606577000	2.254244000	-4.916460000
6	-4.660691000	4.227840000	-2.424264000
1	-5.459841000	3.535269000	-2.706145000
1	-4.868065000	4.565085000	-1.398655000
1	-4.745877000	5.103263000	-3.076664000
6	-1.937014000	5.766850000	-2.960150000
1	-2.770989000	6.192139000	-3.528854000
1	-1.914761000	6.281716000	-1.988804000
1	-1.013728000	6.032841000	-3.483677000
6	0.453845000	3.656635000	-2.545946000
1	0.707591000	4.615122000	-3.003055000
1	0.733596000	3.722963000	-1.484630000
1	1.096153000	2.890971000	-2.992986000
6	-0.808814000	0.803649000	-2.027233000
1	0.137317000	0.746788000	-2.575119000
1	-0.564297000	0.757712000	-0.955711000
1	-1.381414000	-0.099336000	-2.264597000
6	-3.953888000	1.169596000	-1.779955000
1	-3.690921000	0.173852000	-2.151872000

1	-3.963030000	1.116525000	-0.682169000
1	-4.978430000	1.378260000	-2.101045000
1	-4.372511000	1.739304000	-7.659086000
1	-3.638512000	2.831039000	-8.834736000
6	-3.842350000	2.690230000	-7.763930000
1	-4.530637000	3.484351000	-7.456339000
6	-2.558963000	2.714818000	-6.978908000
6	-1.763902000	1.578442000	-6.629759000
6	-0.561421000	2.052437000	-6.028534000
6	-0.615599000	3.482833000	-6.010183000
6	-1.864566000	3.889883000	-6.573282000
6	-2.282570000	5.307085000	-6.862974000
1	-3.372386000	5.415860000	-6.868472000
1	-1.920878000	5.637395000	-7.847934000
1	-1.885533000	6.009635000	-6.123077000
1	0.242354000	5.340825000	-5.267404000
6	0.542436000	4.411087000	-5.761602000
1	1.326343000	3.949007000	-5.158231000
1	1.000560000	4.696412000	-6.719594000
6	0.624097000	1.199670000	-5.661684000
1	0.320865000	0.223149000	-5.269234000

1	1.261176000	1.010076000	-6.537643000
1	1.255075000	1.675715000	-4.905075000
6	-2.078880000	0.142297000	-6.952739000
1	-3.154019000	-0.060328000	-6.923026000
1	-1.724428000	-0.123753000	-7.958816000
1	-1.600043000	-0.545951000	-6.248264000

[Zr(η^5 -Cp*)₂(N₂)] singlet

κ-N			
6	-1.096767000	3.666813000	-2.481442000
6	-2.315349000	4.405464000	-2.544057000
6	-3.407681000	3.484873000	-2.384566000
6	-2.845430000	2.161360000	-2.218511000
6	-1.423738000	2.283597000	-2.296228000
40	-2.394917000	2.914133000	-4.576685000
7	-5.331150000	1.459739000	-4.915334000
7	-4.320846000	1.970893000	-4.799489000
6	-4.848629000	3.856665000	-2.146990000
1	-5.539107000	3.107036000	-2.546205000
1	-5.059838000	3.954411000	-1.071587000
1	-5.103453000	4.815143000	-2.612221000
6	-2.443941000	5.904523000	-2.635614000

1	-3.311388000	6.209184000	-3.233621000
1	-2.575346000	6.353022000	-1.640364000
1	-1.557131000	6.363823000	-3.084190000
6	0.287757000	4.255190000	-2.454241000
1	0.365989000	5.160335000	-3.065319000
1	0.570177000	4.532689000	-1.428751000
1	1.043333000	3.550023000	-2.814095000
6	-0.437408000	1.167438000	-2.069522000
1	0.529797000	1.371861000	-2.539564000
1	-0.250731000	1.017766000	-0.996722000
1	-0.800303000	0.2135581000	-2.467746000
6	-3.608563000	0.934543000	-1.797783000
1	-3.044833000	0.018848000	-2.003492000
1	-3.817810000	0.953024000	-0.717882000
1	-4.570281000	0.847699000	-2.313242000
1	-4.651829000	2.409475000	-7.532762000
1	-3.673661000	2.898006000	-8.920740000
6	-3.848423000	3.081577000	-7.850067000
1	-4.224562000	4.106223000	-7.757326000
6	-2.586277000	2.874481000	-7.053453000
6	-2.031812000	1.590309000	-6.680172000

6	-0.719587000	1.822346000	-6.161706000
6	-0.445172000	3.227848000	-6.234120000
6	-1.594775000	3.875652000	-6.773825000
6	-1.719130000	5.340898000	-7.104575000
1	-2.730312000	5.722020000	-6.916695000
1	-1.505373000	5.527393000	-8.166771000
1	-1.023050000	5.952160000	-6.521203000
1	0.761622000	4.916845000	-5.601341000
6	0.877444000	3.882614000	-5.941796000
1	1.443506000	3.346159000	-5.174351000
1	1.504404000	3.908458000	-6.844363000
6	0.262699000	0.752553000	-5.760293000
1	-0.230936000	-0.095700000	-5.273553000
1	0.793222000	0.355433000	-6.637258000
1	1.022156000	1.130781000	-5.068577000
6	-2.631410000	0.250509000	-7.012179000
1	-3.712321000	0.226469000	-6.840461000
1	-2.463281000	-0.005064000	-8.069010000
1	-2.187741000	-0.549558000	-6.410763000

η-NN

6	-1.035698000	3.238278000	-2.483465000
---	--------------	-------------	--------------

6	-2.038677000	4.241018000	-2.715173000
6	-3.321279000	3.643612000	-2.455623000
6	-3.102869000	2.293603000	-2.049458000
6	-1.705224000	2.033020000	-2.085687000
40	-2.325091000	2.563260000	-4.546986000
7	-3.643115000	0.722045000	-4.647204000
7	-4.366877000	1.637476000	-4.849005000
6	-4.647467000	4.358477000	-2.477405000
1	-5.466050000	3.687146000	-2.755184000
1	-4.889275000	4.782670000	-1.492171000
1	-4.651141000	5.190250000	-3.190996000
6	-1.793170000	5.713555000	-2.925533000
1	-2.569756000	6.179765000	-3.541844000
1	-1.783898000	6.249471000	-1.965347000
1	-0.830437000	5.905458000	-3.408502000
6	0.450656000	3.468129000	-2.394677000
1	0.769924000	4.330714000	-2.985178000
1	0.750248000	3.661143000	-1.354389000
1	1.027601000	2.600876000	-2.735042000
6	-1.030426000	0.751002000	-1.673368000
1	-0.100751000	0.578766000	-2.228200000

1	-0.765198000	0.766341000	-0.606409000
1	-1.676857000	-0.117246000	-1.835117000
6	-4.164454000	1.324769000	-1.610491000
1	-3.916373000	0.294853000	-1.884825000
1	-4.289027000	1.357898000	-0.519215000
1	-5.134144000	1.554364000	-2.060922000
1	-4.505876000	1.813458000	-7.386228000
1	-3.730638000	2.428557000	-8.853632000
6	-3.880773000	2.621144000	-7.781838000
1	-4.456204000	3.548251000	-7.695063000
6	-2.561145000	2.717002000	-7.065474000
6	-1.676576000	1.612130000	-6.811355000
6	-0.464344000	2.138410000	-6.266038000
6	-0.587512000	3.562171000	-6.194977000
6	-1.886137000	3.916591000	-6.676442000
6	-2.401177000	5.320301000	-6.866953000
1	-3.488289000	5.379195000	-6.744045000
1	-2.173979000	5.690027000	-7.876870000
1	-1.949681000	6.022288000	-6.158327000
1	0.159351000	5.447263000	-5.414247000
6	0.523546000	4.528499000	-5.885172000

1	1.278887000	4.092317000	-5.225607000
1	1.039324000	4.827648000	-6.808928000
6	0.774112000	1.341962000	-5.943030000
1	0.531610000	0.340144000	-5.571865000
1	1.402469000	1.208810000	-6.834937000
1	1.392638000	1.834046000	-5.185079000
6	-1.935622000	0.186849000	-7.217148000
1	-2.936127000	-0.148740000	-6.923515000
1	-1.853944000	0.069413000	-8.307073000
1	-1.216338000	-0.500372000	-6.760606000

[Zr(η^5 -Cp*)₂(N₂)] triplet

κ-N			
6	-1.049969000	3.395485000	-2.501405000
6	-2.135868000	4.327427000	-2.613175000
6	-3.347855000	3.612923000	-2.371675000
6	-3.012270000	2.240049000	-2.120741000
6	-1.592837000	2.102955000	-2.208867000
40	-2.391810000	2.701183000	-4.557564000
7	-5.444971000	1.378887000	-4.969552000
7	-4.410217000	1.826606000	-4.832685000
6	-4.721837000	4.219504000	-2.260664000

1	-5.507654000	3.514034000	-2.548827000
1	-4.931855000	4.533879000	-1.227871000
1	-4.827178000	5.104005000	-2.897465000
6	-2.007783000	5.819342000	-2.776997000
1	-2.886863000	6.250177000	-3.267075000
1	-1.899298000	6.319059000	-1.803077000
1	-1.132948000	6.092236000	-3.376727000
6	0.413034000	3.746683000	-2.448690000
1	0.636388000	4.672126000	-2.986080000
1	0.734359000	3.893409000	-1.407373000
1	1.046148000	2.956878000	-2.867876000
6	-0.798644000	0.872328000	-1.858263000
1	0.144756000	0.822139000	-2.412085000
1	-0.546433000	0.853456000	-0.787250000
1	-1.355204000	-0.044778000	-2.076372000
6	-3.966831000	1.162963000	-1.677927000
1	-3.649123000	0.171379000	-2.017508000
1	-4.031213000	1.127105000	-0.581050000
1	-4.980222000	1.328676000	-2.055169000
1	-4.454031000	1.875080000	-7.715510000
1	-3.680593000	2.876625000	-8.945511000

6	-3.865627000	2.784899000	-7.865786000
1	-4.492529000	3.632949000	-7.570240000
6	-2.566809000	2.754065000	-7.104271000
6	-1.808022000	1.579420000	-6.776399000
6	-0.572441000	2.001320000	-6.200188000
6	-0.564344000	3.436981000	-6.181972000
6	-1.802519000	3.901908000	-6.729652000
6	-2.145936000	5.334428000	-7.042528000
1	-3.228090000	5.499163000	-7.051325000
1	-1.763549000	5.628618000	-8.031659000
1	-1.715601000	6.027020000	-6.311638000
1	0.333332000	5.265997000	-5.427059000
6	0.623795000	4.309371000	-5.875433000
1	1.329864000	3.824478000	-5.196137000
1	1.174458000	4.544456000	-6.797770000
6	0.586517000	1.106052000	-5.847589000
1	0.250073000	0.116271000	-5.522399000
1	1.250212000	0.956918000	-6.711989000
1	1.199530000	1.524301000	-5.042072000
6	-2.189663000	0.160395000	-7.106259000
1	-3.273674000	0.010237000	-7.080967000

1	-1.847564000	-0.116861000	-8.114155000
1	-1.744777000	-0.552644000	-6.404301000
η-NN			
6	-1.035280000	3.343269000	-2.547624000
6	-2.100749000	4.301102000	-2.629224000
6	-3.322634000	3.615060000	-2.355211000
6	-3.014959000	2.238076000	-2.113824000
6	-1.603129000	2.063218000	-2.242520000
40	-2.437101000	2.675029000	-4.569859000
7	-4.393515000	1.256370000	-4.823265000
7	-4.814453000	2.324737000	-4.942988000
6	-4.681366000	4.251042000	-2.228115000
1	-5.483497000	3.573280000	-2.536754000
1	-4.882753000	4.542567000	-1.187359000
1	-4.765014000	5.153716000	-2.841805000
6	-1.943669000	5.791185000	-2.784390000
1	-2.818105000	6.244219000	-3.262695000
1	-1.816587000	6.280631000	-1.807540000
1	-1.068646000	6.051311000	-3.388882000
6	0.434611000	3.663637000	-2.485173000
1	0.676479000	4.601573000	-2.990729000

1	0.754026000	3.770697000	-1.438432000
1	1.054949000	2.875934000	-2.925842000
6	-0.831675000	0.814180000	-1.906457000
1	0.099407000	0.741454000	-2.478068000
1	-0.559550000	0.790362000	-0.840412000
1	-1.414261000	-0.089643000	-2.111900000
6	-3.989439000	1.179177000	-1.671683000
1	-3.715068000	0.187479000	-2.046262000
1	-4.023958000	1.117188000	-0.574760000
1	-5.006721000	1.387751000	-2.015816000
1	-4.321254000	1.687246000	-7.790133000
1	-3.592804000	2.760665000	-8.987472000
6	-3.795939000	2.638731000	-7.914179000
1	-4.488183000	3.434924000	-7.620664000
6	-2.514924000	2.683107000	-7.124502000
6	-1.714329000	1.554405000	-6.758246000
6	-0.521391000	2.041717000	-6.144089000
6	-0.585161000	3.475075000	-6.137468000
6	-1.825746000	3.872186000	-6.732246000
6	-2.228411000	5.282153000	-7.075389000
1	-3.316607000	5.393690000	-7.118308000

1	-1.831789000	5.581104000	-8.057528000
1	-1.853149000	6.003995000	-6.342584000
1	0.216275000	5.340763000	-5.360190000
6	0.553355000	4.412415000	-5.834159000
1	1.302466000	3.957524000	-5.181879000
1	1.066370000	4.698601000	-6.763593000
6	0.666832000	1.204690000	-5.748917000
1	0.367148000	0.213725000	-5.392443000
1	1.343727000	1.051938000	-6.602128000
1	1.255198000	1.674829000	-4.954557000
6	-2.023143000	0.114049000	-7.070075000
1	-3.096183000	-0.096237000	-7.018849000
1	-1.686229000	-0.151420000	-8.082527000
1	-1.524723000	-0.566400000	-6.372152000

[Re(η^5 -Cp*)(CO)₂(PMe₃)(N₂)]

κ -N

6	-0.187299000	2.293131000	-0.434115000
6	-1.038935000	3.290488000	0.143014000
6	-2.338358000	2.684946000	0.374301000
6	-2.273529000	1.329479000	-0.069293000
6	-0.953849000	1.077793000	-0.596702000

6	-0.677874000	2.844025000	-3.400255000
8	0.136035000	2.874444000	-4.247657000
7	-2.661138000	4.492636000	-2.294548000
7	-3.097914000	5.533340000	-2.388758000
6	-3.486147000	3.347092000	1.086494000
1	-4.441383000	2.870054000	0.847787000
1	-3.355546000	3.297179000	2.177038000
1	-3.572150000	4.404648000	0.817693000
6	-0.632876000	4.668557000	0.587050000
1	-1.439775000	5.394040000	0.443239000
1	-0.369135000	4.676701000	1.654083000
1	0.234555000	5.029773000	0.027628000
6	1.284281000	2.434553000	-0.710053000
1	1.559671000	3.474731000	-0.904035000
1	1.871857000	2.087256000	0.151000000
1	1.590545000	1.846897000	-1.580076000
6	-0.396877000	-0.265931000	-0.981824000
1	0.411482000	-0.170702000	-1.712611000
1	0.009461000	-0.794015000	-0.106464000
1	-1.162598000	-0.909993000	-1.424865000
6	-3.335663000	0.284953000	0.131823000

1	-3.337907000	-0.458656000	-0.671016000
1	-3.162144000	-0.256258000	1.072420000
1	-4.335766000	0.723853000	0.189780000
75	-1.879265000	2.729854000	-1.939269000
15	-3.468189000	1.767187000	-3.416179000
6	-5.233619000	1.667902000	-2.849285000
6	-3.165827000	0.023846000	-3.970369000
6	-3.675646000	2.627025000	-5.043628000
1	-2.716179000	2.640546000	-5.569265000
1	-4.425783000	2.132791000	-5.671255000
1	-3.980939000	3.663473000	-4.869614000
1	-3.143228000	-0.644824000	-3.104340000
1	-3.942664000	-0.319737000	-4.663411000
1	-2.191559000	-0.029044000	-4.465629000
1	-5.301873000	1.063015000	-1.940563000
1	-5.879793000	1.227088000	-3.617331000
1	-5.592582000	2.675510000	-2.616635000

$\eta\text{-NN}$

6	-0.268622000	2.172075000	-0.213726000
6	-0.877605000	3.444083000	0.010797000
6	-2.306471000	3.245048000	0.164581000

6	-2.567300000	1.850734000	0.033790000
6	-1.317454000	1.171597000	-0.247645000
6	-0.381993000	1.922799000	-3.253260000
8	0.452828000	1.463960000	-3.947834000
7	-2.447234000	4.397552000	-3.156805000
7	-1.402885000	4.660263000	-2.742008000
6	-3.293362000	4.321833000	0.521488000
1	-4.316964000	4.035033000	0.262763000
1	-3.272765000	4.538372000	1.599210000
1	-3.073716000	5.258517000	-0.001116000
6	-0.168350000	4.756449000	0.186045000
1	-0.721226000	5.580521000	-0.275394000
1	-0.047760000	4.992216000	1.252372000
1	0.826108000	4.738144000	-0.267540000
6	1.205746000	1.887731000	-0.291363000
1	1.764680000	2.755574000	-0.651918000
1	1.600638000	1.621728000	0.698985000
1	1.418745000	1.056938000	-0.969659000
6	-1.096151000	-0.316728000	-0.225622000
1	-0.244764000	-0.602442000	-0.850137000
1	-0.895049000	-0.673052000	0.795293000

1	-1.969489000	-0.861970000	-0.595757000
6	-3.878154000	1.170576000	0.313276000
1	-4.015458000	0.272545000	-0.296088000
1	-3.922636000	0.855557000	1.365260000
1	-4.729580000	1.833928000	0.136462000
75	-1.647259000	2.552148000	-2.012395000
15	-3.282621000	1.544579000	-3.420981000
6	-3.379922000	-0.306100000	-3.425032000
6	-3.092019000	1.888574000	-5.229393000
6	-5.055601000	2.017816000	-3.155940000
1	-5.150459000	3.102951000	-3.258357000
1	-5.713326000	1.529235000	-3.884051000
1	-5.374114000	1.737215000	-2.148035000
1	-2.113471000	1.531175000	-5.563284000
1	-3.877368000	1.400608000	-5.817734000
1	-3.136171000	2.968878000	-5.394934000
1	-2.400669000	-0.712633000	-3.695322000
1	-4.129315000	-0.665833000	-4.139522000
1	-3.638643000	-0.674908000	-2.428045000

[Re(η^5 -Cp*)(CO)₂(N₂)]

κ-N

6	-0.072276000	2.146637000	-0.173329000
6	-1.050516000	3.214303000	-0.201754000
6	-2.343400000	2.616152000	-0.006947000
6	-2.183230000	1.199317000	0.138798000
6	-0.766071000	0.907635000	0.030665000
6	-0.080995000	2.331668000	-3.259910000
8	0.793478000	2.696300000	-3.938281000
7	-2.888806000	2.518123000	-3.125044000
7	-3.753662000	2.979601000	-3.675817000
6	-3.641438000	3.366637000	0.106329000
1	-4.491643000	2.757477000	-0.212992000
1	-3.819525000	3.663387000	1.148501000
1	-3.637396000	4.277744000	-0.498665000
6	-0.747154000	4.687390000	-0.237011000
1	-1.567537000	5.259075000	-0.680423000
1	-0.583119000	5.078228000	0.777344000
1	0.153259000	4.899561000	-0.820416000
6	1.420003000	2.334038000	-0.193251000
1	1.715564000	3.161786000	-0.843961000
1	1.790625000	2.555882000	0.816686000
1	1.933782000	1.436547000	-0.547378000

6	-0.134276000	-0.435771000	0.273510000
1	0.844201000	-0.512559000	-0.207781000
1	0.007495000	-0.604983000	1.349695000
1	-0.754907000	-1.250183000	-0.110794000
6	-3.256127000	0.211131000	0.506543000
1	-3.047710000	-0.781660000	0.097689000
1	-3.334414000	0.108753000	1.598170000
1	-4.236642000	0.520729000	0.133832000
75	-1.420319000	1.764827000	-2.010531000
6	-1.549773000	0.047203000	-2.853388000
8	-1.606959000	-1.039879000	-3.269150000

η -NN

6	-0.027568000	2.257726000	-0.096484000
6	-1.036101000	3.265703000	-0.067660000
6	-2.325199000	2.609970000	0.070333000
6	-2.101992000	1.202443000	0.125650000
6	-0.678854000	0.961878000	-0.030324000
6	0.143213000	1.646150000	-3.115359000
8	1.116402000	1.472968000	-3.733605000
7	-2.839001000	3.295580000	-3.141026000
7	-1.826954000	3.810549000	-3.245646000

6	-3.638798000	3.320094000	0.243247000
1	-4.483754000	2.668166000	0.006654000
1	-3.760996000	3.663151000	1.279825000
1	-3.711823000	4.200967000	-0.402366000
6	-0.821474000	4.753447000	-0.059158000
1	-1.593388000	5.277729000	-0.631332000
1	-0.852903000	5.144160000	0.967217000
1	0.146715000	5.023618000	-0.488756000
6	1.458996000	2.481610000	-0.105013000
1	1.722934000	3.436122000	-0.568112000
1	1.847438000	2.489129000	0.922368000
1	1.982564000	1.692488000	-0.651269000
6	0.013980000	-0.359164000	0.164836000
1	0.966953000	-0.393385000	-0.370491000
1	0.220603000	-0.541042000	1.228906000
1	-0.596605000	-1.188560000	-0.202835000
6	-3.136468000	0.143637000	0.388213000
1	-2.899388000	-0.789421000	-0.129888000
1	-3.188981000	-0.076746000	1.462982000
1	-4.131706000	0.459152000	0.063866000
75	-1.360985000	1.896618000	-1.961764000

6	-2.223139000	0.446659000	-2.862158000
8	-2.750621000	-0.486940000	-3.319778000

[Os(NH₃)₅(N₂)]

κ-N

76	-0.773064000	-0.488455000	-0.001082000
7	-2.967755000	-0.426390000	0.005252000
1	-3.328960000	0.353383000	-0.553273000
1	-3.438865000	-1.259073000	-0.360704000
7	-0.782470000	-0.505467000	-2.197803000
1	-0.453193000	0.385331000	-2.583156000
1	-0.189556000	-1.217784000	-2.634578000
7	-0.823524000	-2.693694000	0.018489000
1	0.101880000	-3.133293000	0.028117000
1	-1.305795000	-3.092581000	0.829733000
7	1.423088000	-0.533986000	-0.011454000
1	1.815261000	0.398237000	0.154897000
1	1.856447000	-1.130289000	0.700391000
7	-0.763693000	-0.462022000	2.194706000
1	-0.852675000	0.492054000	2.558690000
1	-1.521038000	-0.986306000	2.643748000
1	-3.376576000	-0.279671000	0.933175000

1	-1.706456000	-0.641370000	-2.619030000
1	1.843137000	-0.836099000	-0.895768000
1	0.092219000	-0.824087000	2.625746000
1	-1.295402000	-3.108106000	-0.791274000
7	-0.725503000	1.448617000	-0.018225000
7	-0.691585000	2.571497000	-0.026595000

η-NN

76	0.758858000	0.171998000	0.107439000
7	-0.606200000	-1.554032000	0.144263000
1	-1.574687000	-1.247758000	0.010053000
1	-0.442570000	-2.257631000	-0.582210000
7	0.681307000	0.186843000	-2.081978000
1	-0.061197000	0.812554000	-2.413996000
1	1.527506000	0.520441000	-2.553774000
7	2.389264000	-1.234954000	0.119654000
1	3.210883000	-0.906297000	0.636920000
1	2.153150000	-2.129774000	0.560230000
7	2.269885000	1.771517000	0.066431000
1	1.825353000	2.681016000	-0.091741000
1	2.802305000	1.889603000	0.933938000
7	0.705966000	0.269197000	2.294814000

1	-0.038100000	0.900304000	2.613214000
1	0.516342000	-0.616327000	2.773920000
1	-0.619218000	-2.080538000	1.023252000
1	0.474872000	-0.714042000	-2.524340000
1	2.979322000	1.685979000	-0.667676000
1	1.555366000	0.625278000	2.743638000
1	2.745422000	-1.485729000	-0.808615000
7	-1.270425000	1.173547000	0.100787000
7	-0.525433000	2.034605000	0.080060000

[Fe(NH₃)₅(N₂)]

κ-N			
26	-0.766144000	-0.498339000	-0.015562000
7	-2.832148000	-0.397267000	-0.045010000
1	-3.190995000	0.483280000	0.338267000
1	-3.252056000	-0.442386000	-0.978683000
7	-0.743848000	-0.664636000	-2.076296000
1	-1.044361000	0.190224000	-2.556051000
1	0.174546000	-0.863179000	-2.485200000
7	-0.849963000	-2.555337000	0.122162000
1	-0.222180000	-3.044129000	-0.523936000
1	-0.605364000	-2.935935000	1.041667000

7	1.299212000	-0.567547000	0.030537000
1	1.703424000	-0.140467000	0.869931000
1	1.730084000	-1.496269000	-0.011004000
7	-0.804887000	-0.356236000	2.047979000
1	-0.679405000	0.600988000	2.393175000
1	-1.680456000	-0.661473000	2.483952000
1	-3.326576000	-1.120622000	0.486718000
1	-1.348449000	-1.396768000	-2.461811000
1	1.741121000	-0.047275000	-0.734289000
1	-0.082970000	-0.893717000	2.538363000
1	-1.769077000	-2.962736000	-0.076598000
7	-0.680968000	1.345219000	-0.152996000
7	-0.632618000	2.455053000	-0.235315000

η-NN

26	0.766926000	0.161066000	0.103803000
7	-0.526037000	-1.465172000	0.099925000
1	-1.169957000	-1.487879000	0.896723000
1	-1.158595000	-1.474298000	-0.706147000
7	0.753921000	0.153320000	-1.956237000
1	-0.019827000	0.691541000	-2.361950000
1	1.586720000	0.558449000	-2.395080000

7	2.312783000	-1.165376000	0.181722000
1	3.217614000	-0.756347000	0.434906000
1	2.182621000	-1.912926000	0.870914000
7	2.170679000	1.694516000	0.073218000
1	1.842885000	2.526501000	-0.426641000
1	2.427207000	2.057844000	0.996495000
7	0.698976000	0.264983000	2.160366000
1	0.273740000	1.131027000	2.509691000
1	0.154705000	-0.472512000	2.619205000
1	-0.092365000	-2.393383000	0.093492000
1	0.667440000	-0.768913000	-2.395420000
1	3.068026000	1.469599000	-0.367691000
1	1.607758000	0.226096000	2.632805000
1	2.500630000	-1.659982000	-0.696360000
7	-1.261886000	1.131511000	0.103452000
7	-0.545993000	1.986137000	0.013092000

[Ru(NH₃)₅(N₂)]

κ-N

44	-0.767358000	-0.499650000	-0.013579000
7	-2.966439000	-0.395402000	-0.052411000
1	-3.308274000	0.512273000	0.275832000

1	-3.373104000	-0.503977000	-0.985492000
7	-0.735810000	-0.647935000	-2.209573000
1	-1.033973000	0.222071000	-2.659879000
1	0.189240000	-0.841897000	-2.602542000
7	-0.856356000	-2.679044000	0.126667000
1	-0.240803000	-3.157018000	-0.537428000
1	-0.591213000	-3.046959000	1.044549000
7	1.431567000	-0.565999000	0.030938000
1	1.823221000	-0.170351000	0.890036000
1	1.849377000	-1.496776000	-0.050221000
7	-0.806743000	-0.346594000	2.182849000
1	-0.695506000	0.619313000	2.504077000
1	-1.680453000	-0.666393000	2.609419000
1	-3.446397000	-1.087034000	0.530007000
1	-1.344575000	-1.373911000	-2.597054000
1	1.854796000	-0.012618000	-0.719444000
1	-0.069838000	-0.872090000	2.661258000
1	-1.786046000	-3.068031000	-0.053743000
7	-0.679288000	1.498810000	-0.149380000
7	-0.630102000	2.609657000	-0.224890000

η-NN

44	0.783301000	0.149730000	0.101354000
7	-0.637895000	-1.534556000	0.109533000
1	-1.329619000	-1.456927000	0.860091000
1	-1.200255000	-1.576454000	-0.744753000
7	0.794234000	0.127100000	-2.097688000
1	0.064083000	0.732028000	-2.486428000
1	1.662574000	0.462608000	-2.523373000
7	2.400119000	-1.239183000	0.178726000
1	3.315022000	-0.791017000	0.282689000
1	2.334951000	-1.895383000	0.962544000
7	2.246498000	1.801112000	0.089336000
1	1.793064000	2.689225000	-0.141905000
1	2.706029000	1.968699000	0.988607000
7	0.697482000	0.244039000	2.297087000
1	0.128766000	1.032530000	2.621078000
1	0.281130000	-0.578357000	2.742580000
1	-0.229706000	-2.467707000	0.210256000
1	0.629029000	-0.789593000	-2.522826000
1	3.009370000	1.705863000	-0.586738000
1	1.601713000	0.363910000	2.761850000
1	2.481138000	-1.822744000	-0.659126000

7	-1.399491000	1.260983000	0.047747000
7	-0.673565000	2.109897000	-0.006359000

[Rh(η^5 -Cp)Cl₂(N₂)]

κ -N

6	-0.102973000	2.353803000	0.232030000
6	-1.151479000	3.295951000	0.139429000
6	-2.370518000	2.580086000	-0.159350000
6	-2.062650000	1.168584000	-0.132942000
6	-0.670591000	1.028576000	0.062319000
1	0.943420000	2.580113000	0.388410000
1	-1.046514000	4.371768000	0.163723000
1	-3.349158000	3.017132000	-0.298963000
1	-2.760049000	0.370962000	-0.348408000
1	-0.120049000	0.097206000	0.070394000
45	-1.079155000	2.240364000	-1.825796000
7	0.486045000	1.742426000	-3.175352000
7	1.273597000	1.499345000	-3.912675000
17	-1.198162000	4.318415000	-3.000749000
17	-2.539102000	1.191525000	-3.401044000

η -NN

NA

[Ir(η^5 -Cp)Cl₂(N₂)]

κ -N

6	-0.086082000	2.350754000	0.208155000
6	-1.145717000	3.294711000	0.127472000
6	-2.373588000	2.579584000	-0.144690000
6	-2.056736000	1.167925000	-0.144780000
6	-0.654566000	1.023619000	0.038173000
1	0.959341000	2.581172000	0.361509000
1	-1.039095000	4.370473000	0.154116000
1	-3.356347000	3.014625000	-0.254500000
1	-2.753219000	0.368677000	-0.357985000
1	-0.107429000	0.090804000	0.042546000
77	-1.064863000	2.233713000	-1.821928000
7	0.422676000	1.759996000	-3.100310000
7	1.238083000	1.501202000	-3.807777000
17	-1.201700000	4.306379000	-3.051438000
17	-2.528097000	1.212625000	-3.447537000

η -NN

NA

[Co(η^5 -Cp)Cl₂(N₂)]

κ -N

6	-0.080256000	2.361516000	0.052545000
6	-1.141706000	3.292746000	0.044385000
6	-2.362014000	2.579406000	-0.180036000
6	-2.045322000	1.184428000	-0.225519000
6	-0.644395000	1.045215000	-0.115966000
1	0.971864000	2.598330000	0.140829000
1	-1.040502000	4.367725000	0.076795000
1	-3.343455000	3.015366000	-0.299550000
1	-2.745896000	0.388693000	-0.432578000
1	-0.091582000	0.117017000	-0.176671000
27	-1.112721000	2.246266000	-1.760490000
7	0.363738000	1.751207000	-2.831146000
7	1.208645000	1.473914000	-3.490053000
17	-1.194595000	4.228829000	-2.807342000
17	-2.489142000	1.205601000	-3.194177000

η-NN

NA

VITA

Upon completing his studies at Carthage High School, Carthage Texas, in 2016, Cole B. Donald entered Panola College. He received an Associate of Science from this institution in 2018. Following this degree, he attended Stephen F. Austin State University to obtain a Bachelor of Science in Chemistry in 2020. He entered the Graduate Program of Natural and Applied Sciences in 2021 to receive the Master of Science in Natural and Applied Science with Chemistry Focus in 2022.

Permanent Address: 902 Green Briar Dr.

Carthage, TX 75633

The American Chemical Society (ACS) Style Guide

This thesis was typed by Cole B. Donald